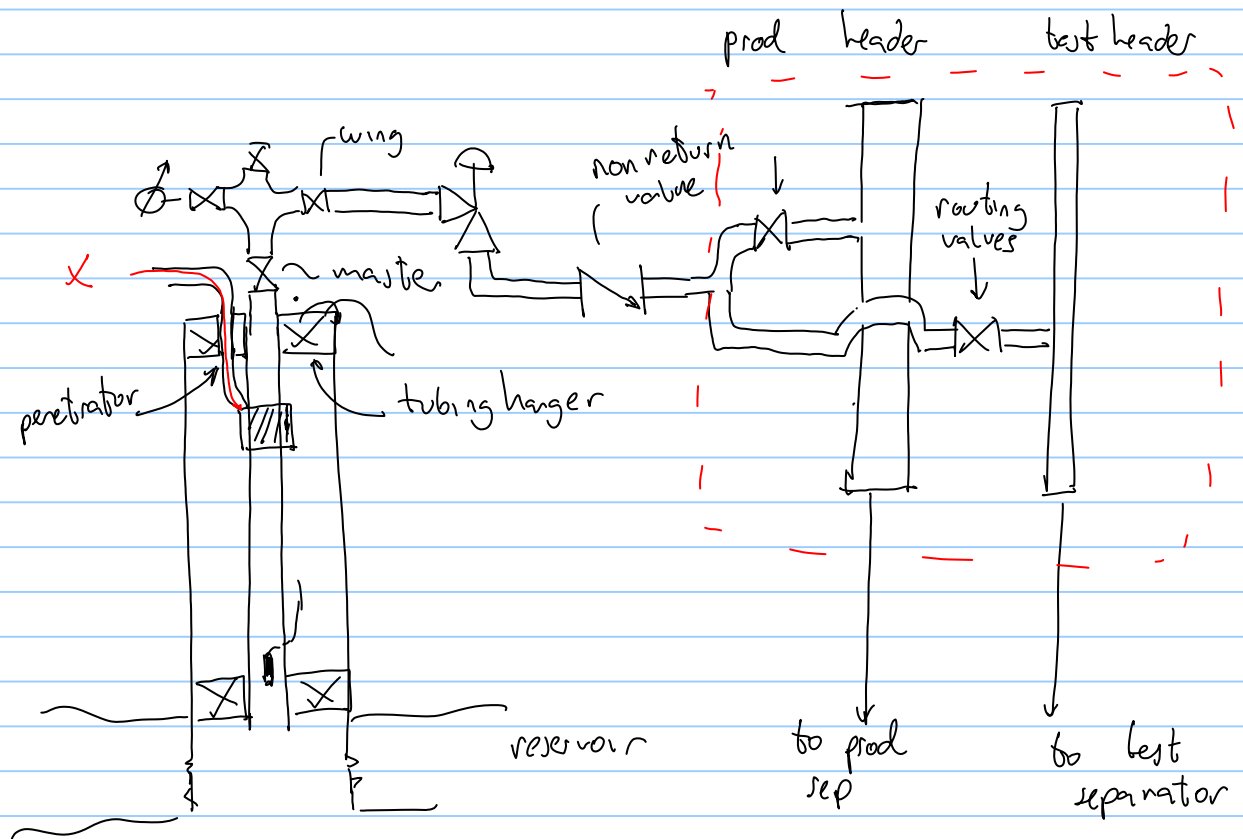


Day 2 :



