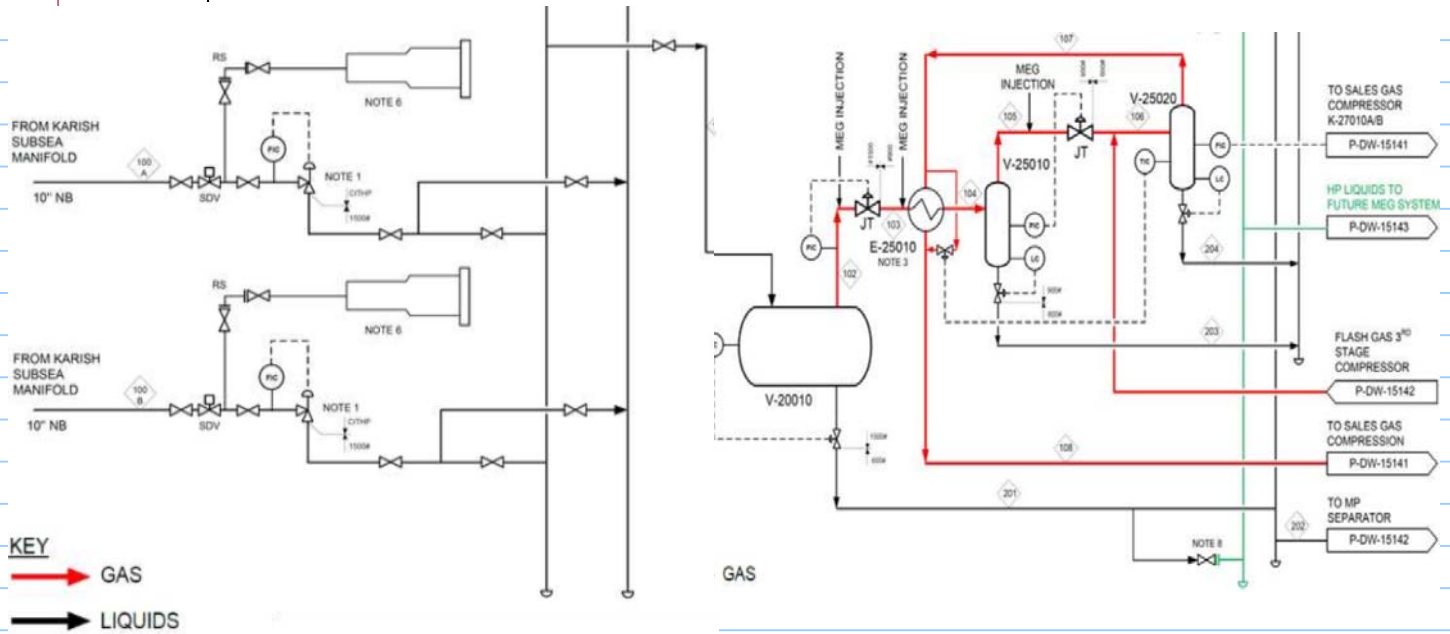
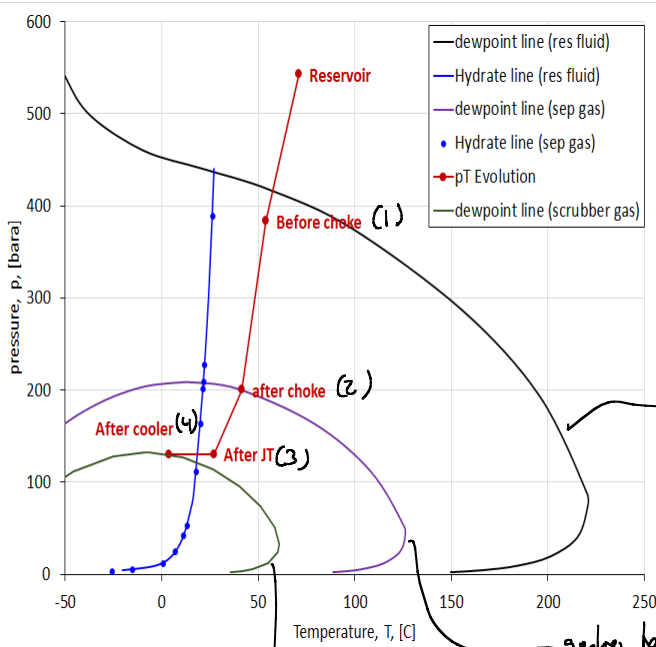
