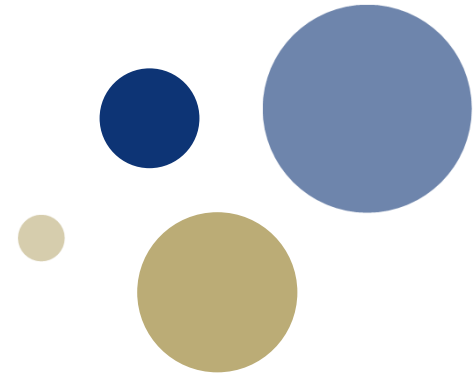




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Science and Technology



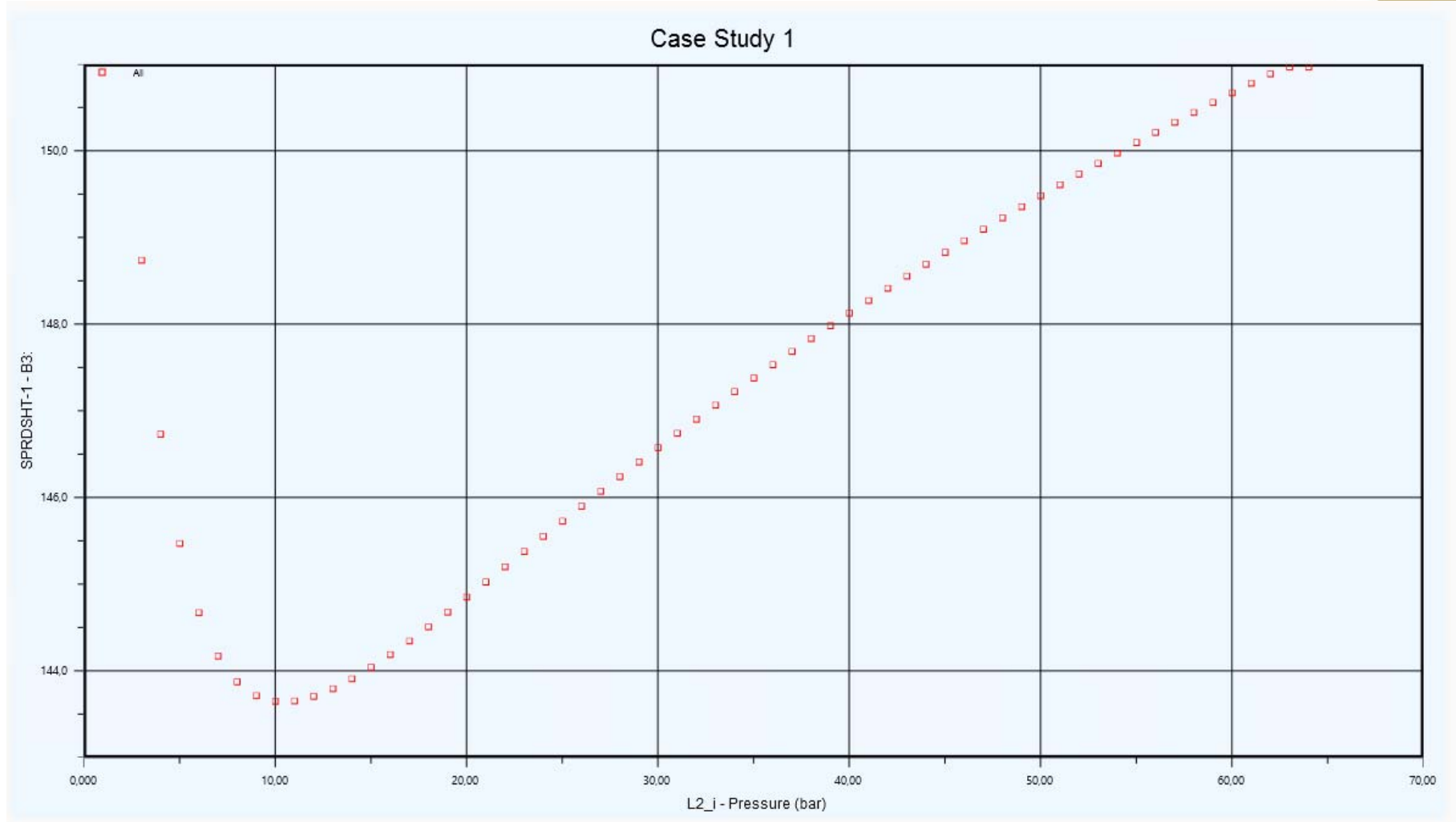
Production Technology

Field Processing and Systems

Postdoc Mariana Díaz
01 /22/2019



Summary



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Multivariable Case Studies



Flowsheet Case (Main) - Solver Active x Case Study 1 x Case Study 2 x +

Variable Selection Case Study Setup Results Plots

Independent Variables Find Variables

Name	Tag	Current Value	Units	Delete
L3_i - Pressure		3,000	bar	X
L3_i - Temperature		19,00	C	X
L2_i - Pressure		20,00	bar	X
L2_i - Temperature		65,00	C	X

Dependent Variables Find Variables

Name	Tag	Current Value	Units	Delete
GOR - B3:		146,2		X

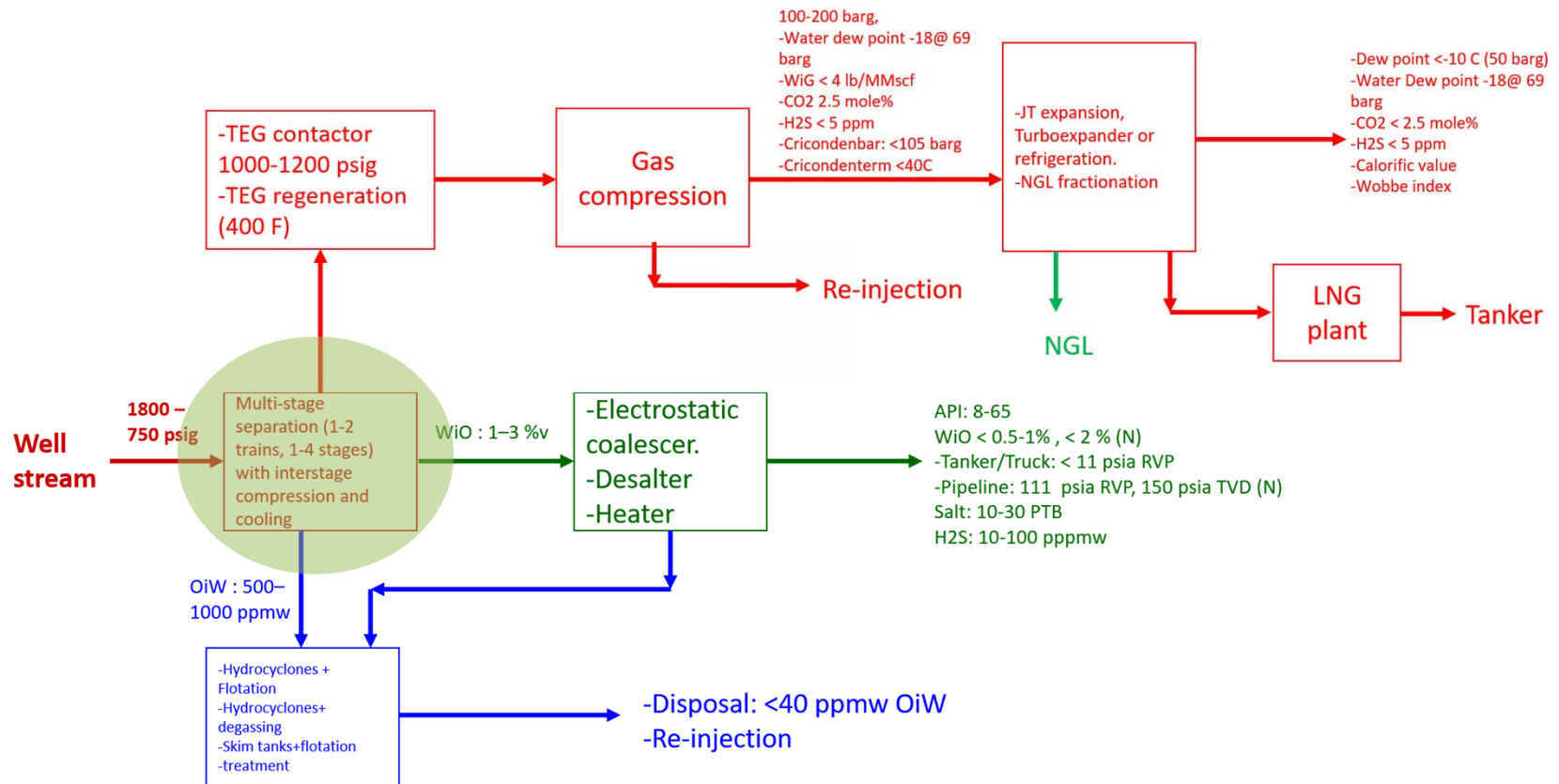
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SEPARATION

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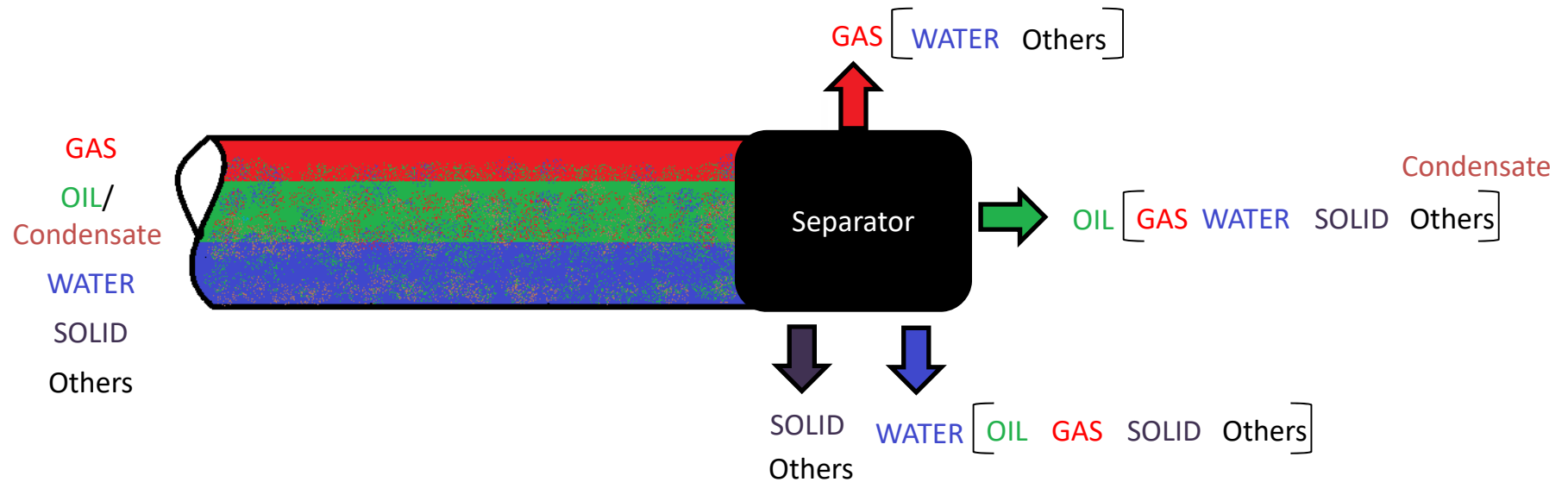
Processing Block Diagram



