

Δp

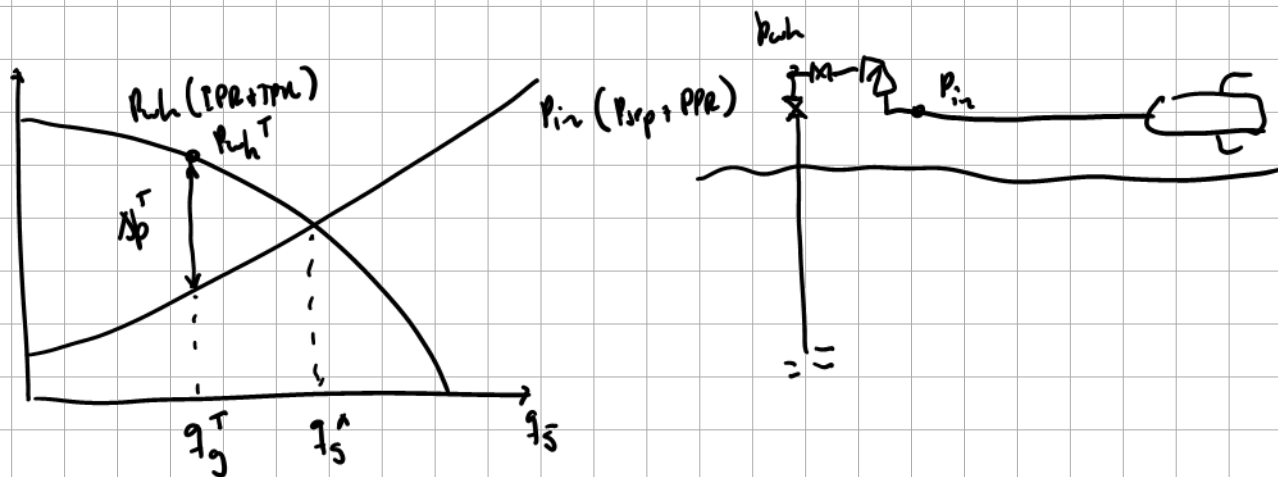
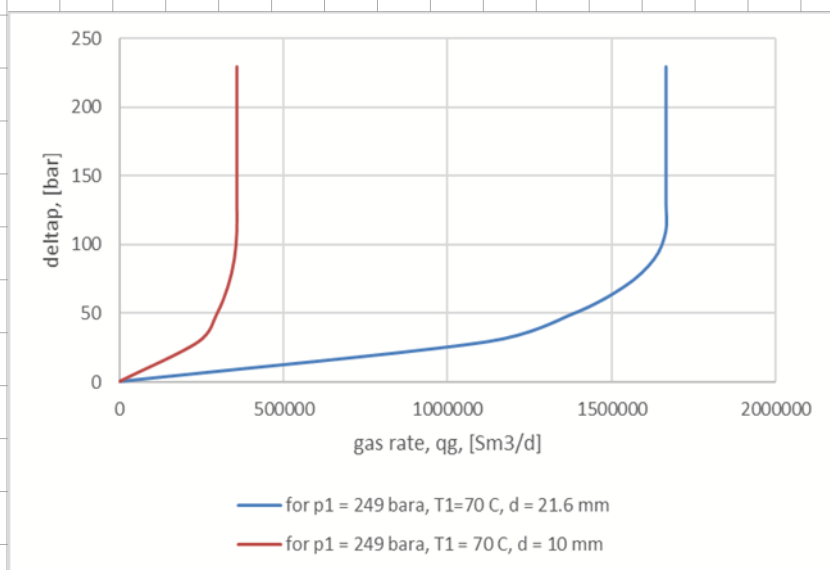
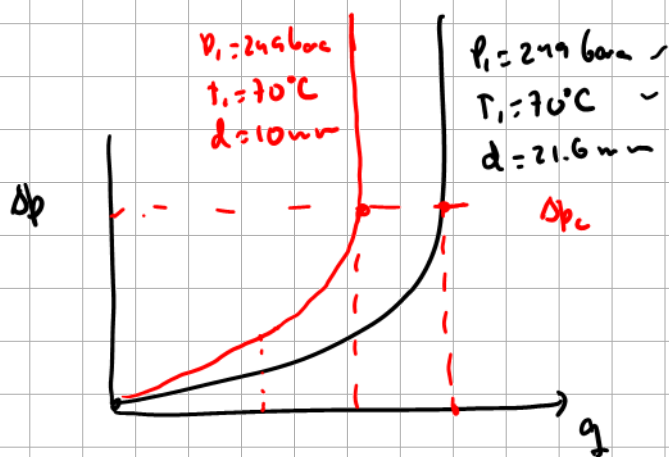
$$y = \frac{p_2}{p_1}$$