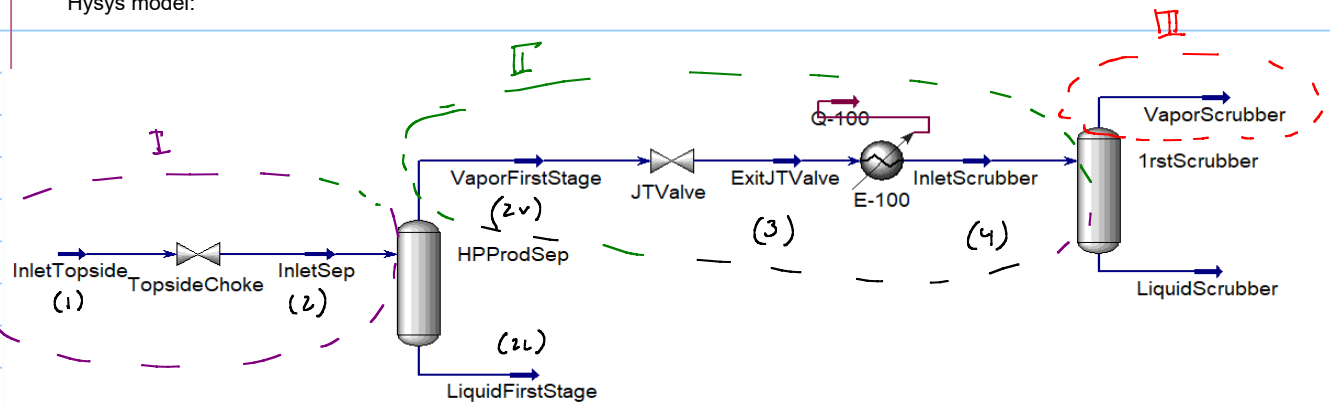


