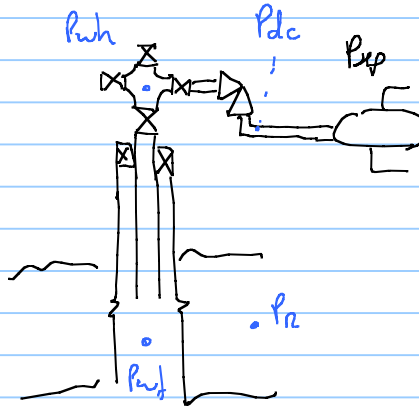


Notes for Youtube video nr. 6

Simplified dry gas production system



Reservoir model



Dry gas material balance

$$P_R = P_{Ri} \frac{z_R}{z_i} \left(1 - \frac{G_p}{G} \right)$$

$q_i = f(t)$

R_F recovery factor

gas deviation factor

$$z = f(P_R, T_R)$$

$\frac{T_R}{T_c}$

$\frac{P_R}{P_c} \rightarrow f(\text{gas composition})$

MB dry gas equation is implicit

- Given R_F , assume P_R
- with P_R compute z_R
- verify that $\varepsilon = P_R - P_{Ri} \frac{z_R}{z_i} (1 - R_F) = 0 \leq \text{Tolerance}$
- if not,

3.3.2 Z-Factor Correlations. Standing and Katz⁴ present a generalized Z-factor chart (**Fig. 3.6**), which has become an industry standard for predicting the volumetric behavior of natural gases. Many empirical equations and EOS's have been fit to the original Standing-Katz chart. For example, Hall and Yarborough^{21,22} present an

accurate representation of the Standing-Katz chart using a Carnahan-Starling hard-sphere EOS,

$$Z = ap_{pr}/y, \dots \dots \dots (3.42)$$

where $\alpha = 0.06125t \exp[-1.2(1-t)^2]$, where $t = 1/T_{pr}$.

The reduced-density parameter, y (the product of a van der Waals covolume and density), is obtained by solving

$$f(y) = 0 = -ap_{pr} + \frac{y + y^2 + y^3 - y^4}{(1-y)^3} - (14.76t - 9.76t^2 + 4.58t^3)y^2 + (90.7t - 242.2t^2 + 42.4t^3)y^{2.18+2.82t}, \dots \dots \dots (3.43)$$

$$\text{with } \frac{df(y)}{dy} = \frac{1 + 4y + 4y^2 - 4y^3 + y^4}{(1-y)^4} - (29.52t - 19.52t^2 + 9.16t^3)y + (2.18 + 2.82t)(90.7t - 242.2t^2 + 42.4t^3) \times y^{1.18+2.82t}, \dots \dots \dots (3.44)$$

The derivative $\partial Z/\partial p$ used in the definition of c_g is given by

$$\left(\frac{\partial Z}{\partial p}\right)_T = \frac{\alpha}{p_{pc}} \left[\frac{1}{y} - \frac{ap_{pr}/y^2}{df(y)/dy} \right], \dots \dots \dots (3.45)$$

Equation approximation to Z chart

to predict T_c, p_c we will use

Sutton correlations

Sutton⁷ suggests the following correlations for hydrocarbon gas mixtures.

$$T_{pcHC} = 169.2 + 349.5\gamma_{gHC} - 74.0\gamma_{gHC}^2 \dots \dots \dots (3.47a)$$

$$\text{and } p_{pcHC} = 756.8 - 131\gamma_{gHC} - 3.6\gamma_{gHC}^2 \dots \dots \dots (3.47b)$$

$$\gamma_g = \frac{M_{wgas}}{M_{wair}(28.97)}$$

$$M_{wgas} = \sum_{i=1}^N z_i M_{wi}$$

$P_n \rightarrow P_{nf}$

IPR equation

low-pressure dry gas equation

$$q_g = C_R (P_n^2 - P_{nf}^2)^n \quad \leftarrow \text{backpressure exponent}$$

laminar $n \sim 1$
turbulent $n \sim 0.5$

inflow coefficient $\{ T_R, K, h, s \text{ (skin factor)}$



= pseudosteady state regime
(boundary dominated flow)
page 37 of compendium

• $p_{wf} \rightarrow p_{wh}$

Dry gas tubing equation

$$q_g = C_T \left(\frac{p_{wf}^2}{e^S} - p_{wh}^2 \right)^{0.5}$$

tubing coefficient (friction losses) $\rightarrow$ elevation coefficient (hydrostatic losses)