Stanko Milan (2018)

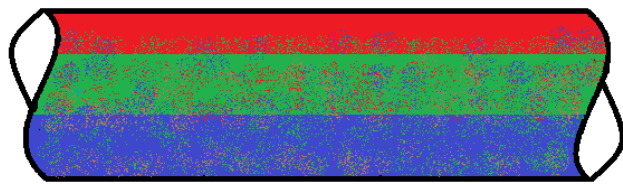
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Separation in Oil and Gas Production



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Characteristics of the flow stream



- Type of fluids/phases



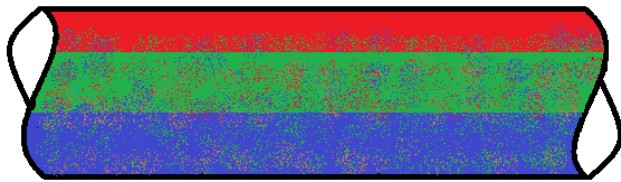
- Type of wells

Dry gas Gas Water

Gas condensate Condensate
Water

Crude oil OIL
Gas
Water
Solid

Characteristics of the flow stream



- **Flow mixture**

Heterogeneous mixtures



Homogeneous mixtures

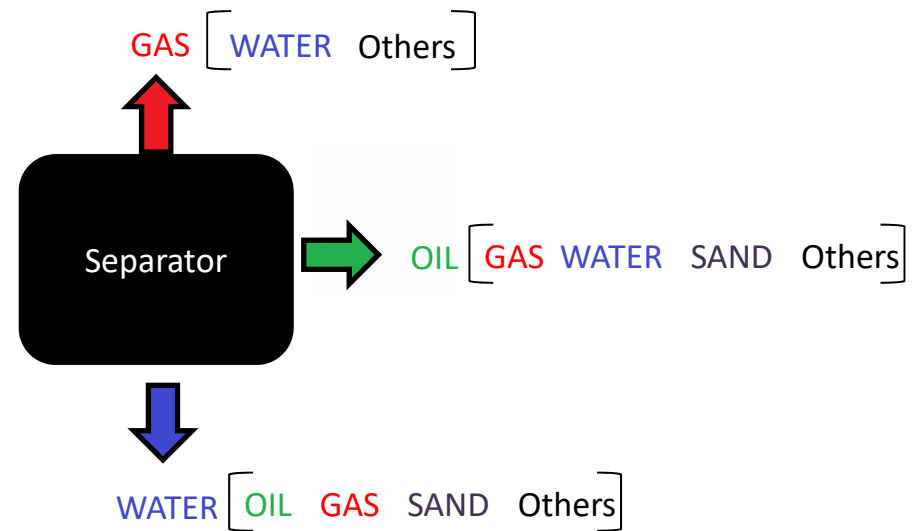
- Emulsion
- Foam



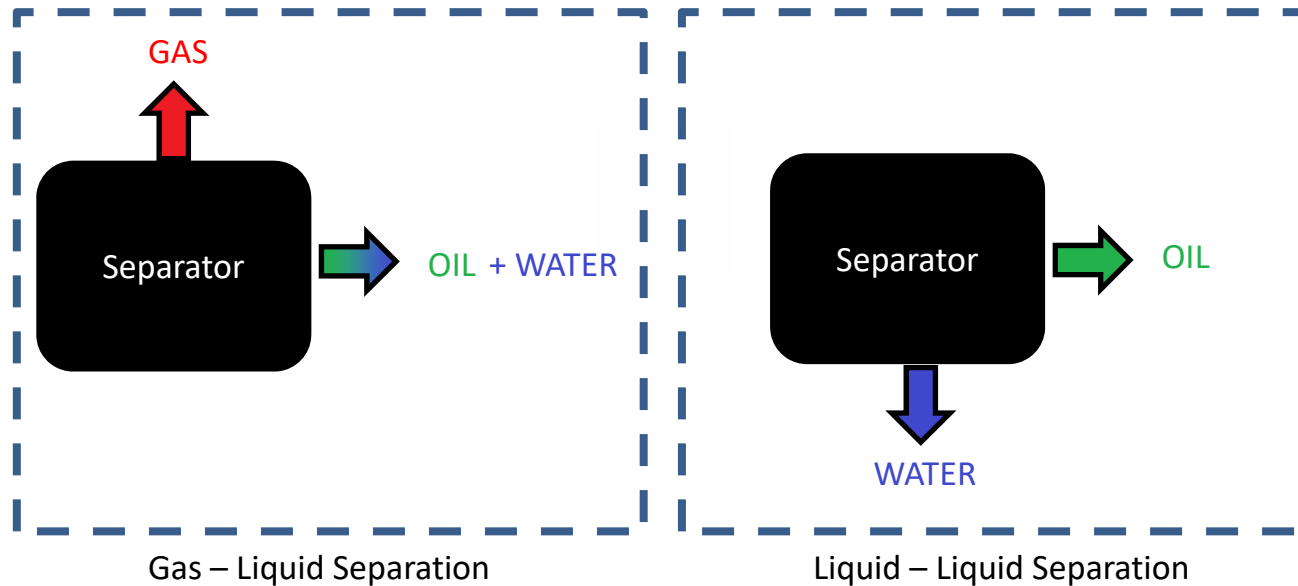
- **Separation variables**

Pressure
Temperature
Density
Composition

Three phase separation



Two phase separation

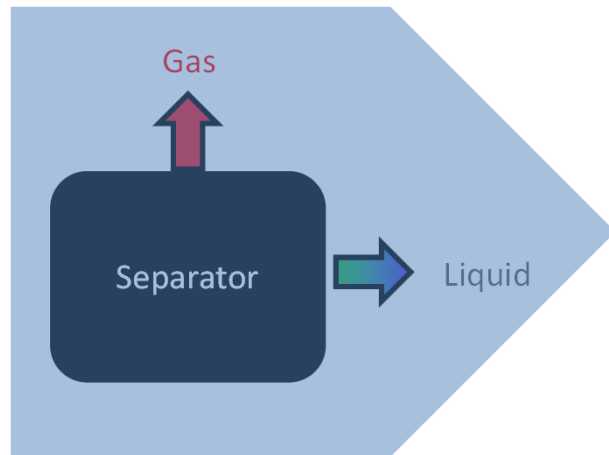


Factor affecting separation



- Gas and liquid flow rate
- Operating and design pressure and temperature
- Slugging or surging
- Physical properties (density, compressibility,...)
- Presence of impurities (paraffin, san, scale,...)
- Foaming tendencies of the oil
- Corrosive tendencies of the liquids or gas