Modeling the topside facilities of the Karish and Tanin field in Hysys



Hysys model:



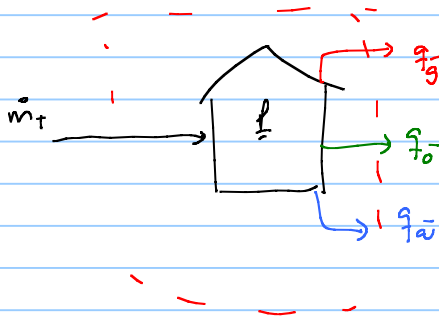
(2), (2v) and (2L) have the same p, T conditions

applies to region I

applies to region II

applies to region III

Estimating inlet mass flow to the processing facilities:



$$\dot{m}_i = \dot{q}_g \rho_g + \dot{q}_o \rho_o + \dot{q}_w \rho_w$$

$\frac{\text{m}^3/\text{d}}{\text{m}^3/\text{d}} \frac{\text{kg}}{\text{m}^3}$ assume q_o and $q_w \ll q_g$

gives:

$$\dot{m}_i \approx \dot{q}_g \rho_g$$

$$\frac{p_{sc}}{\rho_{sc}} = \frac{p_o}{\rho_o} \cdot T_{sc}$$

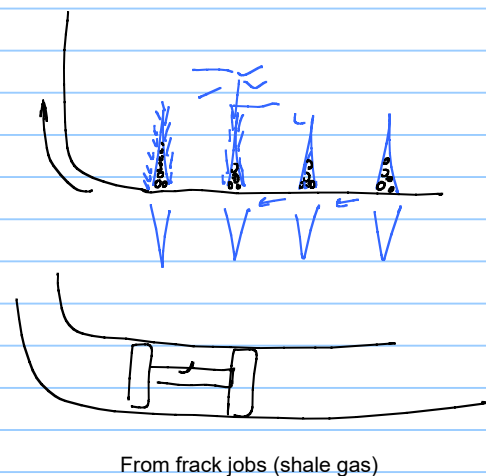
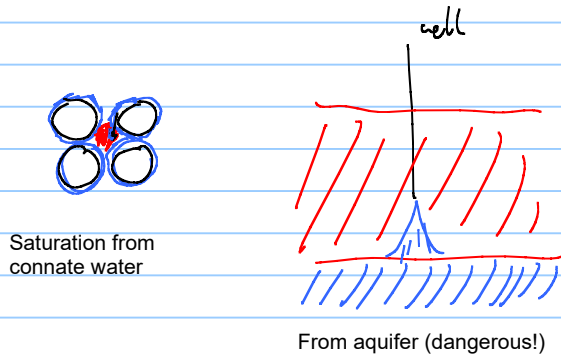
$$\rho_g = \rho_{sc} = \frac{p_{sc} \rho_o}{p_o T_{sc}}$$

B11			$-1.01325/(0.08314 \cdot (15.56 + 273.15)/(84 \cdot 28.97))$
	A	B	C
1	Class exercise, TPG4230, Prof. Milan Stanko (NTNU)		
2			
3	Initial separator pressure	200	bara
4	Gas specific gravity	0.612	[-]
5	Reservoir Temperature, T_R	71	°C
6	Initial reservoir pressure, p_{Ri}	542.5	bara
7	$P_{\text{downstreamTvalve}}$	130	bara
8	$T_{\text{arrivalFPSO}}$	54	°C
9			
10	q_{Karish}	8.33E+06	[Sm ³ /d]
11	sc gas density	7.48E-01	[kg/m ³]
12	massflow	6.24E+06	[kg/d]

q_{Karish}	8.33E+06	[Sm ³ /d]
sc gas density	7.48E-01	[kg/m ³]
massflow	6.24E+06	[kg/d]

• water content in natural gas

Sources of water in gas wells

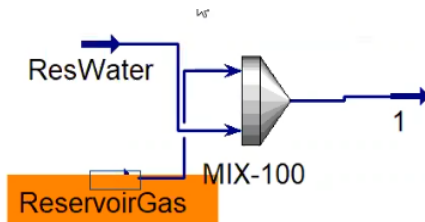


In this field we assume water comes from saturation by connate water only

How to do it in Hysys?

OLD approach:

- Two streams, reservoir water and reservoir gas at same p and T
- Increase molar rate of res water until stream 1 just becomes two phase.
- Read the composition of stream 1



How to do it in Hysys?

NEW approach, use the saturation unit

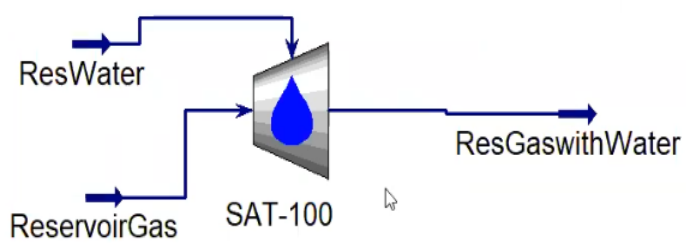


FIG. 20-4

Water Content of Hydrocarbon Gas

mg water/Sm³ of wet gas at 15 °C and 101.325 kPa (abs)

Warning: Dashed lines are meta-stable equilibrium. Actual equilibrium is lower water content. Angle is a function of composition.

Position of this line is a function of gas composition.

Water contents of natural gases with corrections for salinity and relative density. After McKetta and Wehe, Hydrocarbon Processing, August, 1956.

71C

$p_a = 540 \text{ bar}$

Stream Saturator: SAT-100

Design Worksheet

Design

Humidity Indicators

Relative Humidity [%]	100.00	
Mole fraction H ₂ O	0.0019	
Mass fraction H ₂ O	0.0017	
H ₂ O content @ reference conditions	5.444e+005	H ₂ O Content Units
H ₂ O content @ standard conditions	1438	mg/m ³ wet
H ₂ O content @ normal conditions	1506	
Water Dewpoint [C]	71.00	
Wet bulb temperature [C]	70.50	
Partial pressure [bar]	1.017	