$$q_g = 0$$

$$p_{wf} = p_{wh} e^{S/2}$$

Page 156, Appendix A of compendium

$$q_{sc} = \left(\frac{\pi}{4} \right) \cdot \left(\frac{R}{M_{air}} \right)^{0.5} \cdot \left(\frac{T_{sc}}{p_{sc}} \right) \cdot \left(\frac{D^5}{f_M \cdot L \cdot \gamma_g \cdot Z_{av} \cdot T_{av}} \right)^{0.5} \cdot \left[(p_{wf}^2 - p_t^2 \cdot e^S) \cdot \left(\frac{S}{e^S - 1} \right) \right]^{0.5}$$

$$C_T = \left(\frac{\pi}{4} \right) \cdot \left(\frac{R}{M_{air}} \right)^{0.5} \cdot \left(\frac{T_{sc}}{p_{sc}} \right) \cdot \left(\frac{D^5}{f_M \cdot L \cdot \gamma_g \cdot Z_{av} \cdot T_{av}} \right)^{0.5} \cdot \left(\frac{S \cdot e^S}{e^S - 1} \right)^{0.5}$$

$$S = 2 \cdot L \cdot C_a = 2 \cdot \frac{M_g}{Z_{av} \cdot R \cdot T_{av}} \cdot L \cdot g \cdot \cos(\alpha)$$

Comments about friction equation

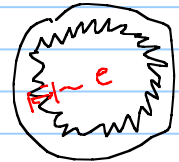
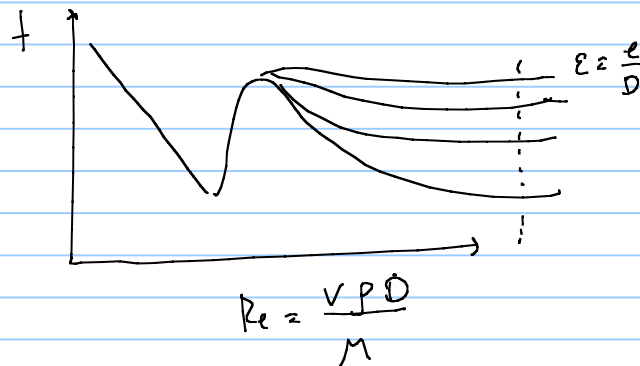
to compute G

$$T_{av} \rightarrow \frac{T_{wf} + T_{wh}}{2}$$

An estimate of T_{wh} is needed

$$Z_{av} \rightarrow \frac{Z_{wf} + Z_{wh}}{2}$$

f_m friction factor



$V \approx f(q_{local})$ for gas $V \uparrow \uparrow$ ρ is low compared
 $q_{local} \uparrow (p)$ liquid $V = [0.5 - 4] \text{ m/s}$ to liquid
 gas $V = [5 - 40] \text{ m/s}$

M_g is $\ll M_L$

$Re_g \gg$

always in fully
turbulent regime

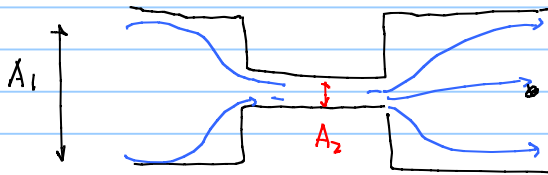
$f_m = f(\epsilon)$ however $\epsilon = f(C.D)$
 due to manufacturing

have equation for dry gas: (page 166)

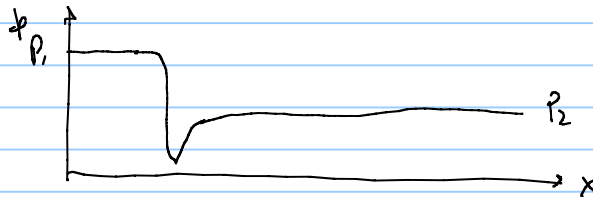
$$q_{\bar{g}} = \frac{p_1 \cdot T_{sc} \cdot A_2 \cdot C_d}{p_{sc}} \cdot \sqrt{2 \cdot \frac{R}{Z_1 \cdot T_1 \cdot M_w} \cdot \frac{k}{k-1} \cdot \left(y^{\frac{2}{k}} - y^{\frac{k+1}{k}} \right)}$$