## Christmas Tree Systems



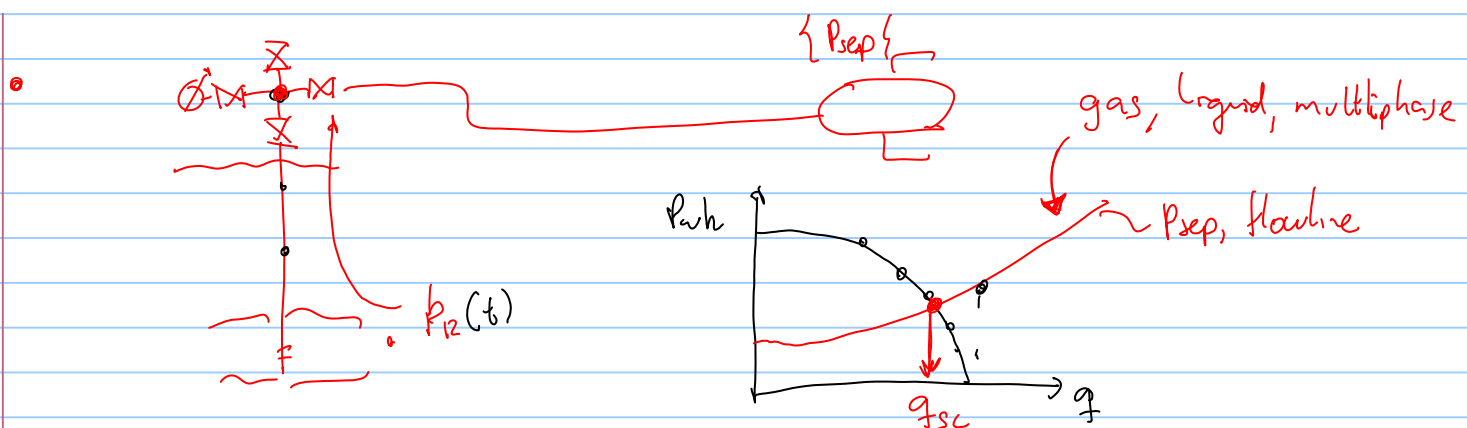
Onshore tree

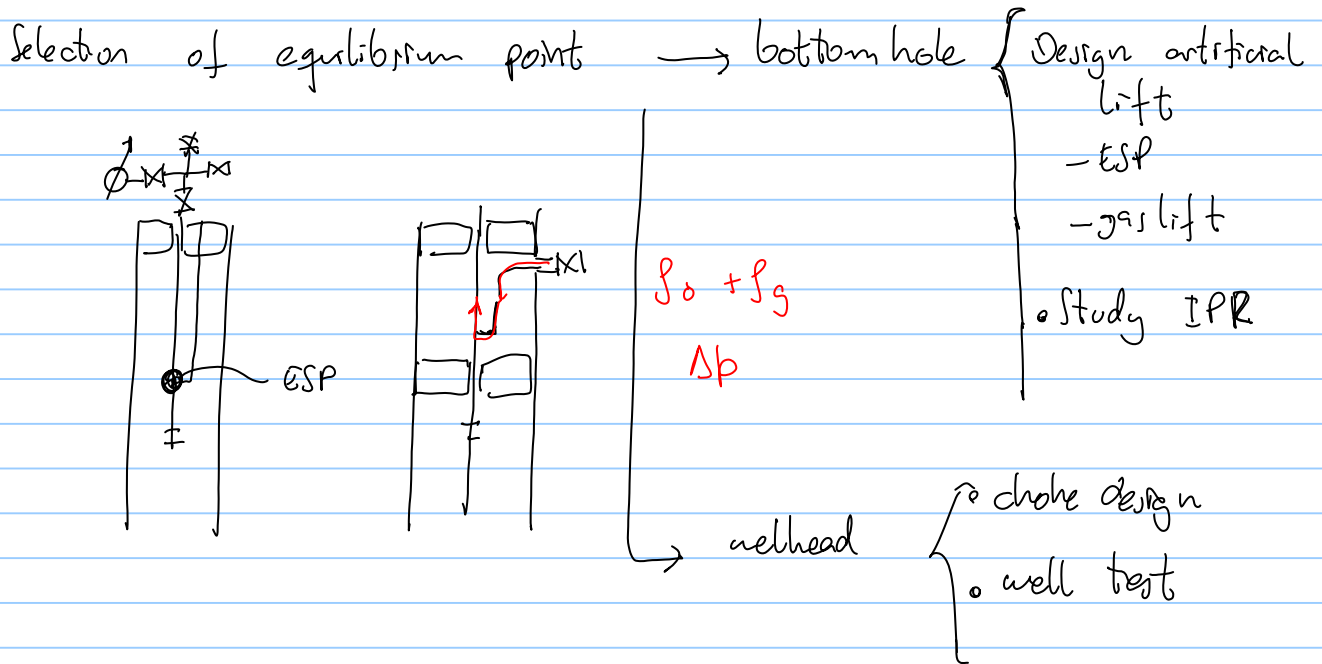


Offshore tree

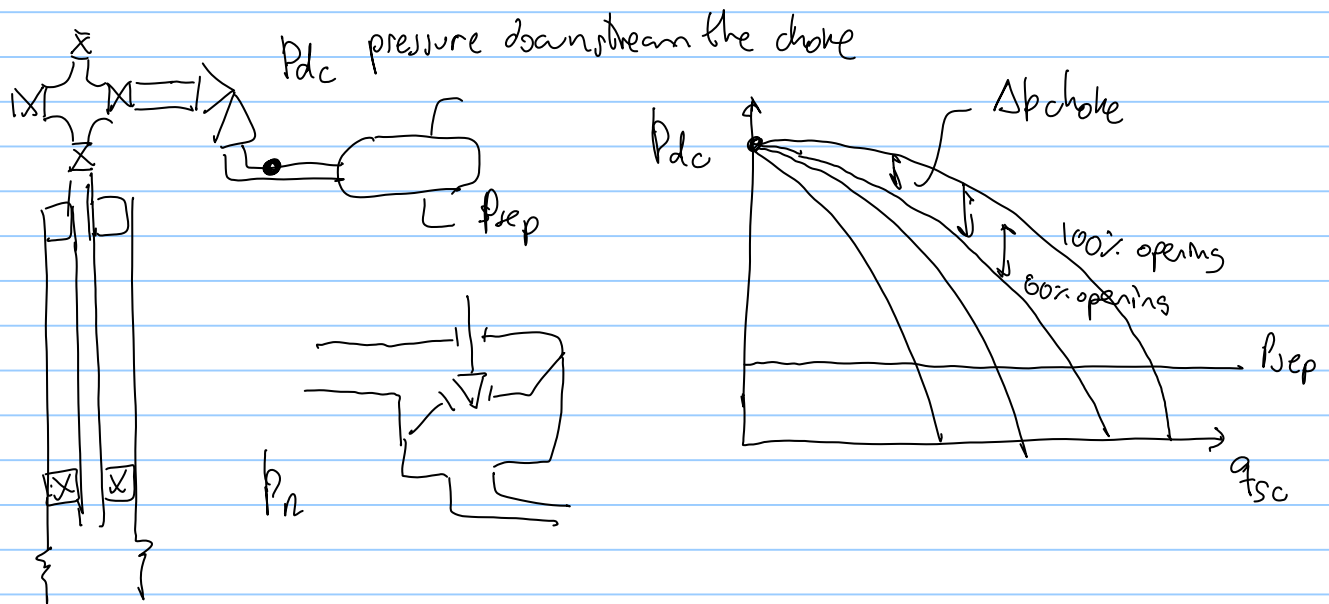


Subsea tree

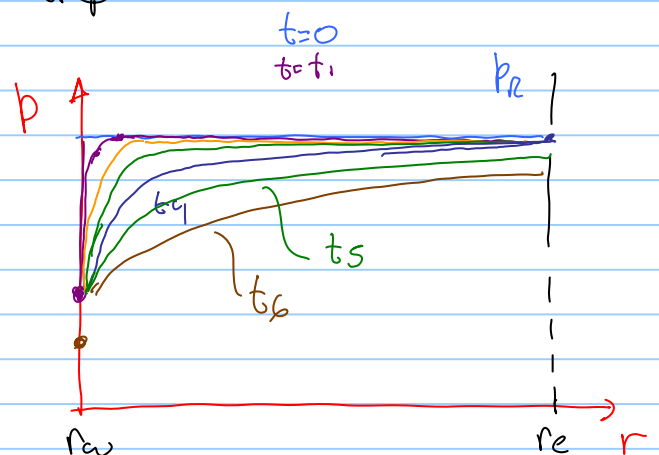
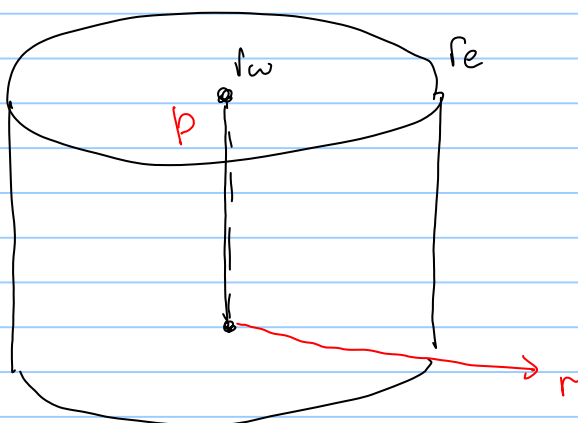