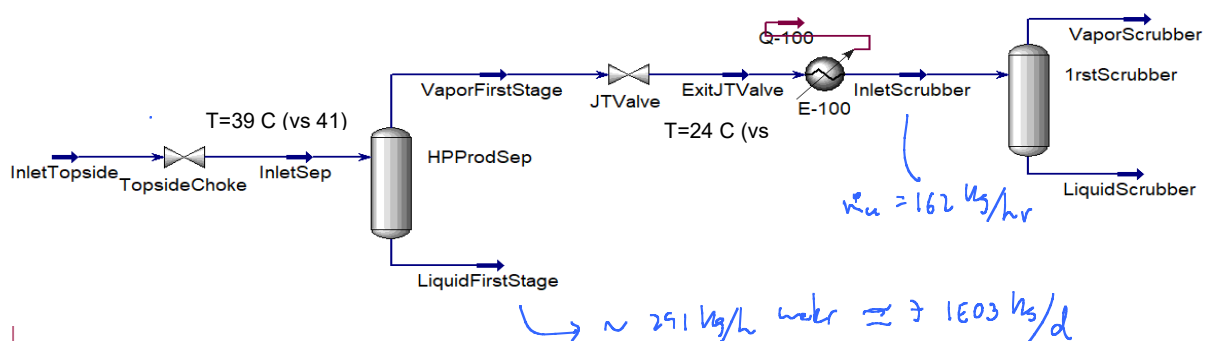
Use process conditions

Reference Conditions

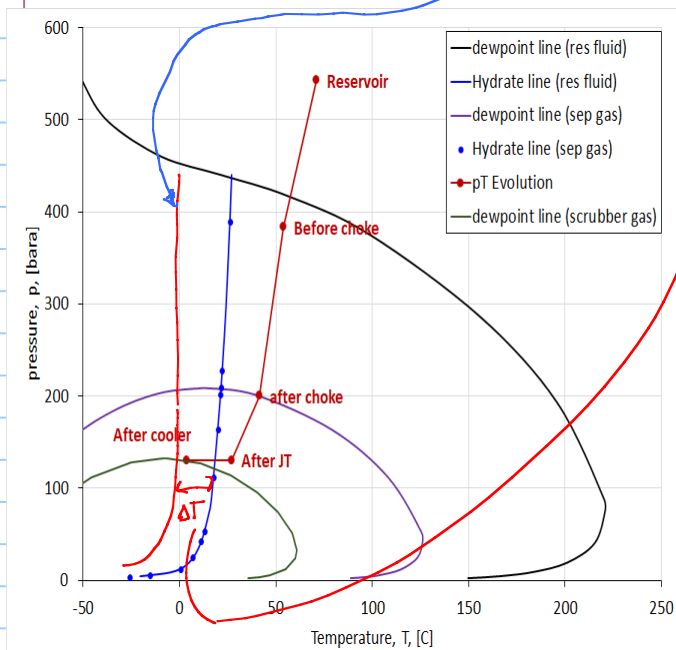
Temperature [C]	71.00	<input checked="" type="checkbox"/>
Pressure [bar]	542.5	<input checked="" type="checkbox"/>
Water content		<input type="checkbox"/>

Component	Mole fraction (res gas)	Mole fraction (res gas sat with water)
N2	6.27E-03	6.26E-03
CO2	7.60E-04	7.59E-04
C1	9.42E-01	0.940683628
C2	1.56E-02	1.56E-02
C3	9.44E-03	9.42E-03
iC4	2.75E-03	2.74E-03
C4	2.97E-03	2.96E-03
iC5	1.80E-03	1.80E-03
C5	1.14E-03	1.14E-03
C6	1.90E-03	1.90E-03
C7	2.77E-03	2.76E-03
C8	1.78E-03	1.78E-03
C9	1.16E-03	1.16E-03
C10	1.00E-03	9.98E-04
C11	9.60E-04	9.58E-04
C12	8.90E-04	8.88E-04
C13	1.11E-03	1.11E-03
C14	6.20E-04	6.19E-04
C15+	4.65E-03	4.64E-03
H2O	0.00E+00	1.87E-03

How does it affect our previous results?