Gas-Liquid Separation

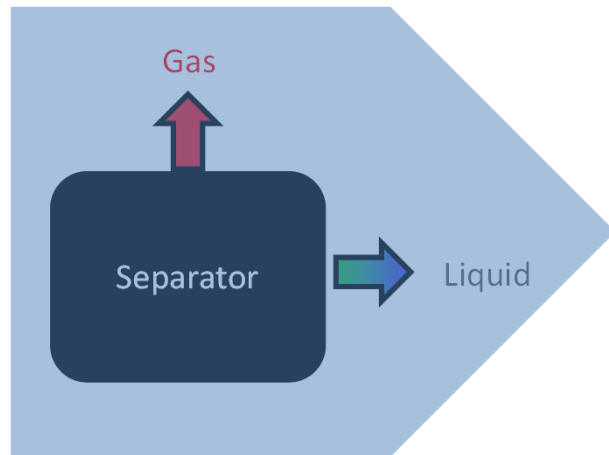


- Gravity tank separators
- Scrubber
- Slug Catcher
- Cyclonic

Gravity Forces

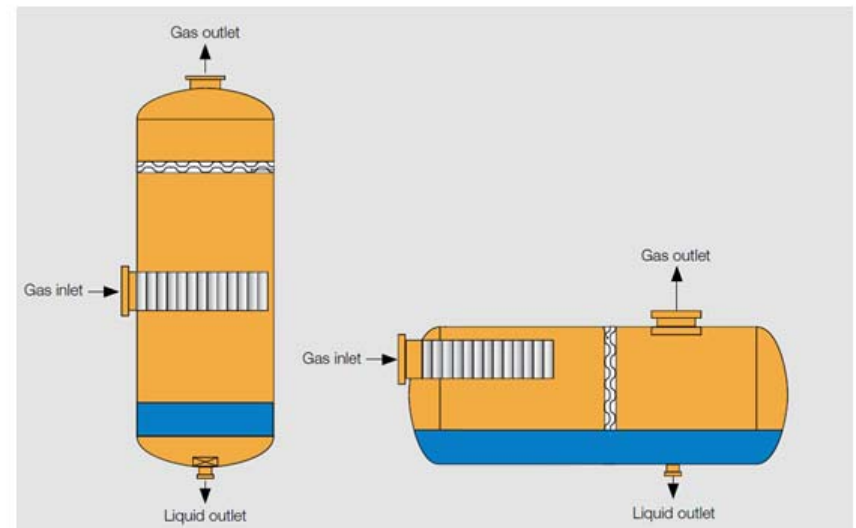
Centrifugal Forces

Gas-Liquid Separation



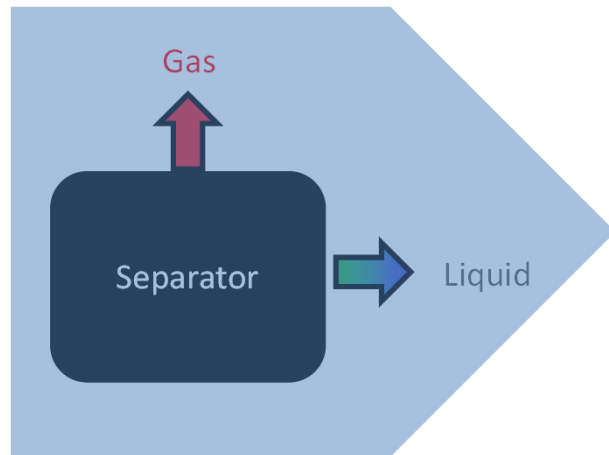
- Gravity tank separators

- Scrubber
- Slug Catcher
- Cyclonic

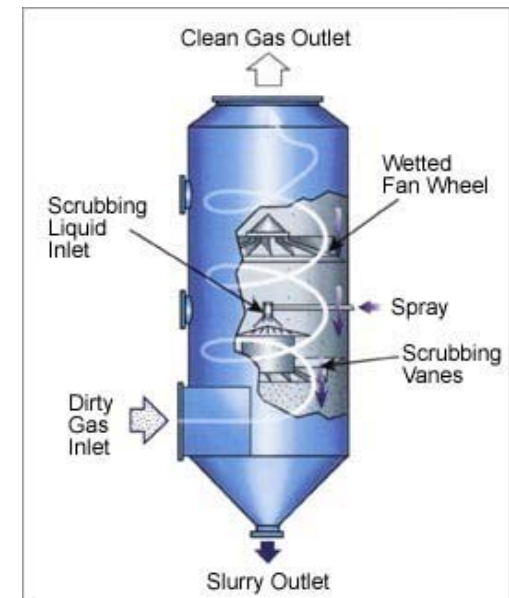


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Gas-Liquid Separation

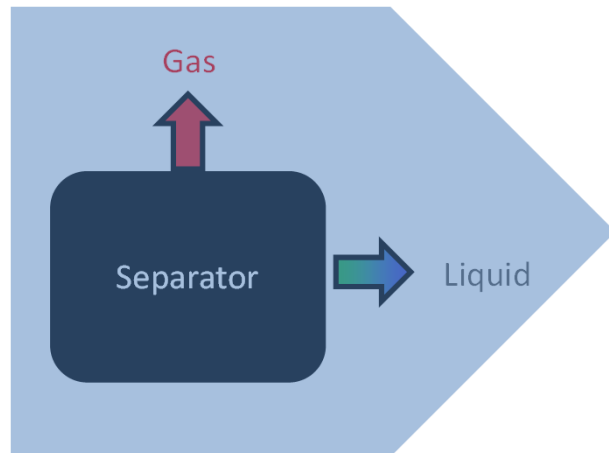


- Gravity tank separators
- **Scrubber**
- Slug Catcher
- Cyclonic

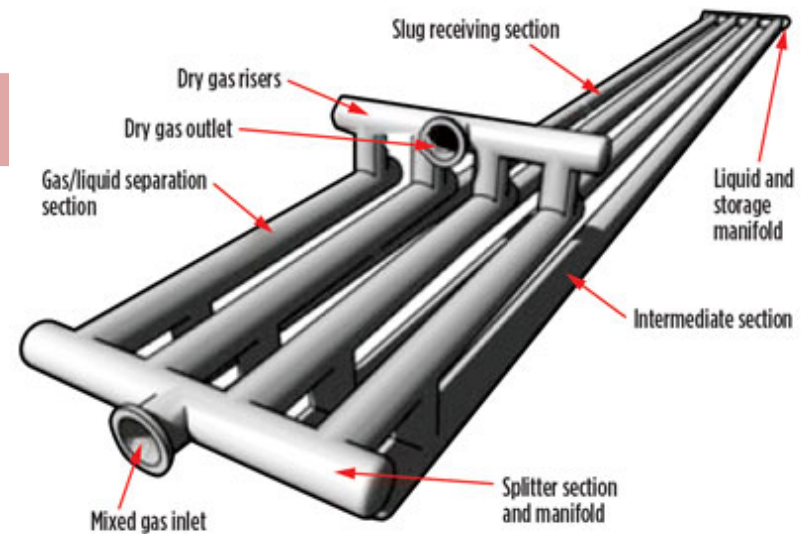


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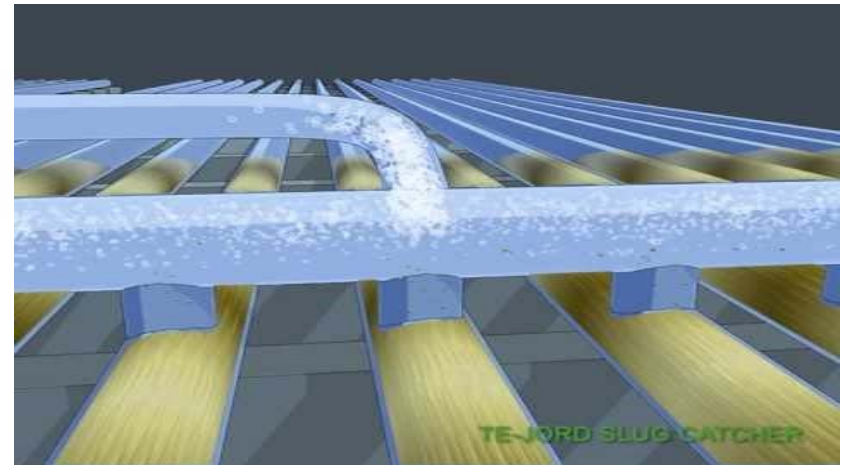
Gas-Liquid Separation



- Gravity tank separators
- Scrubber
- **Slug Catcher**
- Cyclonic

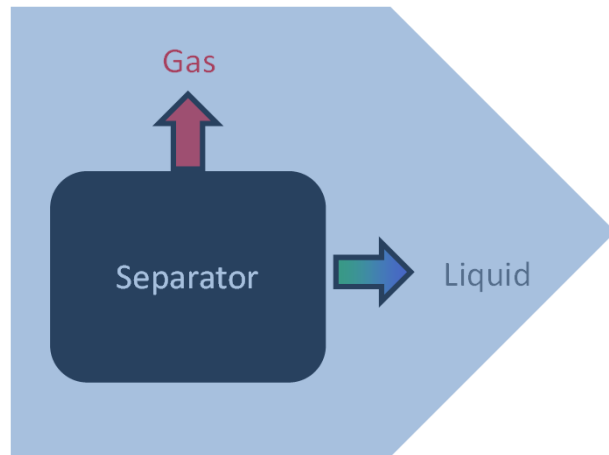


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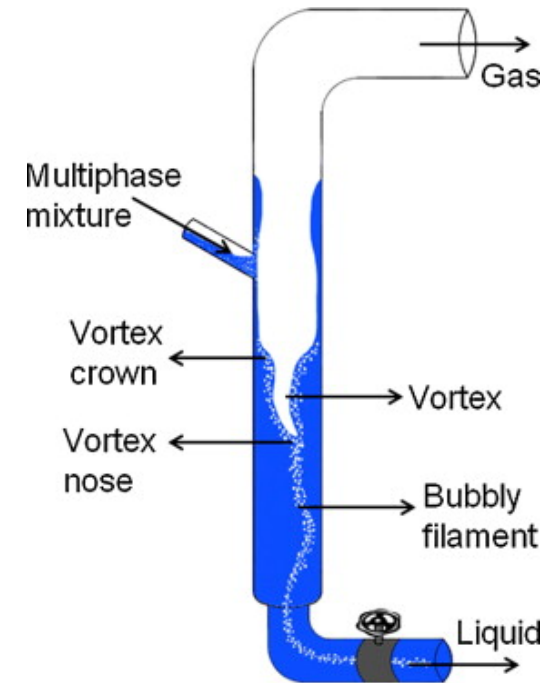