$$y_c = 0.546$$

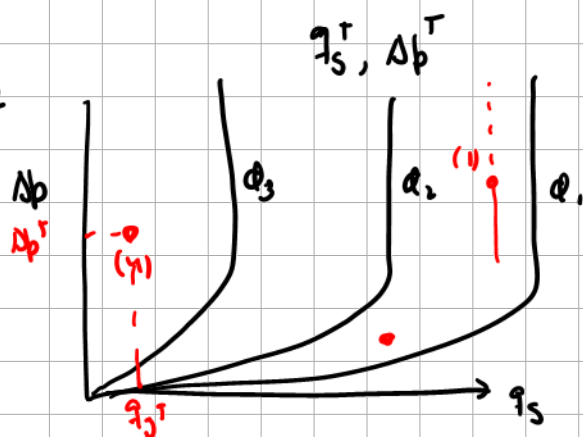
$$\Delta p_c?$$

$$\Delta p = p_1 - p_2 = p_1 (1 - y)$$

$$\Delta p_c = p_1 (1 - y_c)$$



Close performance

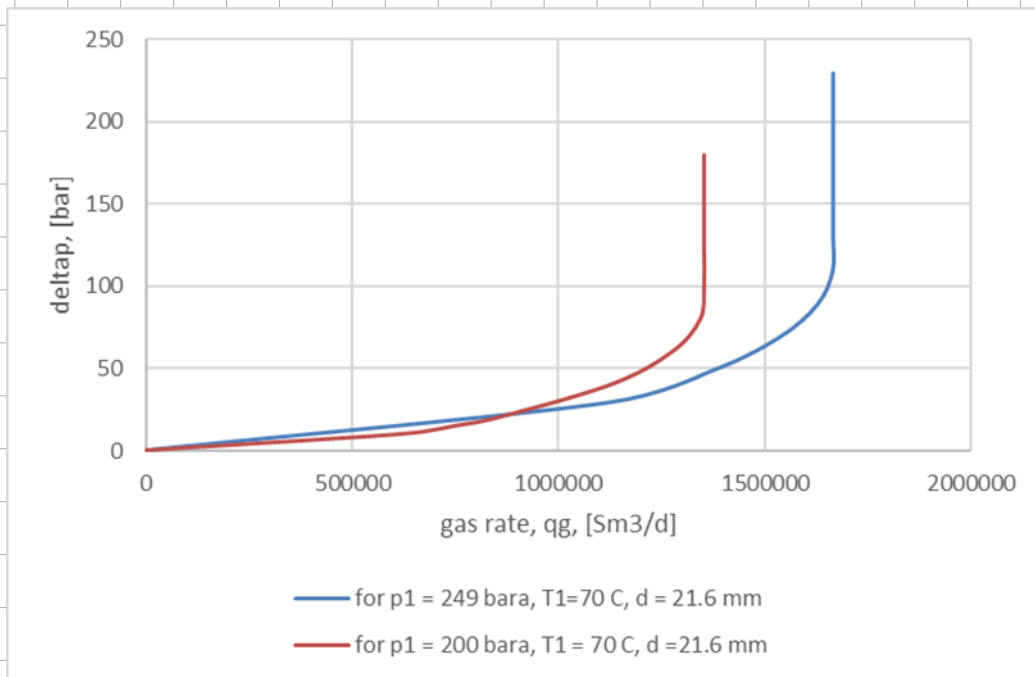
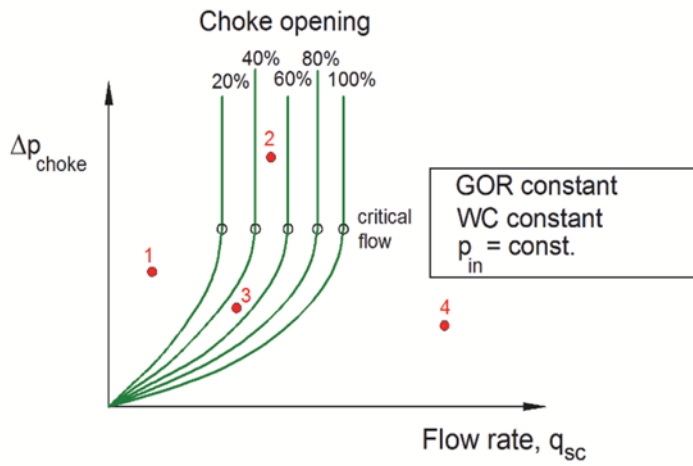