We have free water in the separator gas line, and after the cooler there is a risk of hydrate formation. Therefore it is necessary to inject an inhibitor (e.g. MEG), to move the hydrate line to the left. There should be enough MEG to suppress hydrates in the place where there is most water (just after the cooler)



$$w\% \text{ MEG} = \frac{w_{res}}{w_{res} + w_{H_2O}}$$

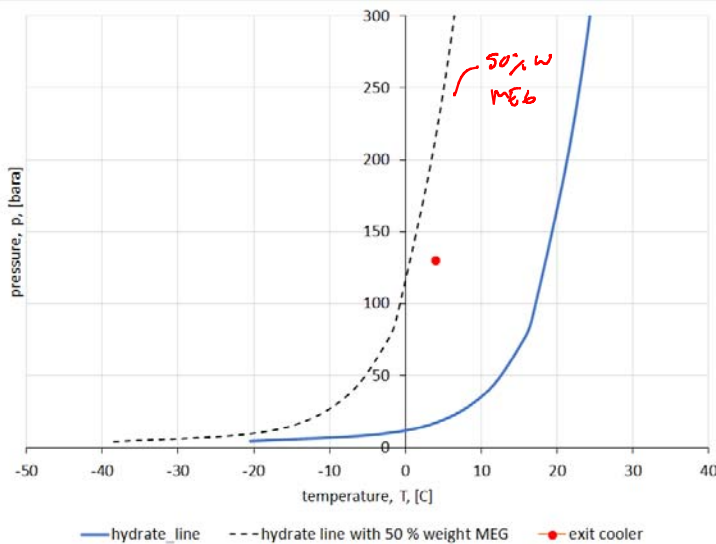
Can be calculated from Hammettschmidt equation

We can use the excel sheet of question 4, quiz 18:

<http://www.ipt.ntnu.no/~stanko/files/Courses/TPG4230/2021/Quizzes/Quiz18.html>

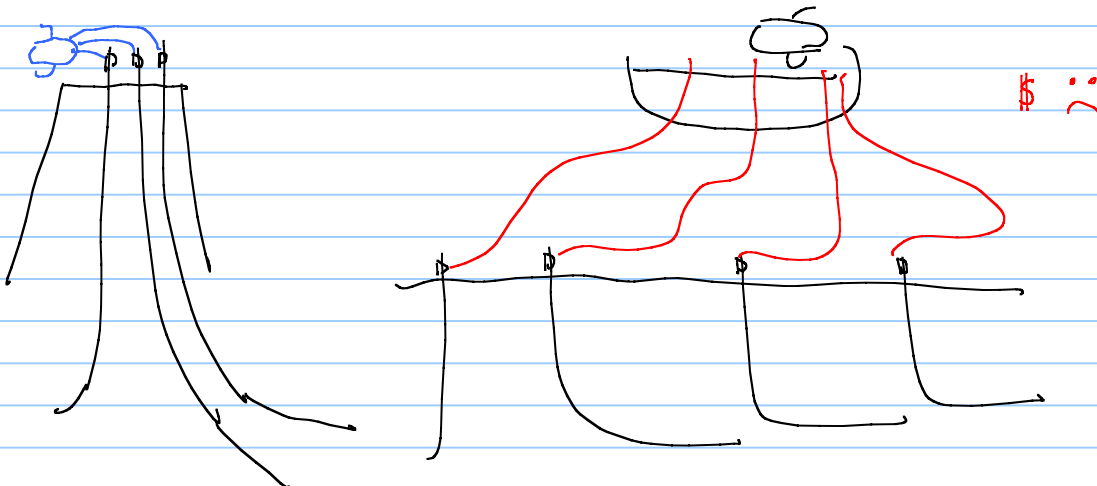
We substitute our hydrate line and calculate the amount of weight %