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Gas-Liquid Separation

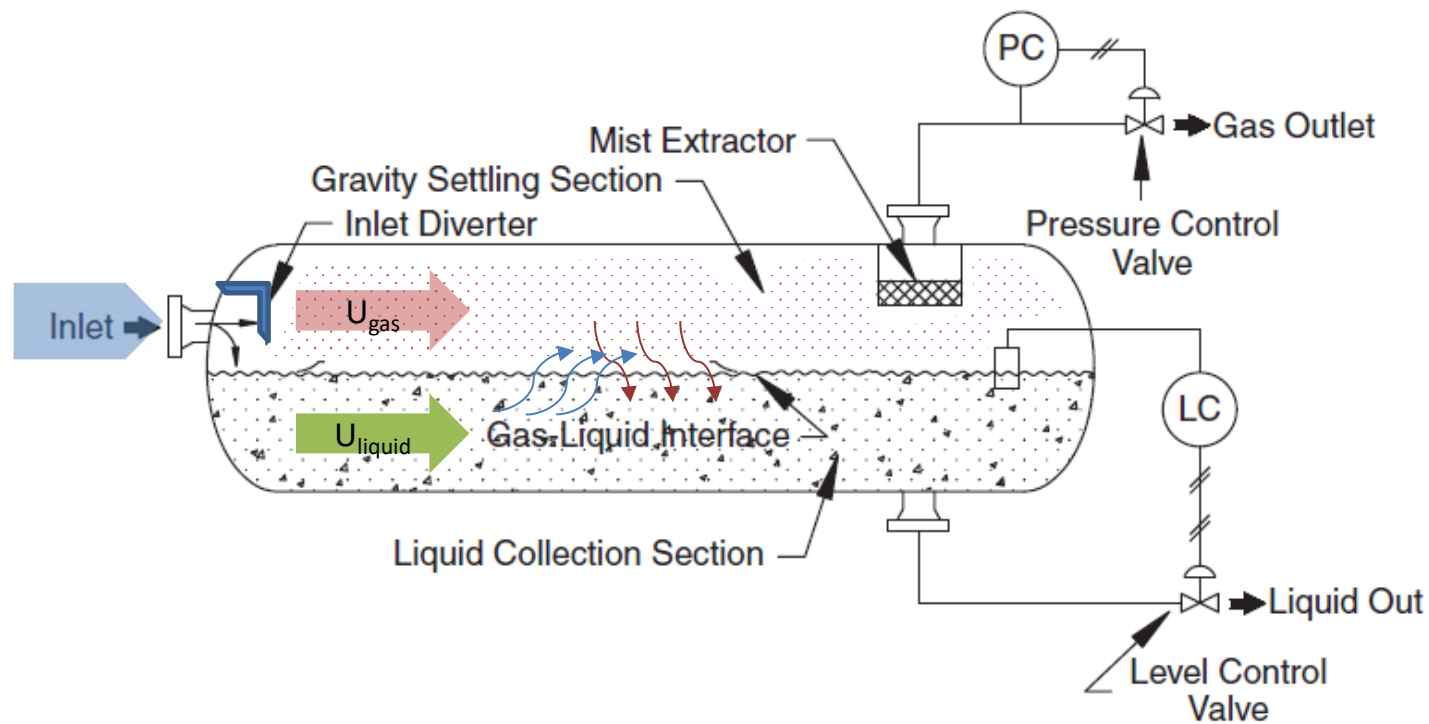


- Gravity tank separators
- Scrubber
- Slug Catcher
- **Cyclonic**



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Gravity tank separator



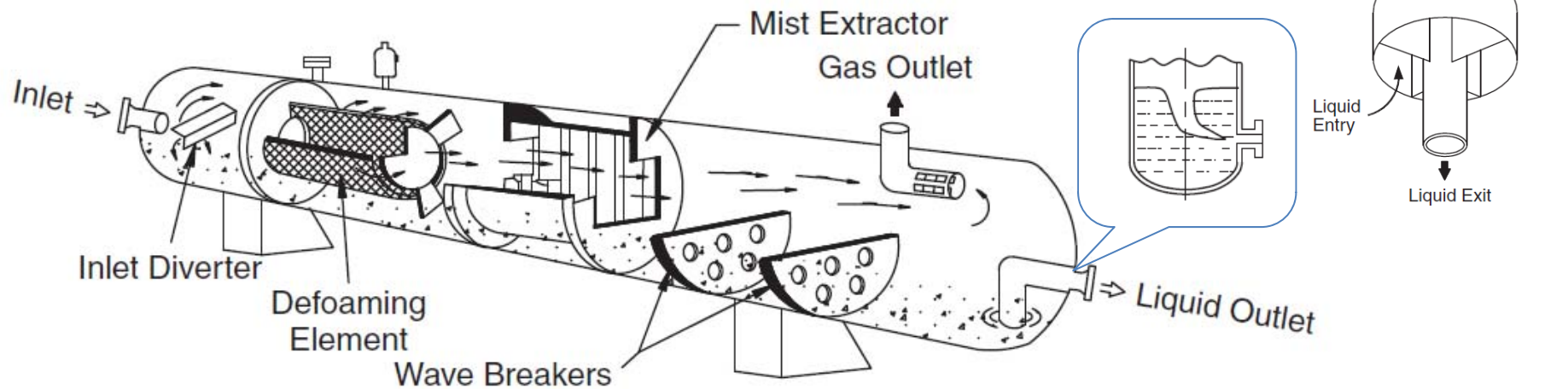
(Stewart and Arnold, 2008)

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Gravity tank separator



Internal elements



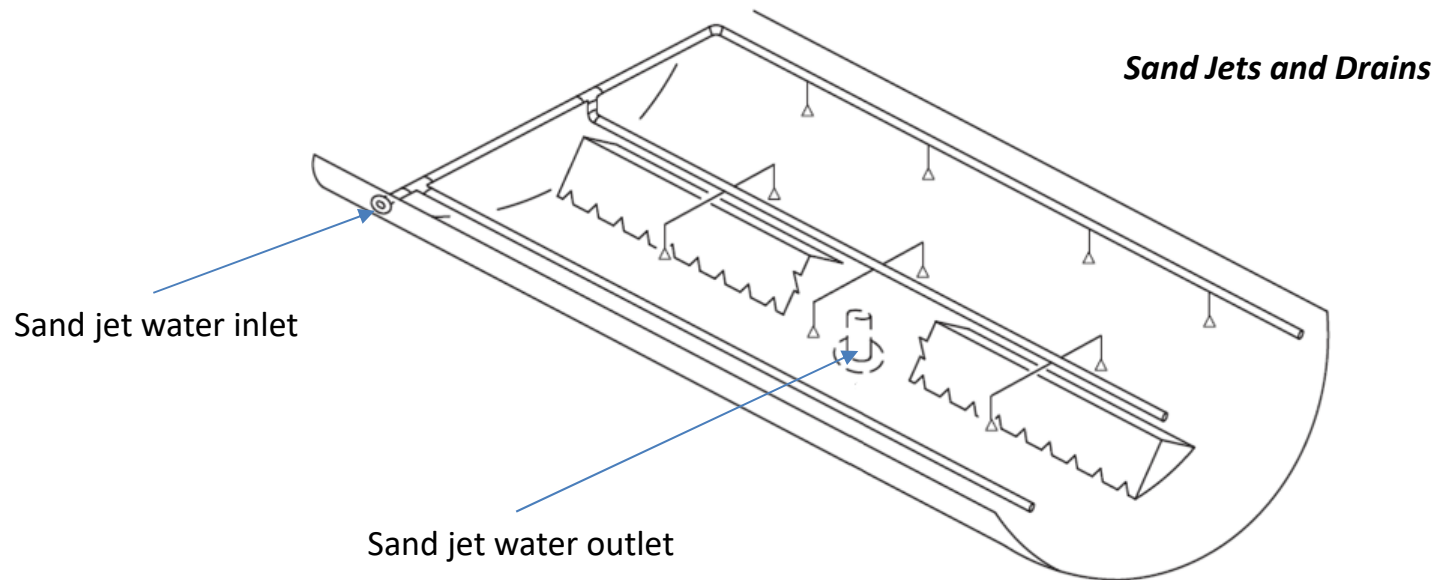
(Stewart and Arnold, 2008)

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Gravity tank separator



Internal elements



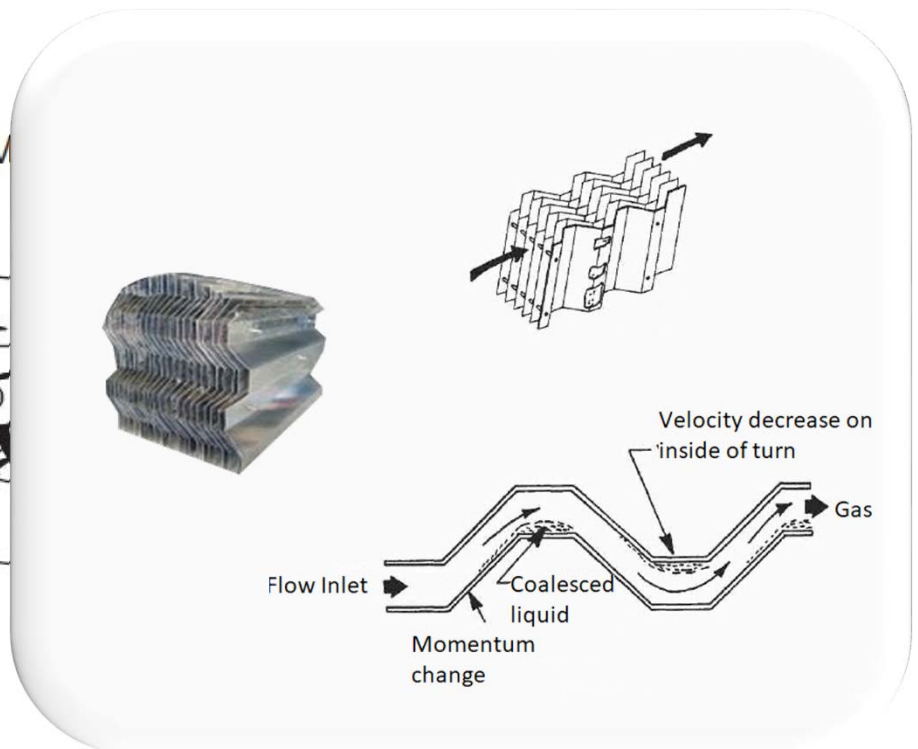
(Stewart and Arnold, 2008)