(1) critical
(2) subcritical
for $p_1 = p_h^T$

$$q_1 > q_2 > q_3$$

(3) too small alone

(4) too large alone



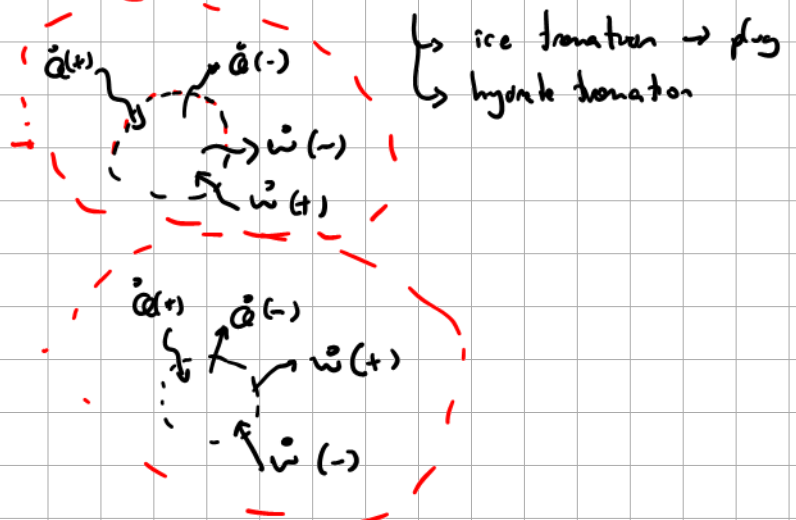
temperature change in choke

low GOR, high WC $\rightarrow$ expansion leads to an increase in temperature (modest)
 high GOR $\rightarrow$ " " " a significant decrease in temperature

$$\dot{Q} + \dot{W} + \dot{E}_{in} - \dot{E}_{out} = 0$$

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$$\dot{Q} - \dot{W} + \dot{E}_{in} - \dot{E}_{out} = 0$$