Flow equilibrium including the choke



• IPR Inflow performance relationship



$t_0 \rightarrow t_4$  transient, infinite acting

$t_4 \rightarrow$  forward  $\rightarrow$  pseudo steady state (PSS)

$t_0 \rightarrow t_1$   $t_{ps}$  time to pseudo steady state

$C_f, C_R, M, K$  } tight formation  $\downarrow K \rightarrow t_{ps}$  is in order of months or years

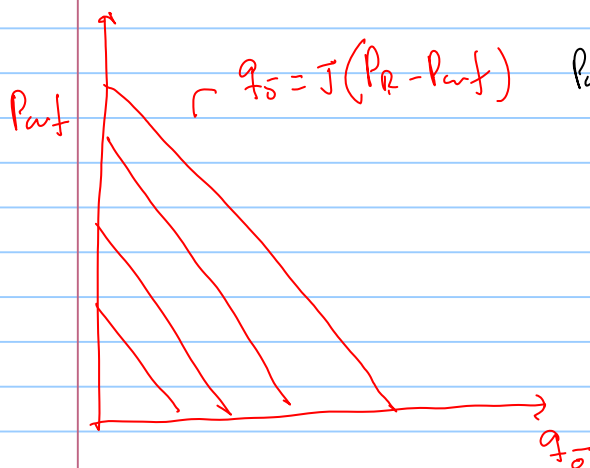
normal formation  $\uparrow K \rightarrow t_{ps}$  hours or days

if  $t_{ps}$  is short { then I use IPR for PSS } \* for this course

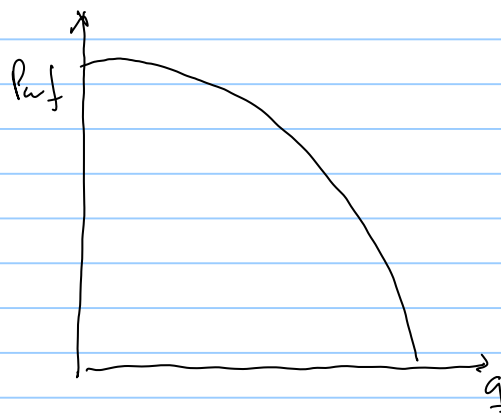
if  $t_{ps}$  is long { then use IPR for transient regime  
 $IPR = f(t)$

pss IPRs

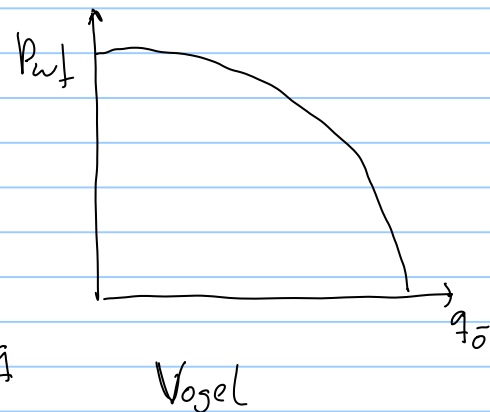
single phase oil



GAS



Saturated oil (mixture oil+gas)



$$q_o = C(P_R^2 - P_{wf}^2)^n$$

$$0.5 \leq n \leq 1$$

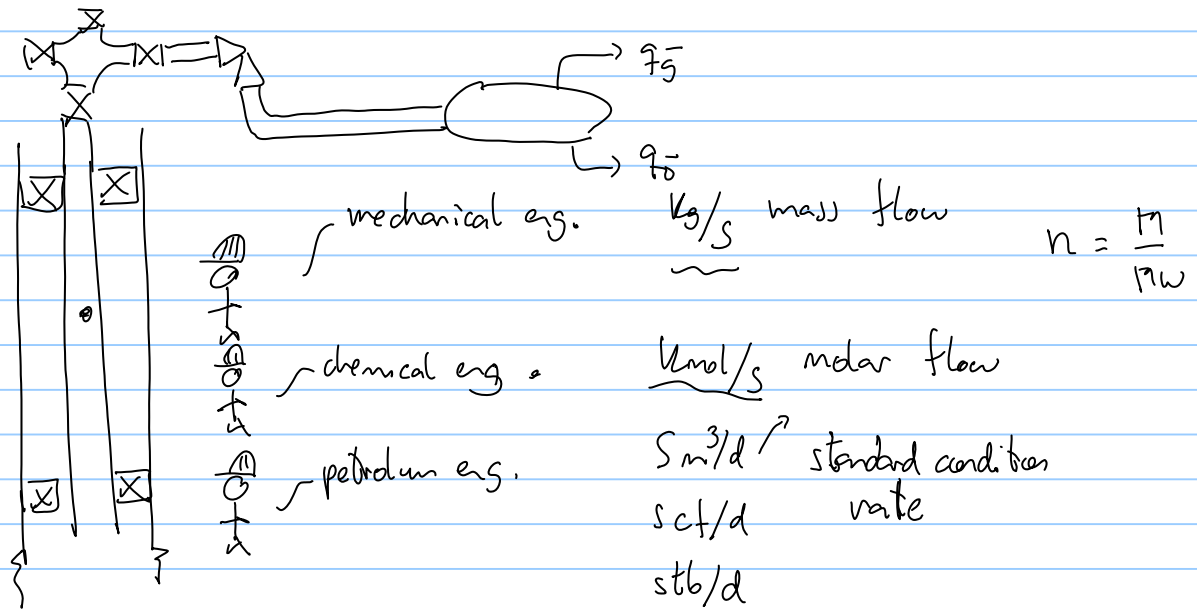
turbulent for low pressure  $p \leq 100$  bara

$$m(p) = 2 \int_0^p \frac{p}{M^2} dp$$

for medium  $\rightarrow$  high pressure

$$q_o = C \left[ m(P_R) - m(P_{wf}) \right]^n \quad \left\{ \begin{array}{l} \text{for the whole} \\ \text{pressure range} \end{array} \right.$$

$$\frac{q_o}{q_{o\max}} = 1 - 0.2 \frac{P_{wf}}{P_R} + 0.8 \frac{P_{wf}^2}{P_R^2}$$