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Gravity tank separator

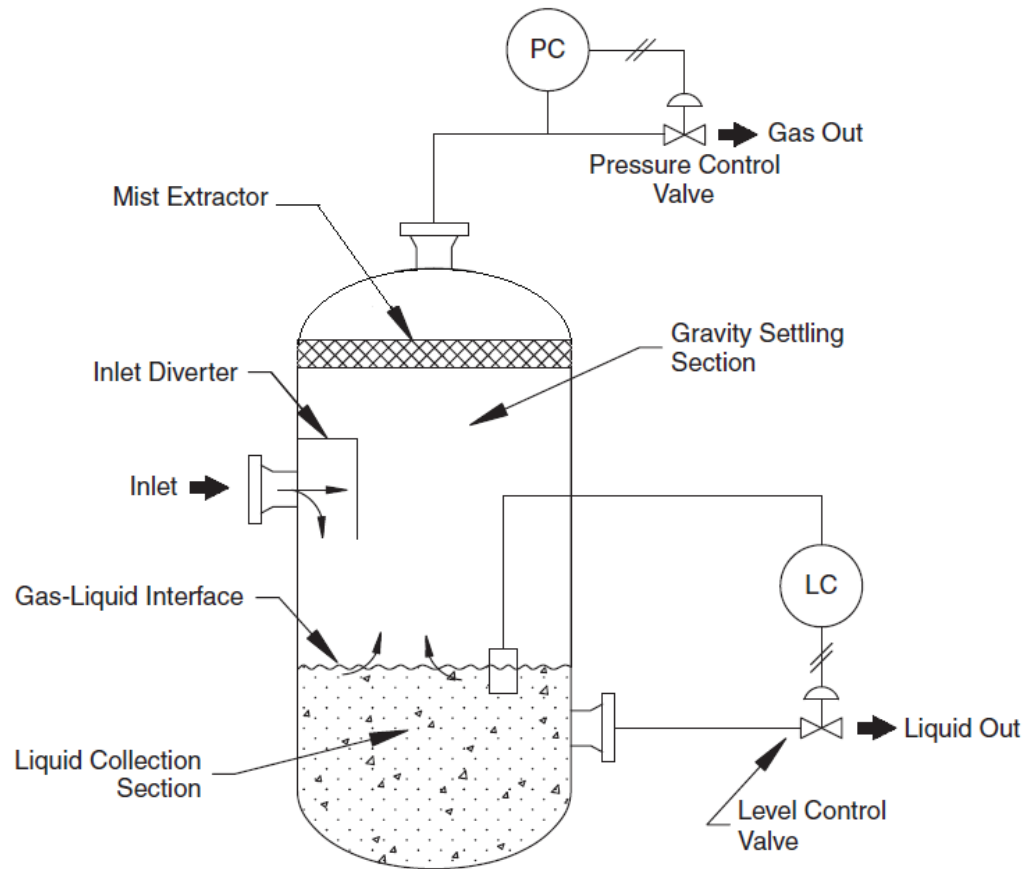


Internal elements



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Vertical Tank

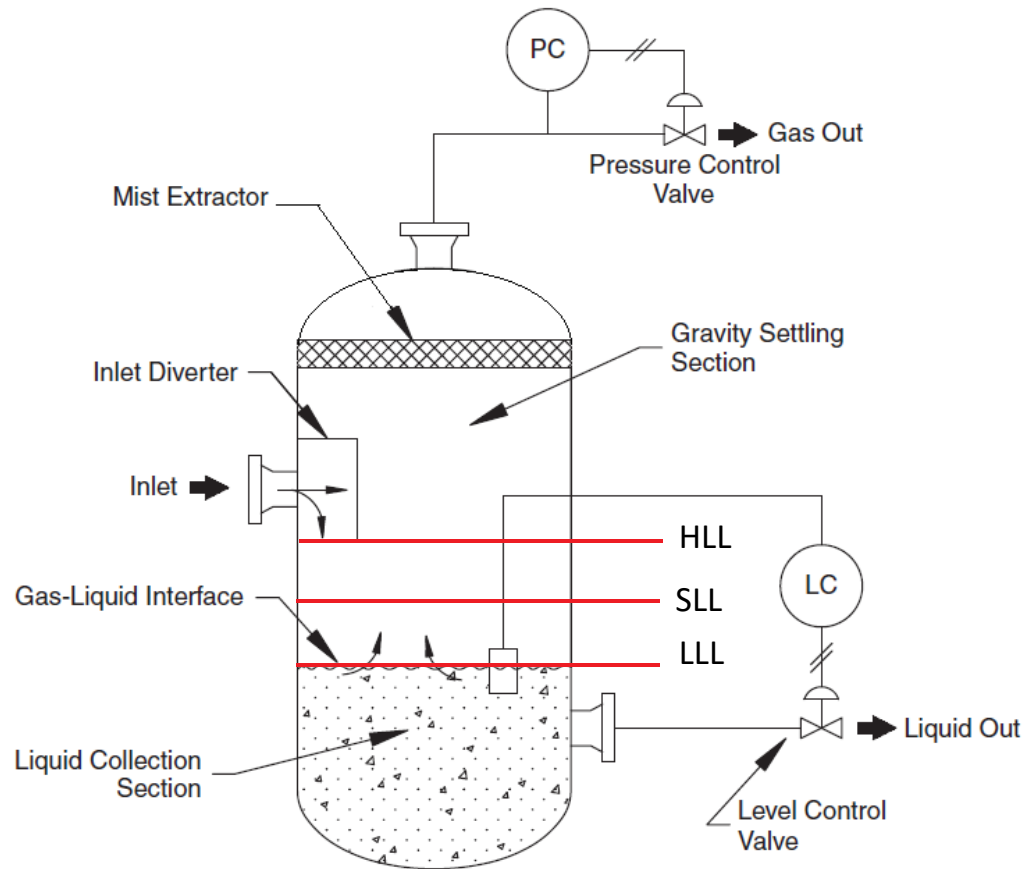


(Stewart and Arnold, 2008)



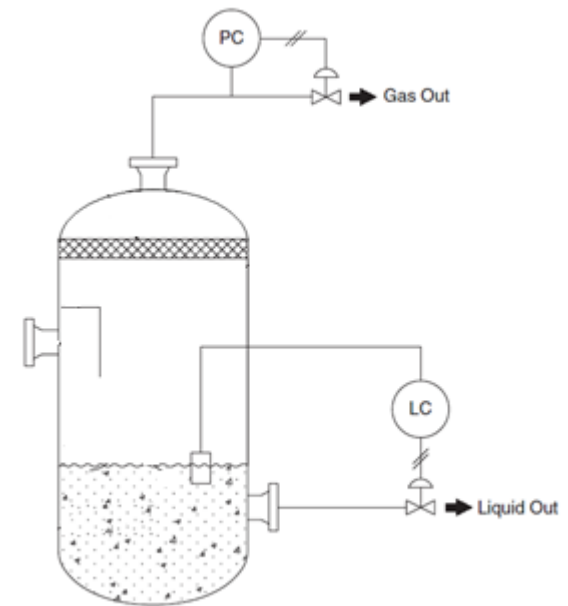
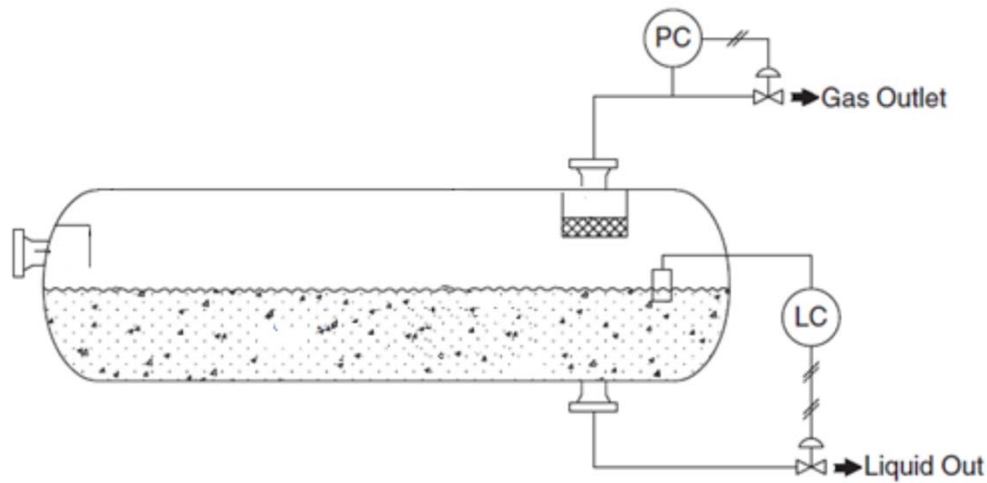
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Control system



(Stewart and Arnold, 2008)

Horizontal vs Vertical



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Horizontal vs Vertical



Horizontal



- Smaller than vertical tanks for a given gas-liquid flow rate
- Commonly used for high gas-liquid ratios and foaming crude
- The interface area is large in horizontal than verticals so it is easier for the gas to come out of the liquid and reach the vapor space



- They are not so good as vertical for solid handling
- Can have less liquid surge than vertical vessels
- surges in horizontal vessels could create internal waves, which could activate the high-level sensor prematurely
- Harder to clean

Vertical



- Commonly used for low to intermediate gas-liquid ratio.
- Suited for production containing solids and sediments
- Save space
- Less tendency for re-evaporation of liquid into the gas



- Wall might need to be thicker due to the distribution of supports
- some relief valves and controls systems may be difficult to service without special ladders and platforms

*High gas-liquid ratio, a vertical separator is a scrubber

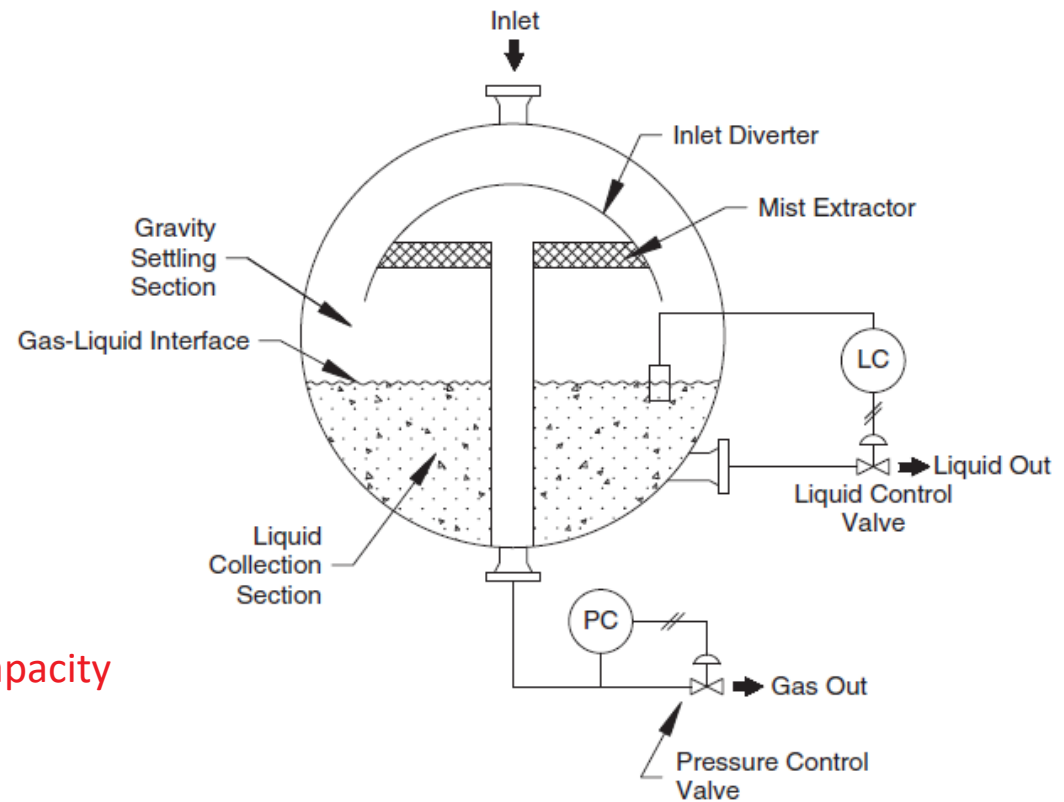
Spherical Separator



- Very efficient pressure containers
- Very compact



- Very difficult to operate and size
- They have limited liquid surge capacity
- Difficult to fabricate
- Limited liquid settling section
- Liquid level control is critical



(Stewart and Arnold, 2008)

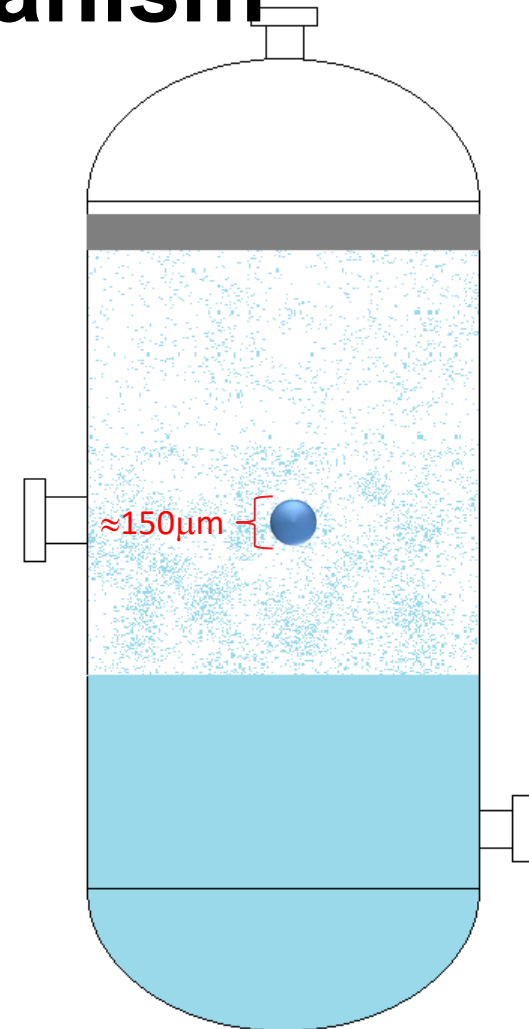
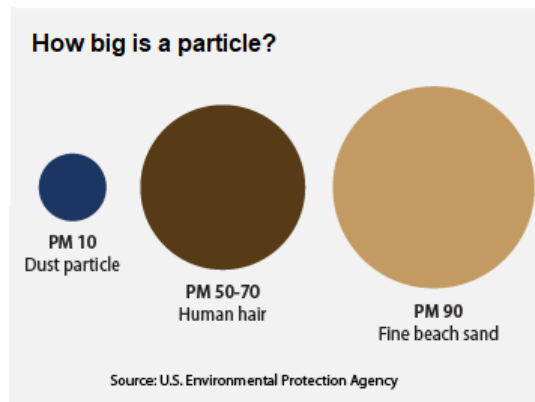
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Bubble and Droplet Mechanism