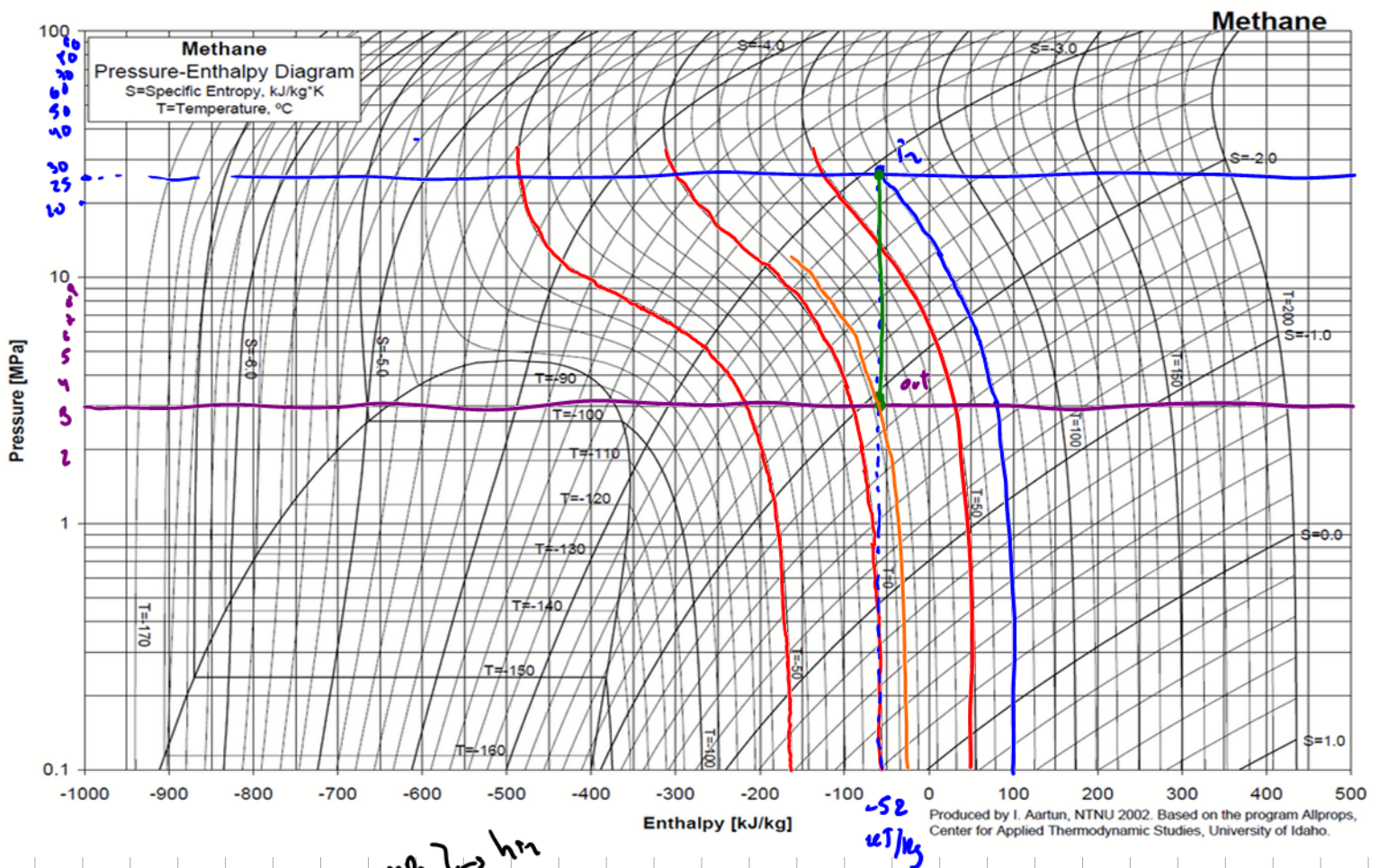
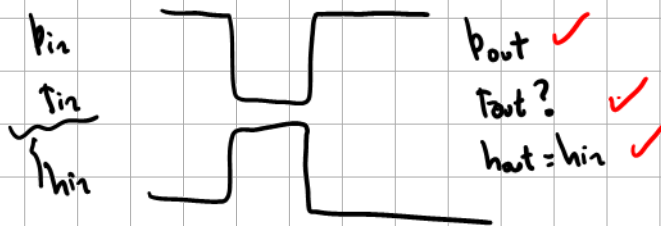
In the above $\dot{Q} = 0$
 $\dot{W} = 0$

$$E_{in} = E_{out}$$

$$\dot{m} \left(h_{in} + \cancel{\frac{V_{in}^2}{2}} + \dots \right) = \dot{m} \left(h_{out} + \cancel{\frac{V_{out}^2}{2}} + \dots \right)$$

$V_{in} \approx V_{out}$

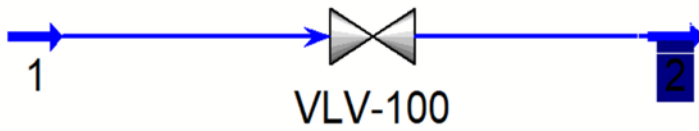
$h_{in} = h_{out}$ iso-enthalpic expansion



$p_{in} = 250 \text{ bara} = 25 \text{ MPa} \rightarrow h_{in}$
 $T_{in} = 70^\circ\text{C}$
 $p_{out} = 30 \text{ bara} = 3 \text{ MPa}$
 $T_{out} = ?$
 $h_{out} = -52 \text{ kJ/kg}$
 $T_{out} = 12^\circ\text{C}$

$1 \text{ MPa} \rightarrow 10 \text{ bara}$
 $1 \text{ GPa} \rightarrow 1000 \text{ Pa}$

$T_{in} = 70^\circ\text{C} \rightarrow T_{out} = 12^\circ\text{C}$



HYSYS Process simulation