$\Delta p \text{ friction} = V$



• Case: dry gas

Ideal gas law

$pV = nRT$

$0.7 \rightarrow 2$

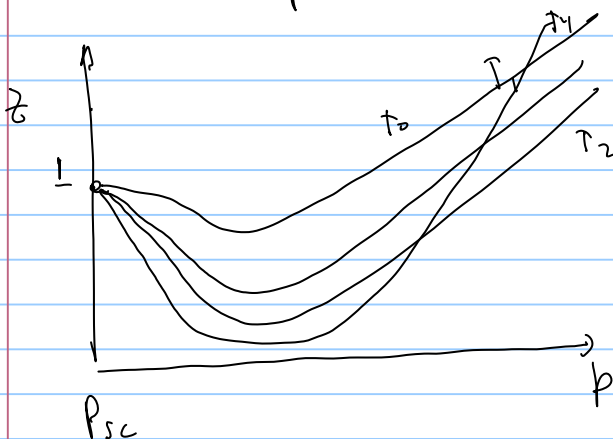
Boyle

Hooke  $\frac{F}{x}$

Charles

insert a correcting factor in the ideal gas eq.

$pV = Z nRT$  ✓



Gay Lussac

Avogadro





Van der Waals corresponding states principles

pseudo-reduced pressure  $p_r = \frac{p}{p_c} \sim p_{si}$

pseudo reduced  $T$   $T_r = \frac{T}{T_c}$  use absolute units only!  $^{\circ}K, ^{\circ}R$

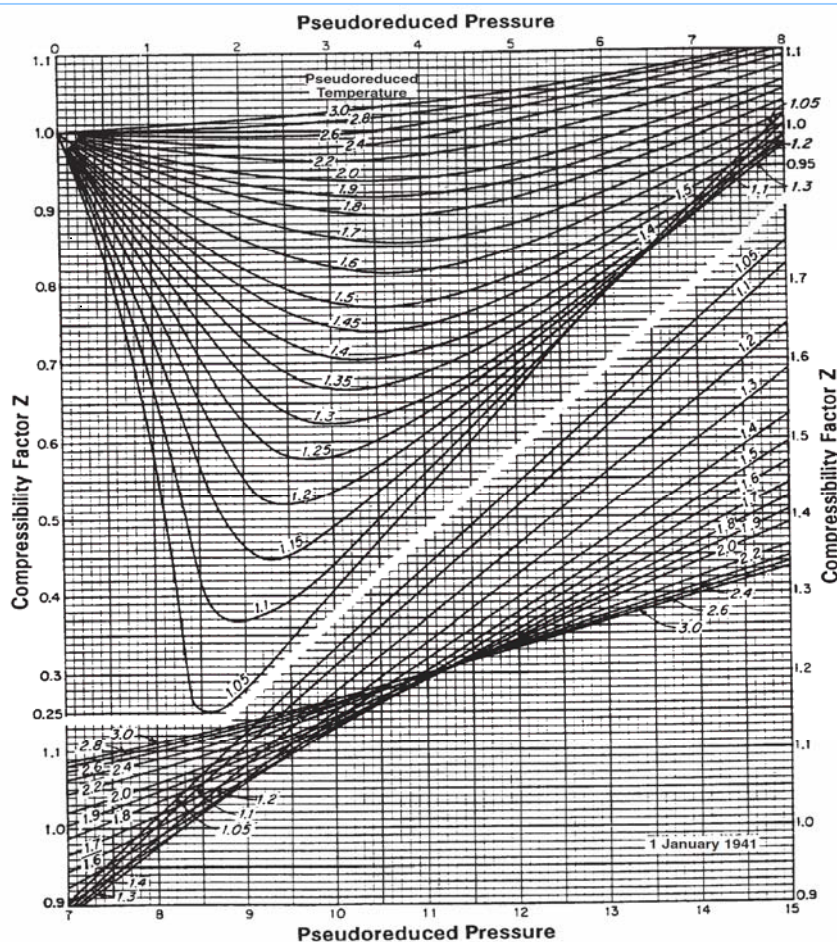
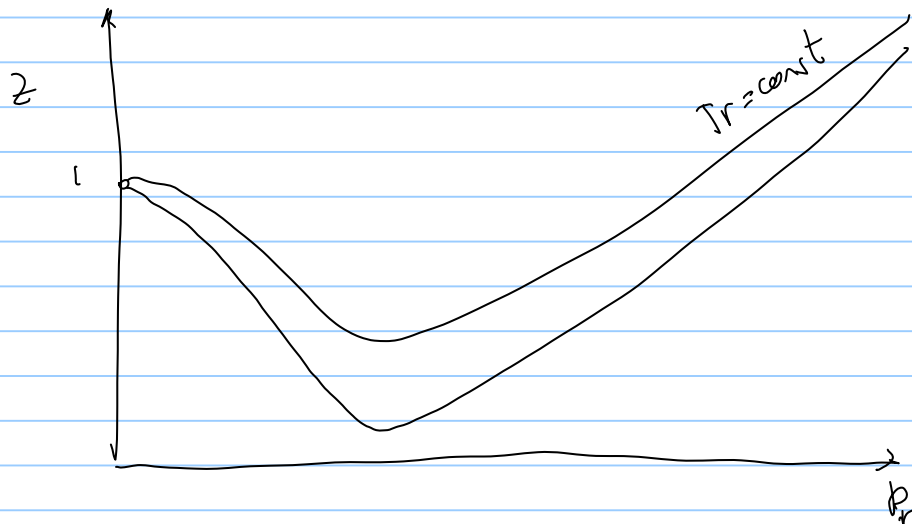