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Bubble and Droplet Mechanism



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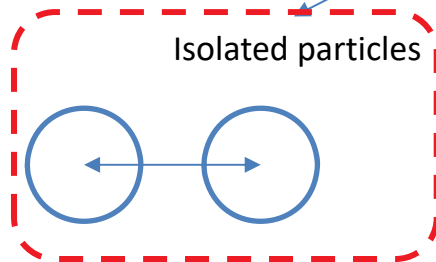
Bubble and Droplet Mechanism



Continuous **vs** disperse flow

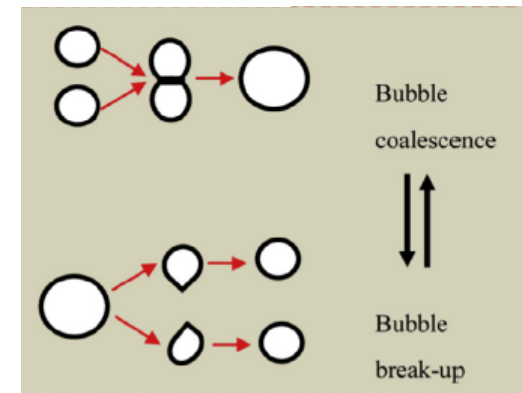
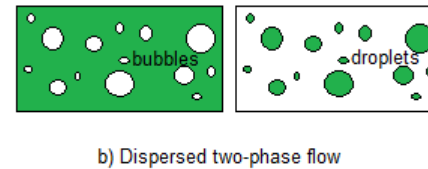
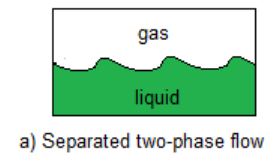
Continuous phase

Disperse phase



Particle-Particle interaction

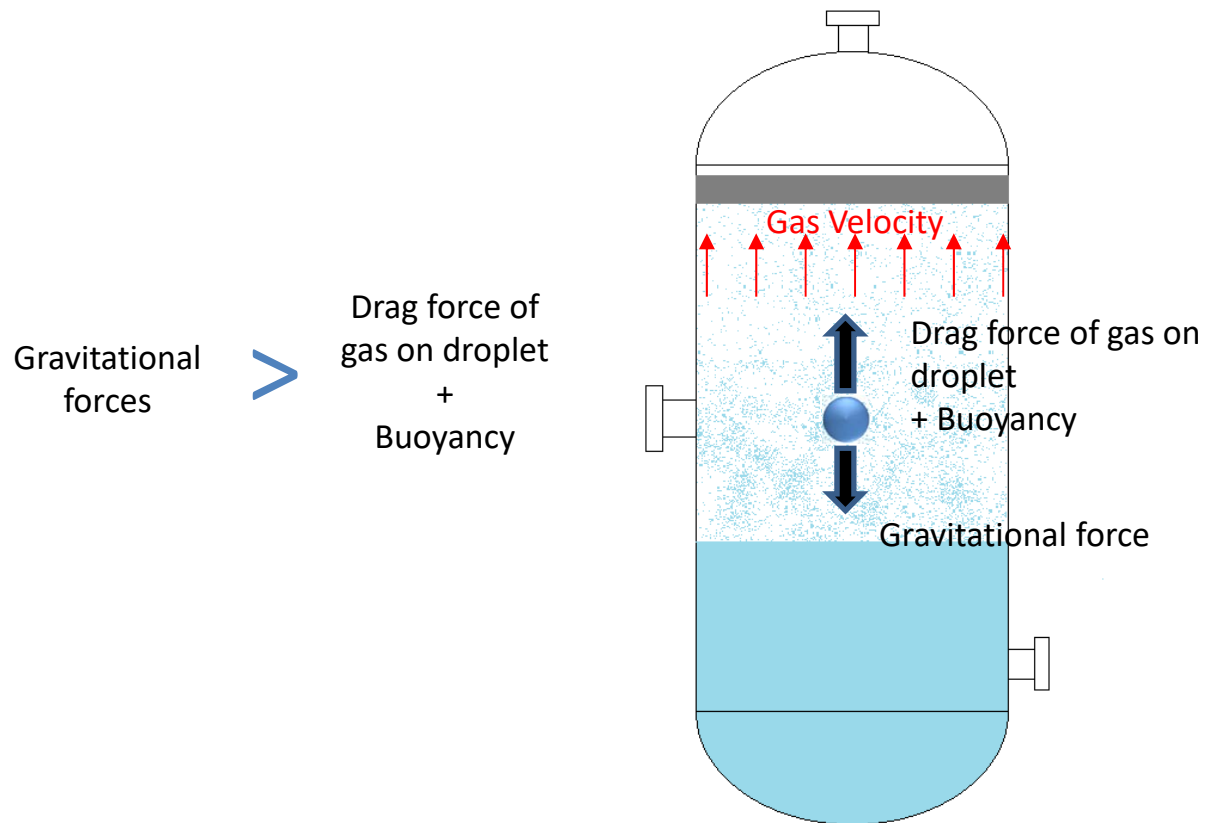
- Momentum
- Heat transfer



(Xu et al., 2015)

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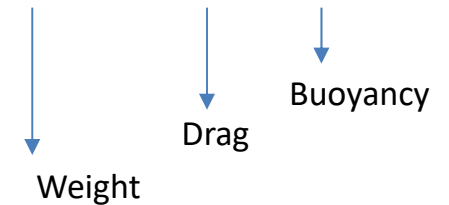
Bubble and Droplet Mechanism



Terminal Velocity (V_t)



$$F_g = F_d + F_b$$



Bubble and Droplet Mechanism



Terminal Velocity (V_t)



$$V_t = \sqrt{\frac{4 D_P g (\rho_P - \rho_f)}{3 C_d \rho_f}}$$

$$F_g = F_d + F_b$$

$$F_g = M_P g = \frac{\pi D_P^3}{6} \rho_P g$$

$$F_d = C_d A_P \underbrace{\frac{1}{2} \rho_f V_t^2}_{\text{Dynamic pressure}}$$

$$F_b = Vol_P \rho_f g = \frac{\pi D_P^3}{6} \rho_f g$$

Bubble and Droplet Mechanism



$$V_t = \sqrt{\frac{4 D_P g (\rho_P - \rho_f)}{3 C_d \rho_f}}$$

1. Droplet settling theory

- Trial and error solution
- Graphical solution

2. Souders-Brown approach

$$K_{SB} = \sqrt{\frac{4 D_P g}{3 C_d}} = V_t \sqrt{\frac{\rho_f}{(\rho_P - \rho_f)}}$$

f(mist extractor, separator geometry, flow rates, fluid properties)

Bubble and Droplet Mechanism



1. Droplet settling theory

$$V_t = \sqrt{\frac{4 D_P g (\rho_P - \rho_f)}{3 C_d \rho_f}}$$



Trial and error solution

$$C_d = f(Re)$$



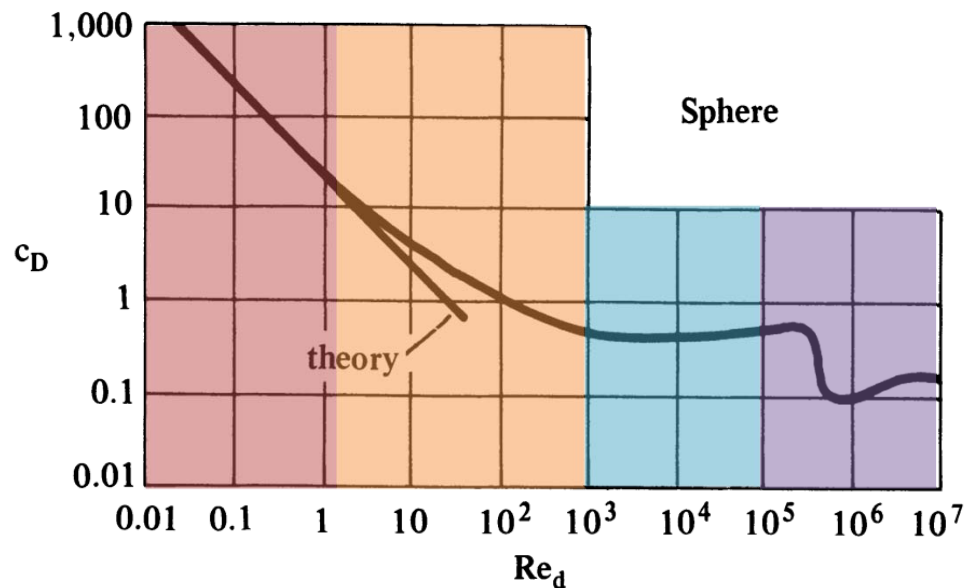
Correlations and plots

$$Re = \frac{D_P V_t \rho_f}{\mu_f}$$

Bubble and Droplet Mechanism



1. Droplet settling theory $C_d = f(Re)$



$$C_d = \frac{24}{Re} \quad Re < 2 \quad \text{Stokes drag coefficient}$$

$$C_d = \frac{24}{Re} (1 + 0.15Re^{0.687}) \quad 2 < Re < 10^3$$

$$C_d = 0.44 \quad 10^3 < Re < 10^5$$

$$C_d = 0.1 \quad 10^5 < Re < 10^6$$

(P.P Wegener, 1997)
(Richardson et al., 2002)