$p_{sc} = 1.01325 \text{ bara}$
 $T_{sc} = 15.56^\circ \text{C}$

$y = \frac{p_2}{p_1}$ (downstream)
 $\bar{p}_1$ (upstream)



if $y > y_c \approx (0.6)$, there is critical flow at the throat



if $y > y_c$ $q_{\bar{g}} = q_{\bar{g}_c} =$

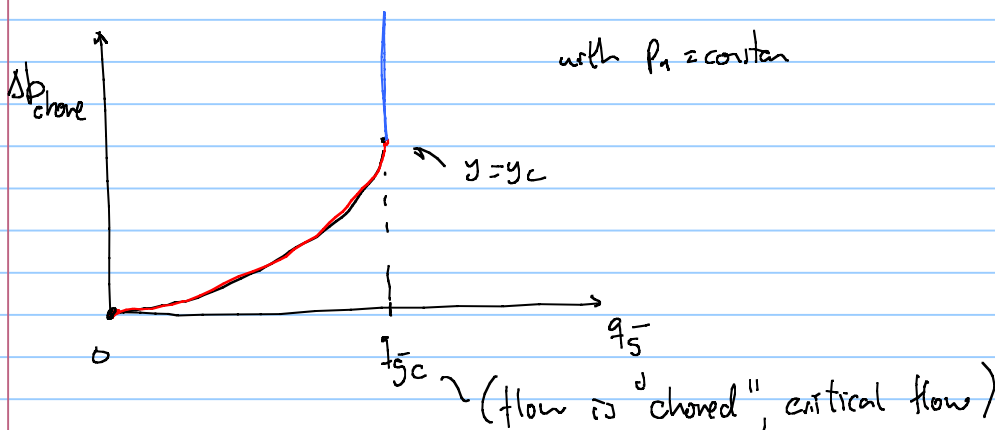
$$q_{\bar{g}} = \frac{p_1 \cdot T_{sc} \cdot A_2 \cdot C_d}{p_{sc}} \cdot \sqrt{2 \cdot \frac{R}{Z_1 \cdot T_1 \cdot M_w} \cdot \frac{k}{k-1} \cdot \left(y_c^{\frac{2}{k}} - y_c^{\frac{k+1}{k}} \right)}$$

in blue

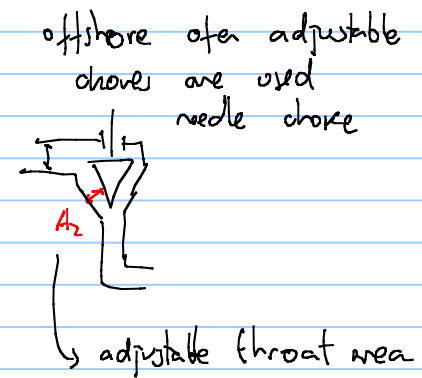
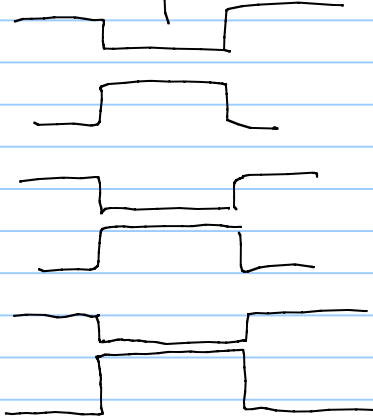
if $y < y_c$

$$q_{\bar{g}} = \frac{p_1 \cdot T_{sc} \cdot A_2 \cdot C_d}{p_{sc}} \cdot \sqrt{2 \cdot \frac{R}{Z_1 \cdot T_1 \cdot M_w} \cdot \frac{k}{k-1} \cdot \left(y^{\frac{2}{k}} - y^{\frac{k+1}{k}} \right)}$$

in red



in onshore fields, bean chokes are often used
 given in $\frac{1}{64}"$



→ P_{sep} flowline → tubing equation can be used for flowline

horizontal flowline, the tubing equation simplifies to

$$q_s = C_{FL} (P_{dc}^2 - P_{sep}^2)^{0.5}$$

$$S=0 \text{ (L'Hopital)}$$

VBA Visual basic for applications

for pipe equations in VBA ① is upstream
 ② is downstream