Fig. 3.6—Standing-Katz<sup>4</sup> Z-factor chart.

$$p_r = 12$$

$$T_r = 1.4$$

Production gas is usually a mixture of several gases (components)

$C_1$	$N_2$	$y_i$ ~ mole fraction
$C_2$	$CO_2$	$C_1$ 0.80
$C_3$	$H_2S$	$C_2$ 0.10
$C_4$		$C_3$ 0.03
$C_5$		$\vdots$
$C_7 +$		$\vdots$
		$\hline \Sigma \downarrow$

$$MW_{mix} = \sum_{i=1}^N y_i MW_i$$

components

Specific gravity of gas

$$\gamma_g = \frac{MW_{mix}}{MW_{air}} \rightarrow 28.97$$

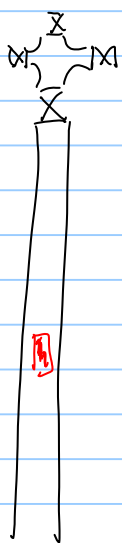
$$P_{cmix} = \sum_{i=1}^N y_i P_{ci}$$

$$T_{cmix} = \sum_{i=1}^N y_i T_{ci}$$

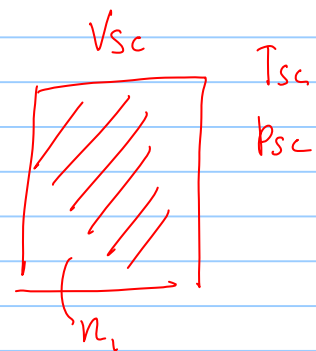
Correlation Sutton

$$P_{cmix} = f(\gamma_g)$$

$$T_{cmix} = f(\gamma_g)$$



$n_i, P_i, T_i$



$$PV = nRTz \quad \text{in S.C.}$$

$$P_{sc} V_{sc} = n R T_{sc} \quad z \approx 1$$

$$R = 8314.3 \frac{J}{K \cdot mol}$$

$$V_{sc} = n \cdot \frac{R T_{sc}}{P_{sc}}$$

$$V_{sc} = n \cdot 23.67 \frac{m^3}{K \cdot mol}$$

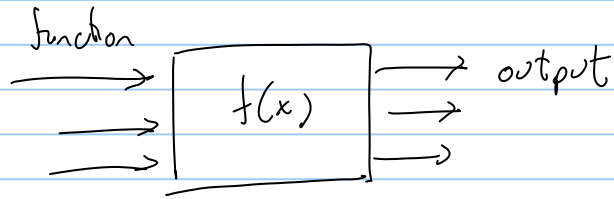
$$\hookrightarrow n \cdot 379.7 \frac{Scf}{K \cdot mol}$$

Excel VBA

access the VBA module

alt + F11

function



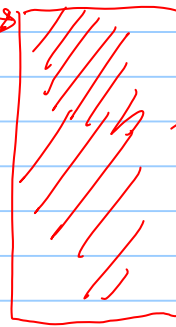
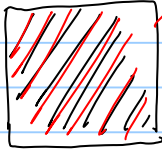
options → trust center → trust center options → macro settings  
 ↳ enable all.

 $V @ P, T$ 

at any point in the production system

 $V_{sc} @ P_{sc}, T_{sc}$ 

no liquid is condensed



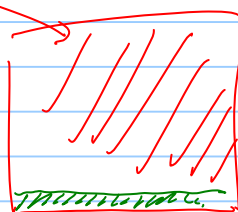
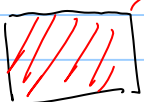
gas volume factor  $B_{g,d} = \frac{V @ P, T}{V_{sc}} = \frac{\frac{n z R T}{P}}{\frac{n \cdot 1 \cdot R T_{sc}}{P_{sc}}} = \frac{z T (P_{sc})}{P T_{sc}}$  ✓

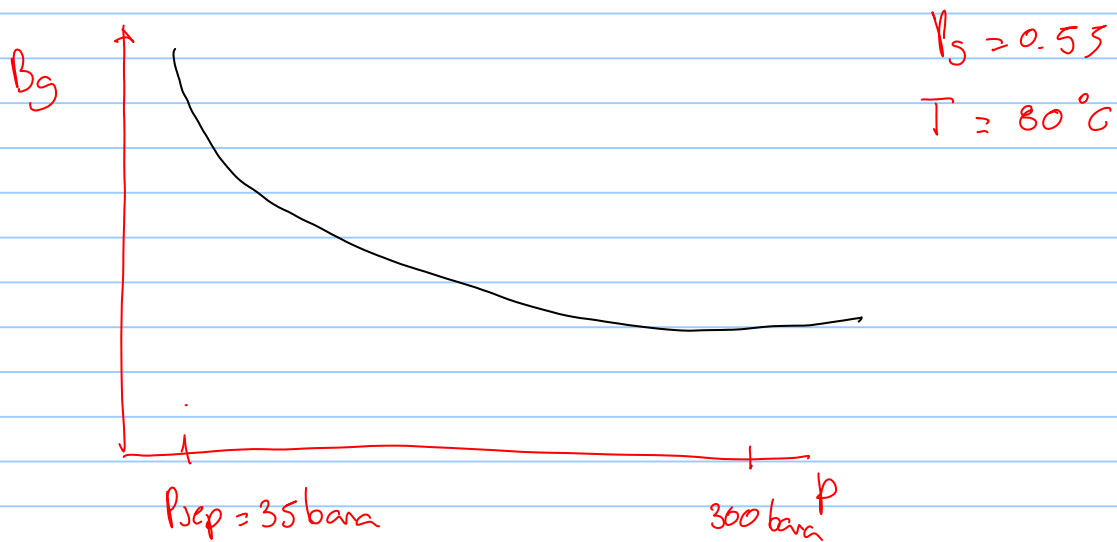
if  $T = T_r$   
 $P = P_r$

gas formation volume factor

if I give you  $q_g @ sc$ find  $q_g @ P, T$ 

$$q_g @ P, T = B_{g @ P, T} \cdot q_{sc}$$

 $V, P, T$  $V_{sc}, P_{sc}, T_{sc}$  $B_{g,wet}$



\* homework 2 :