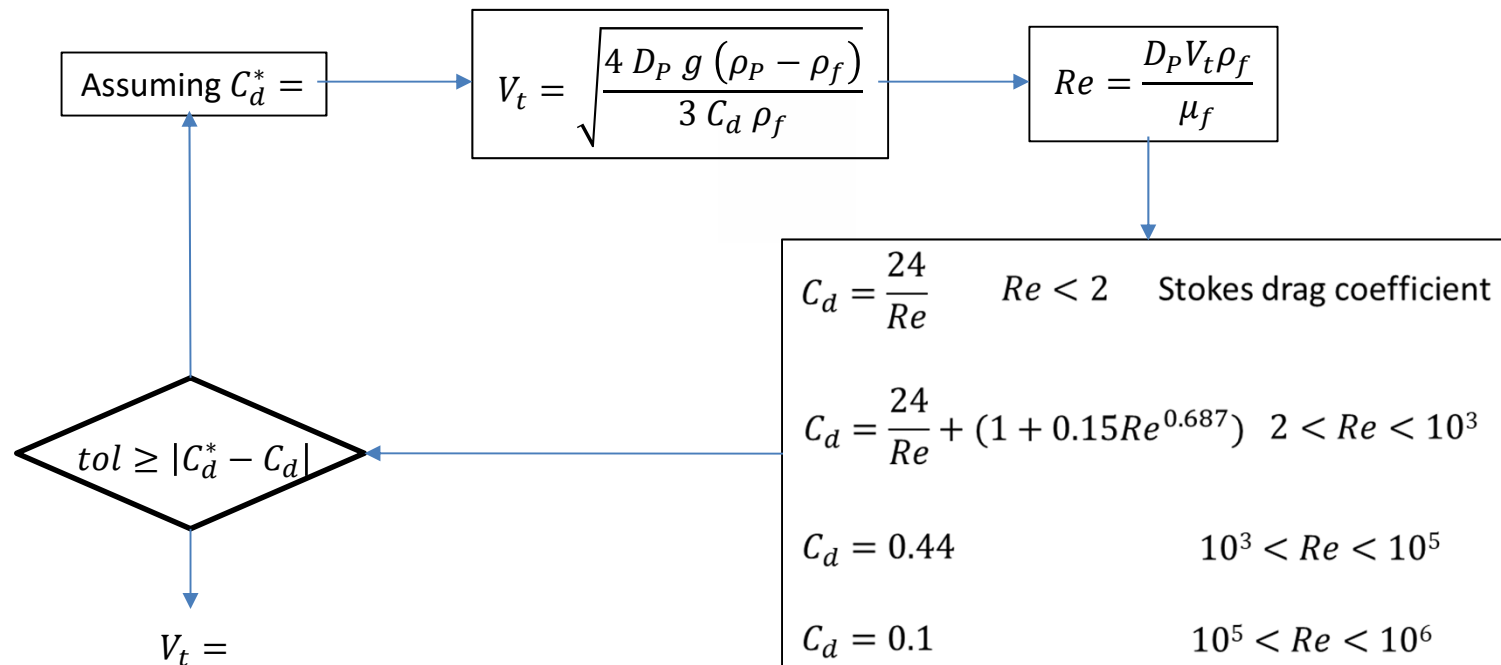
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Bubble and Droplet Mechanism



1. Droplet settling theory

Trial and error solution

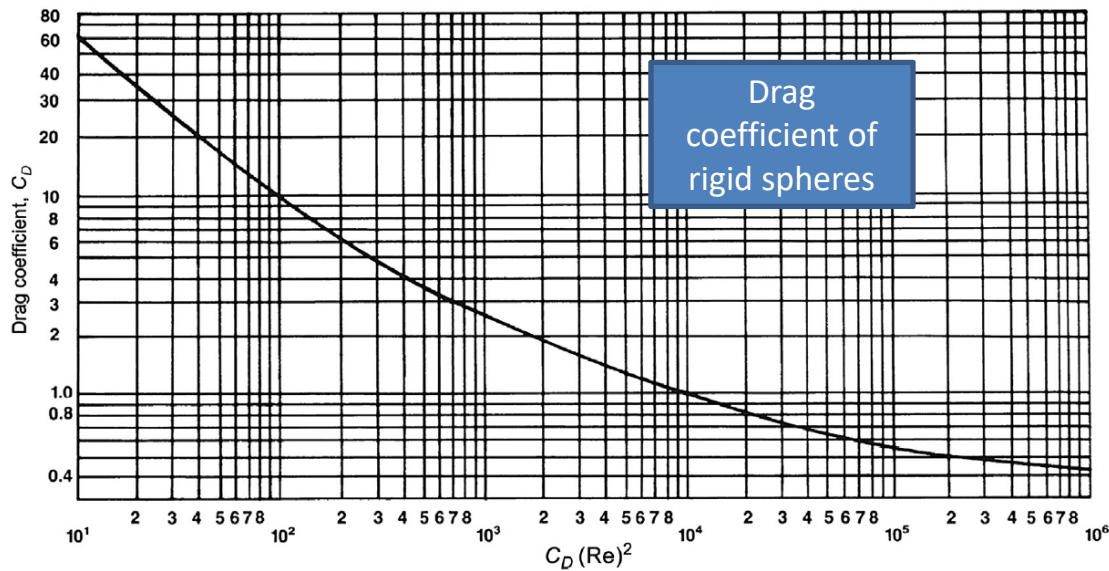


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Bubble and Droplet Mechanism



1. Droplet settling theory *Correlations and plots*



$$C_D Re^2 = \frac{1.31 * 10^7 \rho_f \cancel{D_p^3} (\rho_p - \rho_f)}{\mu^2}$$

$$V_t = \sqrt{\frac{4 D_P g (\rho_P - \rho_f)}{3 C_d \rho_f}}$$

$$Re = \frac{D_P V_t \rho_f}{\mu_f}$$

$$\rho \dots kg/m^3$$

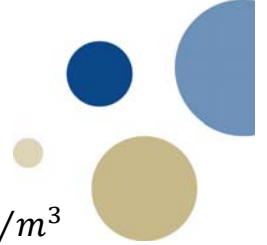
$$\mu \dots mPa \cdot s$$

$$D \dots m$$

(Bahadori, A. 2014)

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Bubble and Droplet Mechanism



$\rho \dots kg/m^3$

$\mu \dots mPa \cdot s$

$D \dots m$

1.2 Droplet settling theory Gravity Settling Laws

$$C_d = \frac{24}{Re}$$

- **Stokes Law** $Re < 2$
 $K_{CR}(Re = 2) = 0.033$

$$V_t = \frac{1000gD_p^2(\rho_p - \rho_f)}{18\mu}$$

$$C_d = \frac{18.5}{Re^{0.6}}$$

- **Intermediate Law** $2 < Re < 500$
 $K_{CR}(Re = 500) = 0.435$

$$V_t = \frac{2.94g^{0.71}D_p^{1.14}(\rho_p - \rho_f)^{0.71}}{\rho_f^{0.29}\mu^{0.43}}$$

$$C_d = 0.44$$

- **Newton's Law** $500 < Re < 2 * 10^5$
 $K_{CR}(Re = 2e5) = 23.64$

$$V_t = 1.74 \sqrt{\frac{D_p g (\rho_p - \rho_f)}{\rho_f}}$$

$$D_p = K_{CR} \left[\frac{\mu^2}{g\rho_f(\rho_p - \rho_f)} \right]^{0.33}$$

$K_{CR} \dots$ proportional constant

(Bahadori, A. 2014)

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Bubble and Droplet Mechanism



2. Souders-Brown approach

$$K_{SB} = \sqrt{\frac{4 D_P g}{3 C_d}} = V_t \sqrt{\frac{\rho_f}{(\rho_P - \rho_f)}} \longrightarrow V_t = K_{SB} \sqrt{\frac{(\rho_P - \rho_f)}{\rho_f}}$$

API 12J recommended range of K_{SB} for vertical and horizontal separators assuming the vessel is equipped with a wire-mesh mist extractor

Type	Height or length (m)	Typical K_{SB} range (m/s)
Vertical	1.52	0.037-0.073
	≥ 3.05	0.055-0.107
Horizontal	3.05	0.122 to 0.152
	Others lengths	$(0.122 \text{ to } 0.152) * (L/3.05)^{0.56}$

NORSOK P-100: $K_{SB} = 0.137 * \left(\frac{L}{6}\right)^{0.58}$

(John M. Campbell, A. 2015)

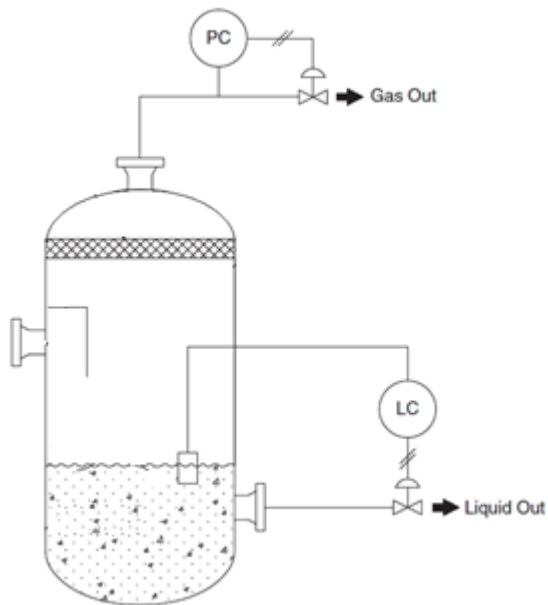
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Separator sizing

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Separator sizing



1. Vapor Capacity → Cross-sectional area
2. Liquid Capacity → Residence time to “de-gas”
3. Operability → Ability to deal with solids, liquid slugs, turn-down, etc.