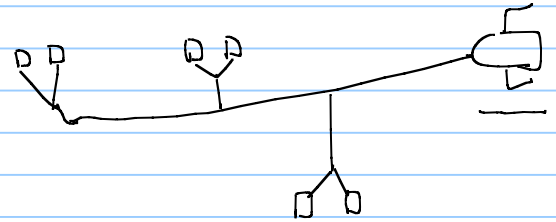
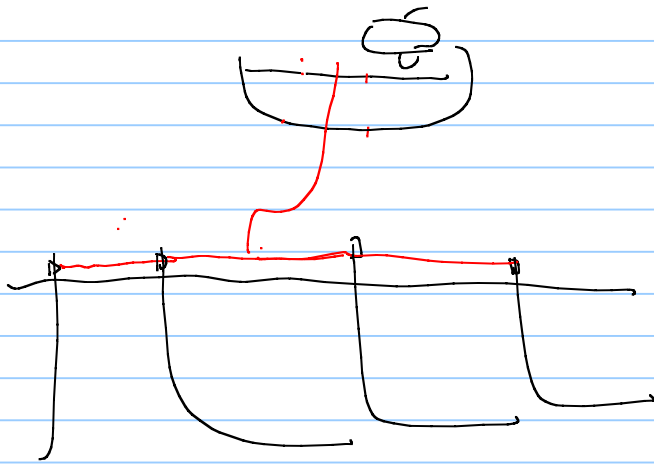
of MEG required such that the outlet of the cooler is outside the hydrate formation zone



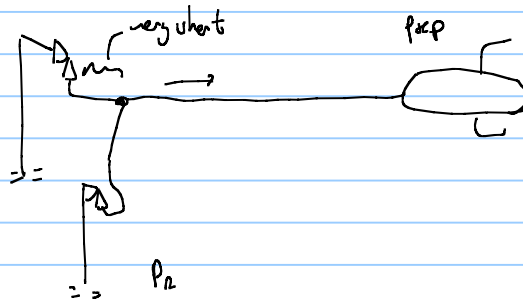
So approximately 162 kg/hr of MEG is needed

- Networks collection of pipes, flowline, pipeline, valves, pumps, take the fluids from wells to the processing facilities.





Example: 2 Dry gas wells

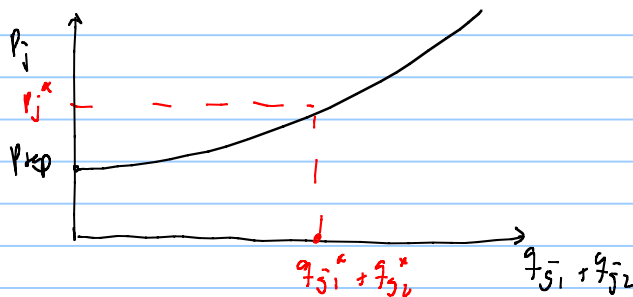
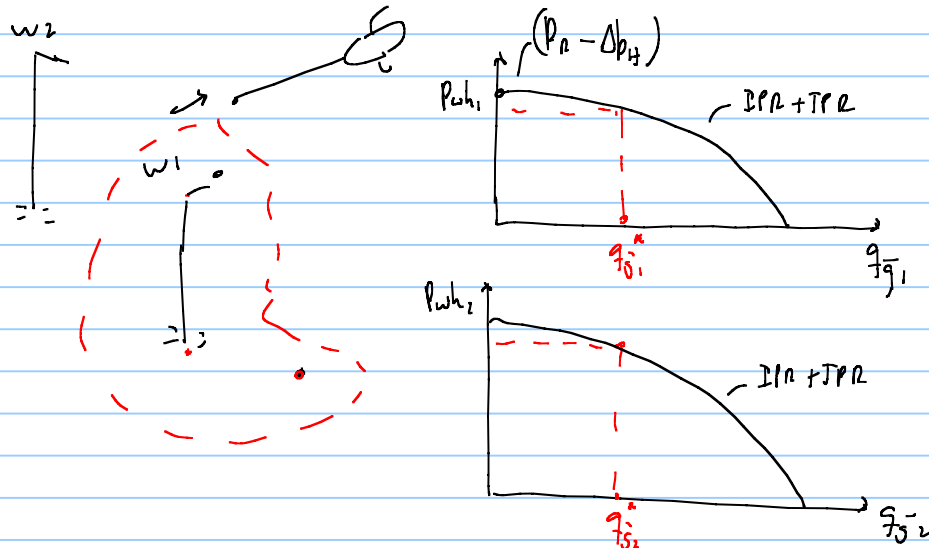
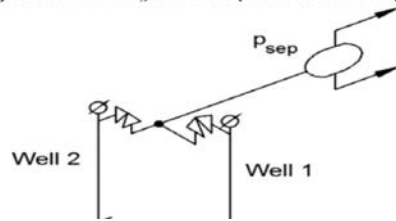