plot  $B_g$  for

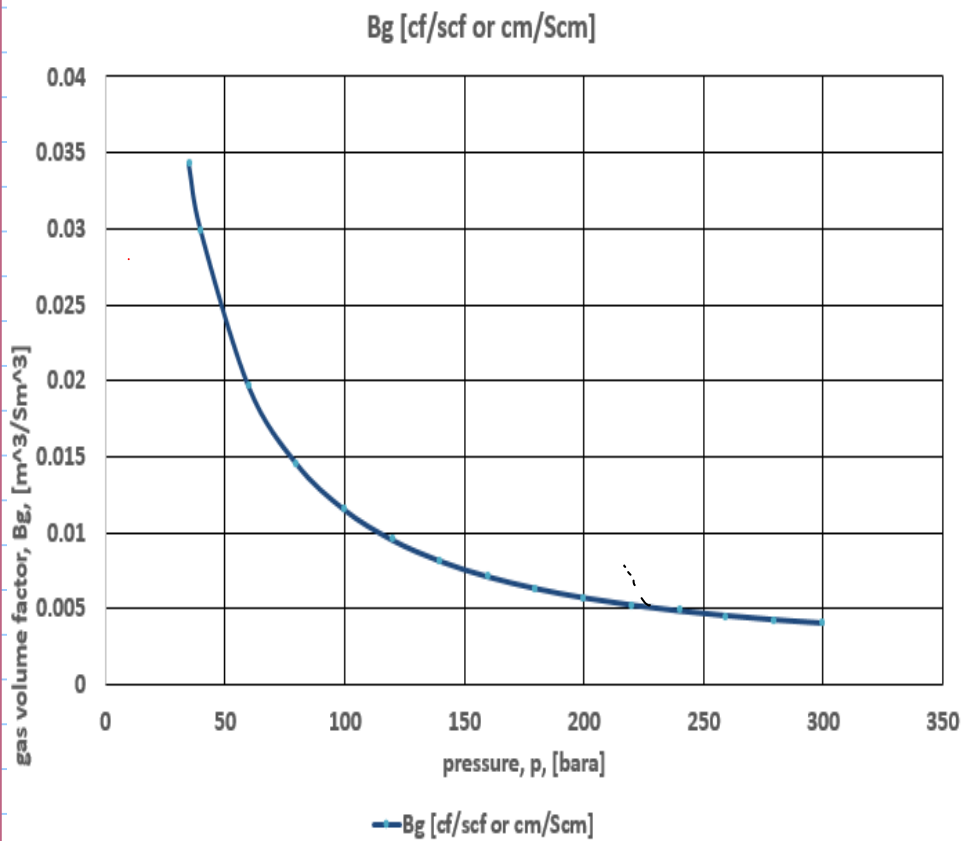
$$T = 120^\circ\text{C}$$

$$T = 60^\circ\text{C}$$

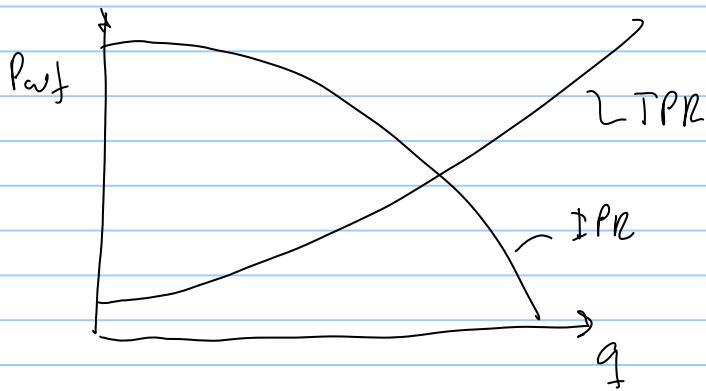
$$T = 80^\circ\text{C}$$

$$p \in [35 \text{ bara} - 300 \text{ bara}]$$

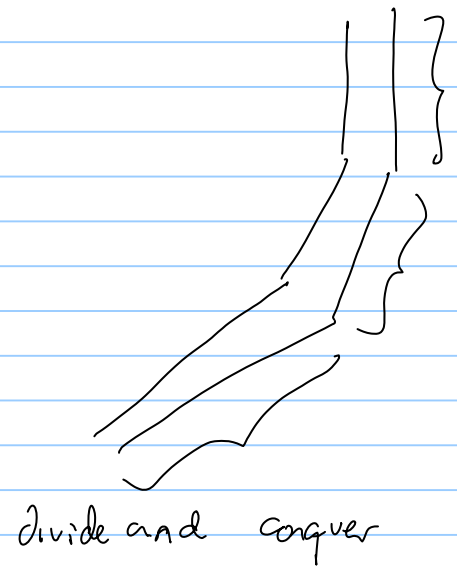
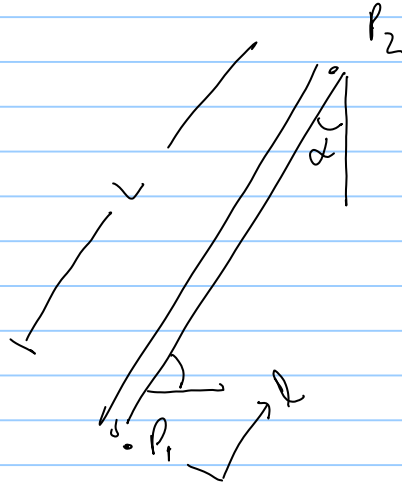
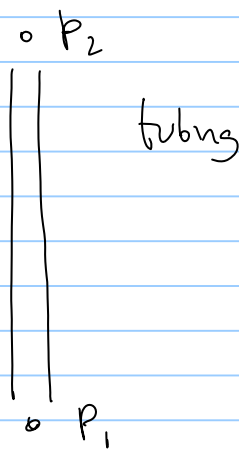
$$\gamma_s = 0.55$$







Development of Dry gas tubing equation  $\leadsto$  flowlines also applicable for



$$\frac{dp}{dl} = \underbrace{-\rho \cdot g \cos \alpha}_{\text{hydrostatic}} - f \rho \frac{u^2}{2\phi}$$

$\downarrow \downarrow \downarrow$        $\downarrow$   
 $\uparrow$        $\downarrow$   
 $\text{hydrostatic}$        $\text{ID}$