L_{sem}/D_v

Vertical separator sizing

1. Vapor Capacity

$$Q_g = V_t A_v \quad (\text{Actual Conditions})$$

$$Q_g^{s.c.} = Q_g \left(\frac{P}{P_{s.c.}} \right) \left(\frac{T_{s.c.}}{T} \right) \frac{1}{z} \quad (\text{s.c.} = \text{standard conditions})$$

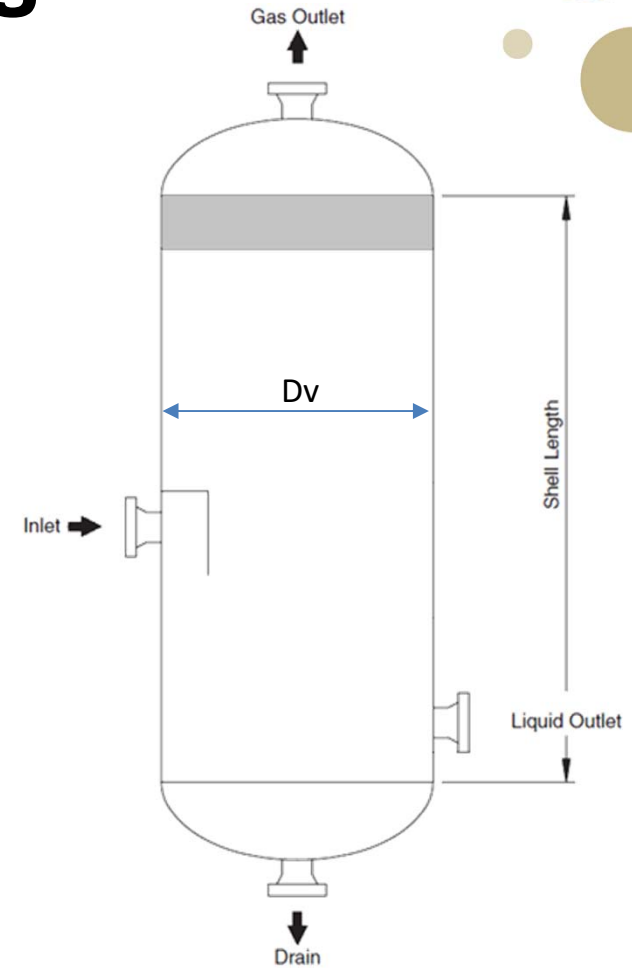
15 C
1 atm (1.01325 bara)

Calculate V_t

Calculate A_v

Calculate D_v

**Minimum value*



(Stewart and Arnold, 2008)

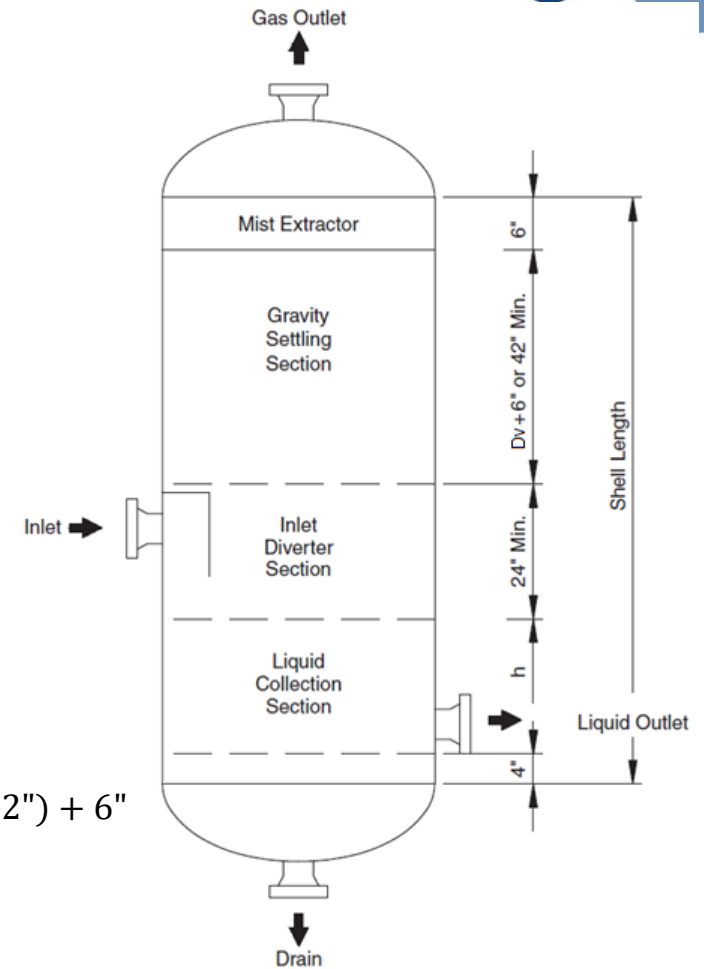
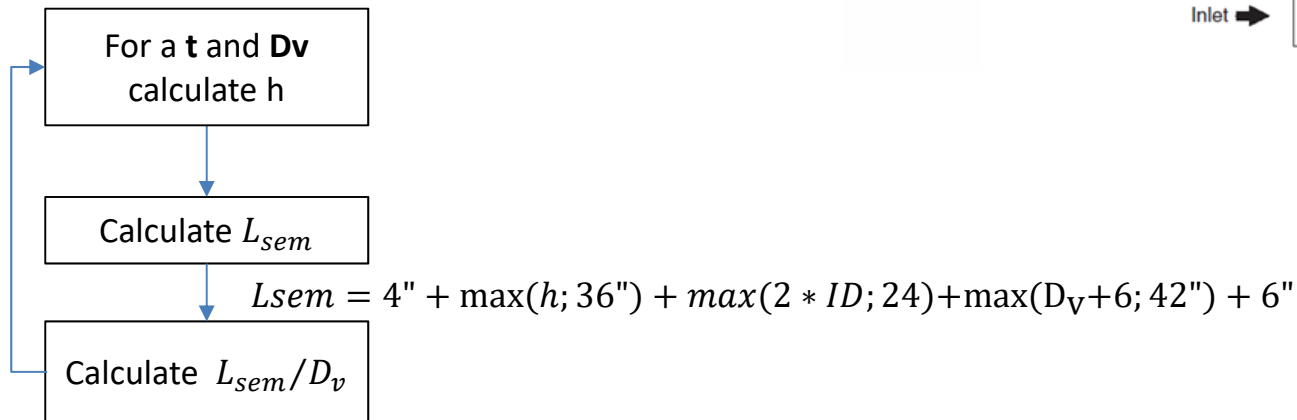
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Vertical separator sizing

1. Vapor Capacity

2. Liquid Capacity

$$t Q_l = h A_v \rightarrow h = \frac{t Q_l 4}{\pi D_v^2} \quad t \dots \text{retention time}$$



(Stewart and Arnold, 2008)

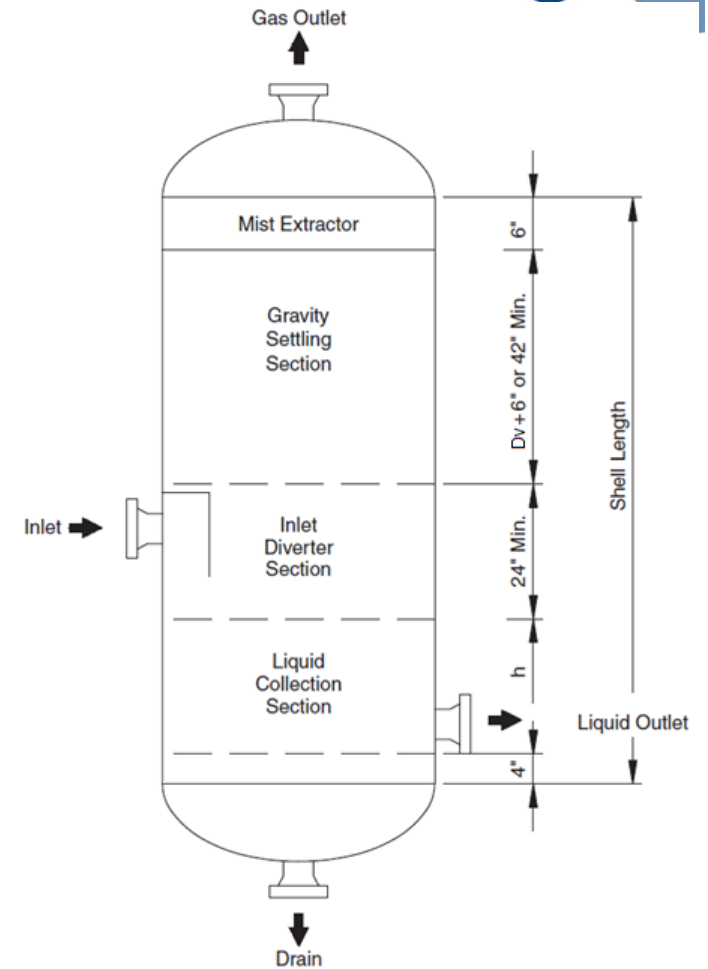
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Vertical separator sizing

1. Vapor Capacity
2. Liquid Capacity
3. Operability

Adjust L_{sem}

Adjust Retention time



(Stewart and Arnold, 2008)

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Design criteria



For the gravity settling section → Droplet size $\approx 140 - 150\mu m$

Vent scrubbers → Droplet size $\approx 300 - 500\mu m$

Retention time in most application → 30 s and 3 min →

Horizontal Liquid volume →

50% full liquid

L/D relation



Vertical Separator are normally $L/D = 2$ to 4

Horizontal Separator are normally $L/D = 2.5$ to 5

Retention Time for Two-Phase Separators

°API Gravity	Retention Time (Minutes)
35+	0.5 to 1
30	2
25	3
20+	4+

1. If foam exists, increase above retention times by a factor of 2 to 4.
2. If high CO_2 exists, use a minimum of 5-minute retention time.

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Design criteria

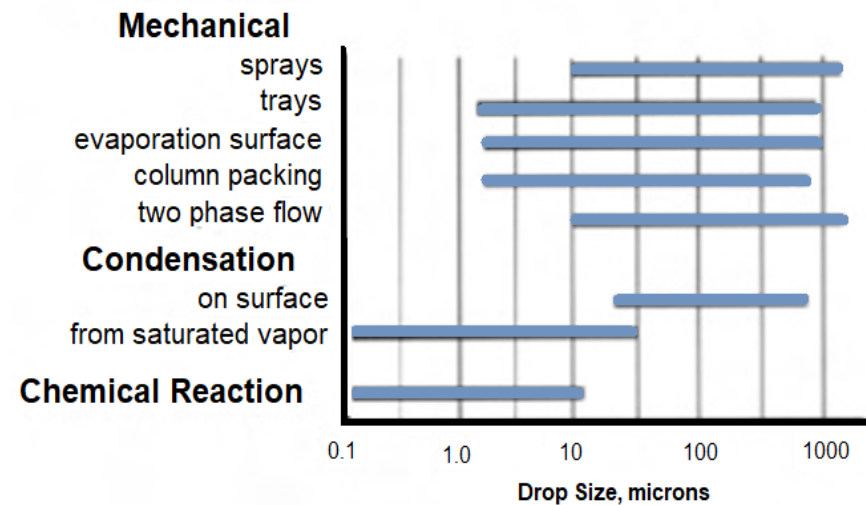


Other factors

Typical Retention Times for Gas/Liquid Separator

Application	Retention Time, minutes
Natural Gas – Condensate separation	2 – 4
Fractionator Feed Tank	10 – 15
Reflux Accumulator	5 – 10
Fractionation Column Sump	2
Amine Flash Tank	5 – 10
Refrigeration Surge Tank	5
Refrigeration Economizer	3
Heat Medium Oil Surge Tank	5 – 10

Typical particle size distribution ranges from entrainment caused by various mechanisms



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