open down $\Delta p_{c1} = 0$ $\Delta p_{c2} = 0$

	Equations	No equations	No unknowns
IPR	$q_{j1} = C_{R1} (p_{a1}^2 - p_{wf1}^2)^{n_1}$ $q_{j2} = C_{R2} (p_{a2}^2 - p_{wf2}^2)^{n_2}$	2	4
TPR	$q_{j1} = C_{T1} \left(\frac{p_{a1}^2}{e^{j_1}} - p_{wh1}^2 \right)^{0.5}$ $q_{j2} = C_{T2} \left(\frac{p_{a2}^2}{e^{j_2}} - p_{wh2}^2 \right)^{0.5}$	2 4	2 6
PPR	$q_{j1} + q_{j2} = C_{P1} (p_j^2 - p_{sep}^2)^{0.5}$	1 5	1 7
	$\Delta p_{c1} = 0$ $p_{wh1} = p_j$	1 6	0 7
	$\Delta p_{c2} = 0$ $p_{wh2} = p_j$	1 7	0 7

PROBLEM 4 (18 POINTS). Network solving. (2017) exam

Consider the gas field with two wells, a manifold a pipeline and a separator shown in the figure below. The wellhead of the wells are very close to the junction so it can be safely assumed that the wellhead pressure and junction pressure are equal when the choke is open.



approach nr. 1

- 1: assume q_{j1}^* , q_{j2}^*
 - 2: Read p_{wh1}^* , p_{wh2}^* , p_j^*
(WPR₁) (WPR₂) (PPR)
 - 3: Verify $p_{wh1}^* = p_{wh2}^* = p_j^*$
- yes
 q_{j1}^* , q_{j2}^* are solution
- not

approach nr. 2

- 1: assume $p_j^* = p_{wh1}^* = p_{wh2}^*$
 - 2: Read q_{j1}^* (WPR₁), q_{j2}^* (WPR₂),
 q_{pipe}^* (PPR)
 - 3: Verify
 $q_{j1}^* + q_{j2}^* = q_{pipe}^*$
- yes
solution
- not

1st iteration

$$p_j = 50 \text{ bara}$$

$$q_{j1}^* = 1.2 \text{ E}06 \text{ m}^3/\text{d}$$

$$q_{j2}^* = 1.57 \text{ E}06 \text{ m}^3/\text{d} + 2.77 \text{ E}06 \text{ m}^3/\text{d} \quad \varepsilon = 0.97 \text{ E}06 \text{ m}^3/\text{d}$$

$$q_{\text{pipe}}^* = 1.8 \text{ E}06 \text{ m}^3/\text{d} \quad 1.8 \text{ E}06 \text{ m}^3/\text{d}$$

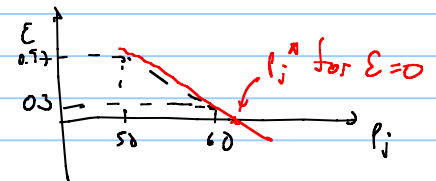
2nd

$$p_j = 60 \text{ bara}$$

$$q_{\text{pipe}}^* = 2.3 \text{ E}06 \text{ m}^3/\text{d} \quad 2.3 \text{ E}06 \text{ m}^3/\text{d} \quad \varepsilon = 0.3 \text{ E}06 \text{ m}^3/\text{d}$$

$$q_{j2}^* = 1.45 \text{ E}06 \text{ m}^3/\text{d} + 2.60 \text{ E}06 \text{ m}^3/\text{d}$$

$$q_{j1}^* = 1.15 \text{ E}06 \text{ m}^3/\text{d}$$



$$\frac{\varepsilon_1 - \varepsilon_2}{p_{j1} - p_{j2}} = \frac{\varepsilon_1 - 0}{p_{j1} - p_j^*}$$

$$p_j^* = \sim$$