$p$  changes along tubing

$$\rho = f(p, T)$$

$$\rho = \frac{p M_w}{z R T}$$

$$\frac{dp}{dl} = - \left( \frac{p M_w}{z R T} \right) g \cos \alpha - f \frac{p M_w}{z R T} \frac{u^2}{2\phi}$$

$$u = \frac{\dot{m}}{\rho \cdot A} \quad \frac{\text{kg/s}}{\frac{\text{kg/m}^3 \cdot \text{m}^2}} = \frac{\text{m}}{\text{s}}$$

$$A = \pi \cdot \phi^2 \cdot 0.25$$

$$\frac{dp}{dl} = - \frac{p M_w}{z R T} g \cos \alpha - f \frac{p M_w}{z R T} \frac{1}{2\phi} \frac{\dot{m}^2 (z R T)^2}{p^2 M_w^2 \pi^2 \phi^4}$$

$$u = \frac{\dot{m} \cdot z R T}{p M_w \cdot \pi \phi^2} \cdot \frac{4}{\pi \phi^2}$$

$$\frac{dp}{dl} = - \frac{p M_w}{z R T} g \cos \alpha - f \frac{\dot{m}^2 z R T^3}{\phi^5 \cdot \pi^2 p M_w}$$

Constant  $B = \frac{\rho_w \cdot g \cos \alpha}{\bar{z} R T}$

$$D = \frac{f \bar{z} R T \dot{m}^2}{\phi^5 \pi^2 \rho_w}$$

$$\frac{dp}{dl} = -B \cdot p - \frac{D}{p} \quad - \quad \frac{dp}{Bp + \frac{D}{p}} = dl$$

B and D are constant and equal to

$$B = \frac{\rho_w g \cos \alpha}{\bar{z} R T}$$

$$\bar{z} = f(\bar{p}, \bar{T}) \quad \bar{T} = \frac{T_1 + T_2}{2}$$

$$D = \frac{\bar{f} \bar{z} R \bar{T} \dot{m}^2}{\phi^5 \pi^2 \rho_w}$$

$$\int_1^2 \frac{dp}{Bp + \frac{D}{p}} = \int_1^2 -dl$$

= -L

$$\int_1^2 \frac{p dp}{Bp^2 + D}$$

change variable

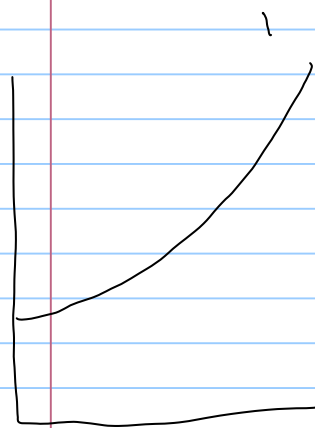
$$U = Bp^2 + D$$

$$dU = 2p B dp$$

$$\frac{1}{2B} \int_1^2 \frac{dU}{U} = \frac{1}{2B} \ln \left( \frac{U_2}{U_1} \right)$$

return the variable change

$$\frac{1}{2B} \ln \left( \frac{Bp_2^2 + D}{Bp_1^2 + D} \right) = -L$$

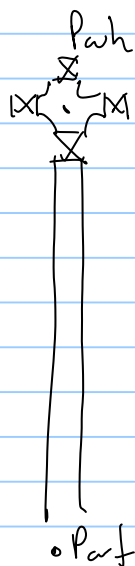


$$\frac{B p_2^2 + D}{B p_1^2 + D} = e^{-L 2 \cdot B}$$

elevation factor  
 $S = 2 L B$

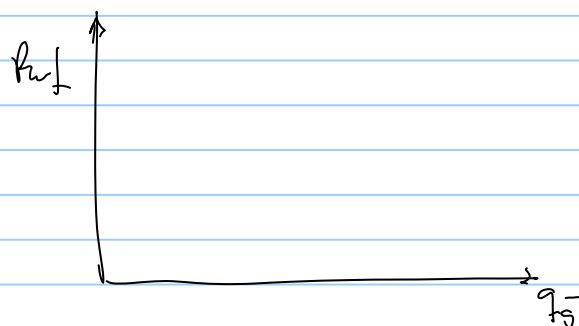
$$\frac{B p_2^2 + D}{B p_1^2 + D} = e^{-S}$$

$$p_1^2 = p_2^2 e^S + \left(\frac{B}{D}\right) e^S - 1$$



$$p_2 = P_{wh}$$

$$p_1 = P_{wf}$$



$$\frac{B}{D} = \frac{\dot{m}^2 \bar{z} R \bar{T} \theta \cdot f \bar{z} R \bar{T}}{M_w \pi^2 \phi^5 M_w g \cos \alpha}$$

$$\frac{B}{D} = \frac{q_{sc}^2 (\bar{z} R \bar{T})^2 \theta \cdot f}{\pi^2 \phi^5 \cos \alpha R^2 g} \left(\frac{p_{sc}}{T_{sc}}\right)^2$$

$$\dot{m} = \rho_{sc} \cdot q_{sc}$$

$$\downarrow$$

$$\frac{kg}{m^3} \cdot \frac{m^3}{d}$$

$$\rho_{sc} = \frac{p_{sc} \cdot M_w}{Z R T_{sc}}$$

$$p_1^2 = p_2^2 e^S + q_{sc}^2 \left( \text{constant} \right)$$

$$p_1^2 = p_2^2 e^S + \frac{q_{sc}^2}{C_r^2}$$

# Tubing flow Equation-Dry gas

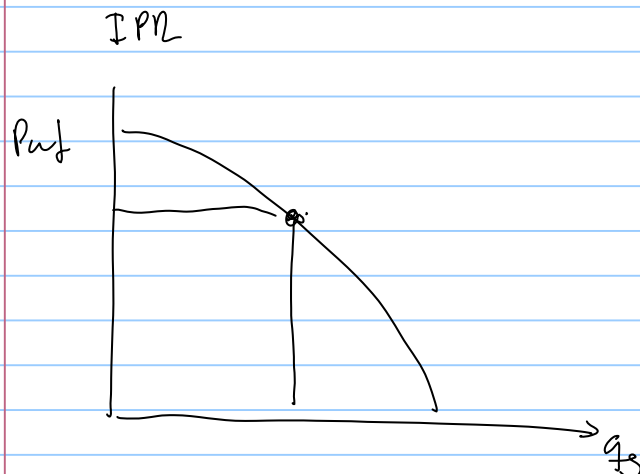
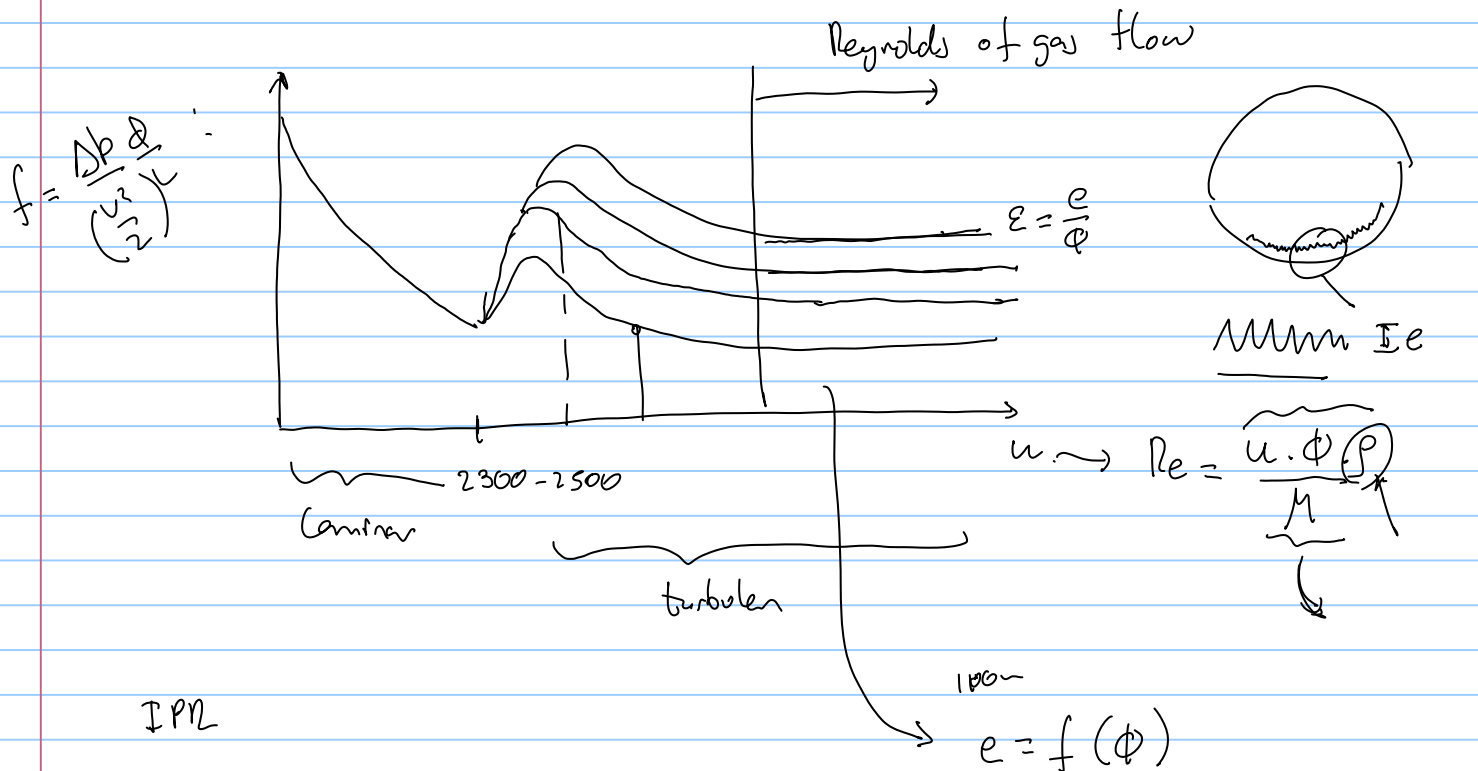
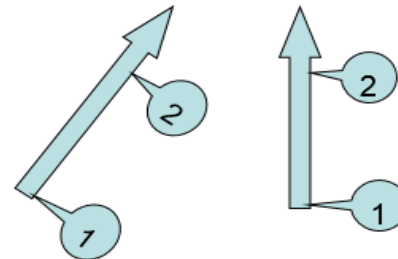
$$q_{sc} = \left( \frac{\pi}{4} \right) \left( \frac{R}{M_{air}} \right)^{0.5} \left( \frac{T_{sc}}{P_{sc}} \right) \left[ \frac{D^5}{\gamma_g f_M Z_{av} T_{av} L} \right]^{0.5} \left( \frac{s e^s}{e^s - 1} \right)^{0.5} \left( \frac{p_1^2}{e^s} - p_2^2 \right)^{0.5}$$

$$\frac{s}{2} = \frac{M_g g}{Z_{av} R T_{av}} H = \frac{(28.97) \gamma_g g}{Z_{av} R T_{av}} H$$

$$q_{gsc} = C_T \left( \frac{p_1^2}{e^s} - p_2^2 \right)^{0.5}$$

$$p_{inlet} = p_1 = e^{s/2} \left( p_2^2 + \frac{q_g^2}{C_T^2} \right)^{0.5}$$

$$p_{wh} = p_2 = \left( \frac{p_1^2}{e^s} - \frac{q_g^2}{C_T^2} \right)^{0.5}$$



Check tubing calculator spreadsheet. Gas exercise.

**CALCULATION OF TUBING FLOW CONSTANT**

height difference	[m]	3000
Internal Diameter	[m]	0.15
Gas gravity	[-]	0.55
Line length	[m]	3000
Inlet temperature	[K]	378
Outlet temperature	[K]	360
Inlet pressure	[bara]	303
Outlet pressure	[bara]	275
Ave. Temperature	[K]	369
Ave. Pressure	[bara]	289
Ave. Compressibility factor	[-]	0.981
Friction factor	[-]	0.012
Elevation Coeff, S		0.156
Tubing flow constant	[Sm <sup>3</sup> /bar]	3.58E+04