

Q & A's

VOLUME FACTORS (B or $b \equiv 1/B$)
(RATIO)

FORMATION VOLUME FACTOR, FVF

Generic Definition: VF

$$B_p \equiv \frac{V_p(p_i, T)}{V_p(p_{sc}, T_{sc})} \quad ; p \in \{g, o, w\}$$

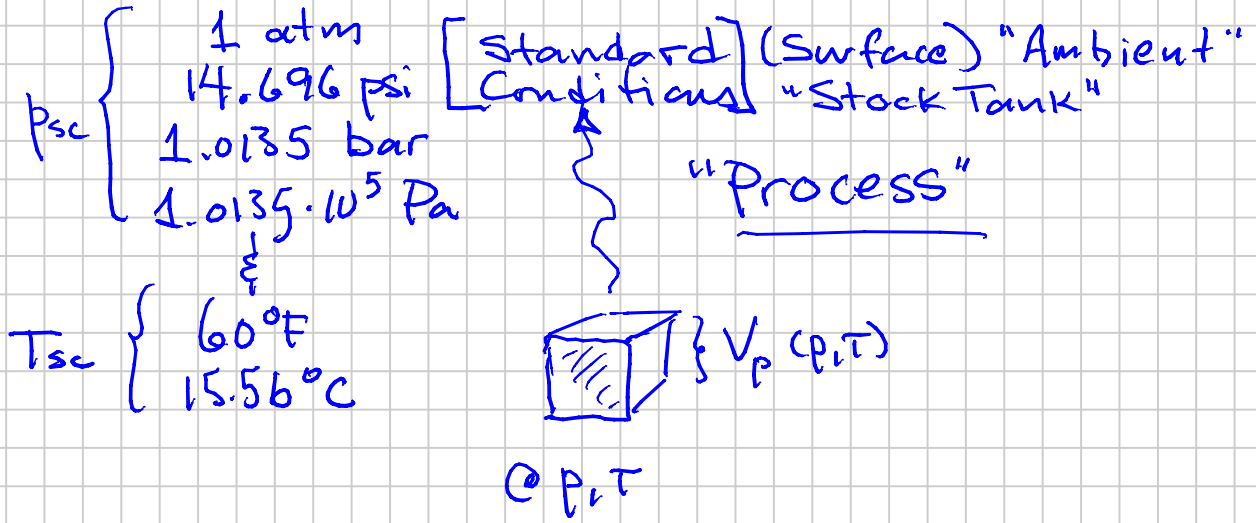
@ S.C.

"Reservoir Rock"
(pores)

B_g

B_o

B_w



b "Shrinkage Factor" (Sometimes used)
Inverse Volume Factor

$$B_{oi} = 1.5 \frac{\text{m}^3}{\text{std m}^3} \quad \frac{\text{RB}}{\text{STB}} \quad \text{Statfjord}$$

reservoir
stock-tank

$$b_{oi} = 0.667 \text{ STB/RB} \quad \text{Sm}^3/\text{m}^3$$

oil "shrinkage" of $V_{oi} \rightarrow V_o$

% reduction in volume from (p_i, T) to (p_{sc}, T_{sc})

$$\text{Shrinkage: } 1 - b = 33.3\%$$

Shrinkage of oil @ (p_i, T) to lesser oil (STO) @ STC (p_{sc}, T_{sc}) is due to the "loss of" mass at STC becoming surface gas.

$$\Rightarrow B_o > 1 \text{ RB/STB}$$

$$B_{oi} \sim 1^+ - 4 \\ (1.2 - 2.5)$$

GAS FVF B_g (b_g)

Assumption: (1) Ideal Gas Law

$$pV = nRT$$

absolute
psia

absolute units
K °R

kPa(a)

bar(a)

Assumption

$$\left. \begin{aligned} (2) \quad V_g(p_i, T) &= n_g \\ V_g(p_{sc}, T_{sc}) &= n_{\bar{g}} \end{aligned} \right\} =$$

$$V_g = \frac{n_g R T_i}{p}$$

$$V_{\bar{g}} = \frac{n_{\bar{g}} R T_{sc}}{p_{sc}}$$

$$F_{gi} = \frac{n_{\bar{g}}}{n_g}$$

$$n_{\bar{g}} = n_g$$

$$\text{Derive } B_g = \frac{V_g(p_i, T)}{V_g(p_{sc}, T_{sc})} = \frac{\gamma_g RT / p}{\gamma_g RT_{sc} / p_{sc}} = \underbrace{\frac{p_{sc}}{T_{sc}} \cdot \frac{T}{p}}$$

Example Calculations :

$$p_{Ri} = 1500 \text{ psia}$$

(Initial Reservoir Pressure)

$$T_R = 160 \text{ } ^\circ\text{F}$$

(Reservoir Temperature)

$$p_{sc} = 1 \text{ atm} = 14.696 \text{ psia}$$

$$T_{sc} = 60 \text{ } ^\circ\text{F}$$

$$14.50377 \frac{\text{psi}}{\text{bar}}$$

Determine B_{gi} & b_{gi}

units B_g

b_g

$$\text{m}^3 / \text{Sm}^3$$

$$\text{Sm}^3 / \text{m}^3$$

$$\text{ft}^3 / \text{scf}$$

$$\text{scf} / \text{ft}^3$$

$$\text{RB} / \text{Mscf}$$

$$\text{Mscf} / \text{RB}$$

$$T_c = (T_{op} - 32) / 1.8$$

$$T_K = T_c + 273.15$$

$$T_R = T_F + 459.67$$

$$35.31 \frac{\text{ft}^3}{\text{m}^3}$$

$$5.6146 \text{ ft}^3 / \text{bbl}$$

$$\text{Mscf} = 10^3 \text{ scf}$$

SI: (bar, °C)

Fields (psia, °F)

$$\frac{p_{sc}}{T_{sc}} = \frac{1.0135 \text{ bara}}{15.56 + 273.15} =$$

$$\frac{p_{sc}}{T_{sc}} = \frac{14.696}{60 + 459.67} = 0.02828$$

$$0.003510$$

$$B_g \ll 1$$

$$B_g = 0.00351 \frac{T(K)}{p(\text{bara})}$$

$$B_g = 0.02828 \frac{T(^{\circ}\text{R})}{p(\text{psia})}$$

$$\left[\frac{\text{m}^3}{\text{Sm}^3} \right] 0.0117$$

$$\left[\frac{\text{ft}^3}{\text{scf}} \right] 0.0117$$

$$b_g = 85.6 \frac{\text{Sm}^3}{\text{m}^3}$$

$$b_g = 85.6 \frac{\text{scf}}{\text{ft}^3}$$

$$b_g \gg 1$$

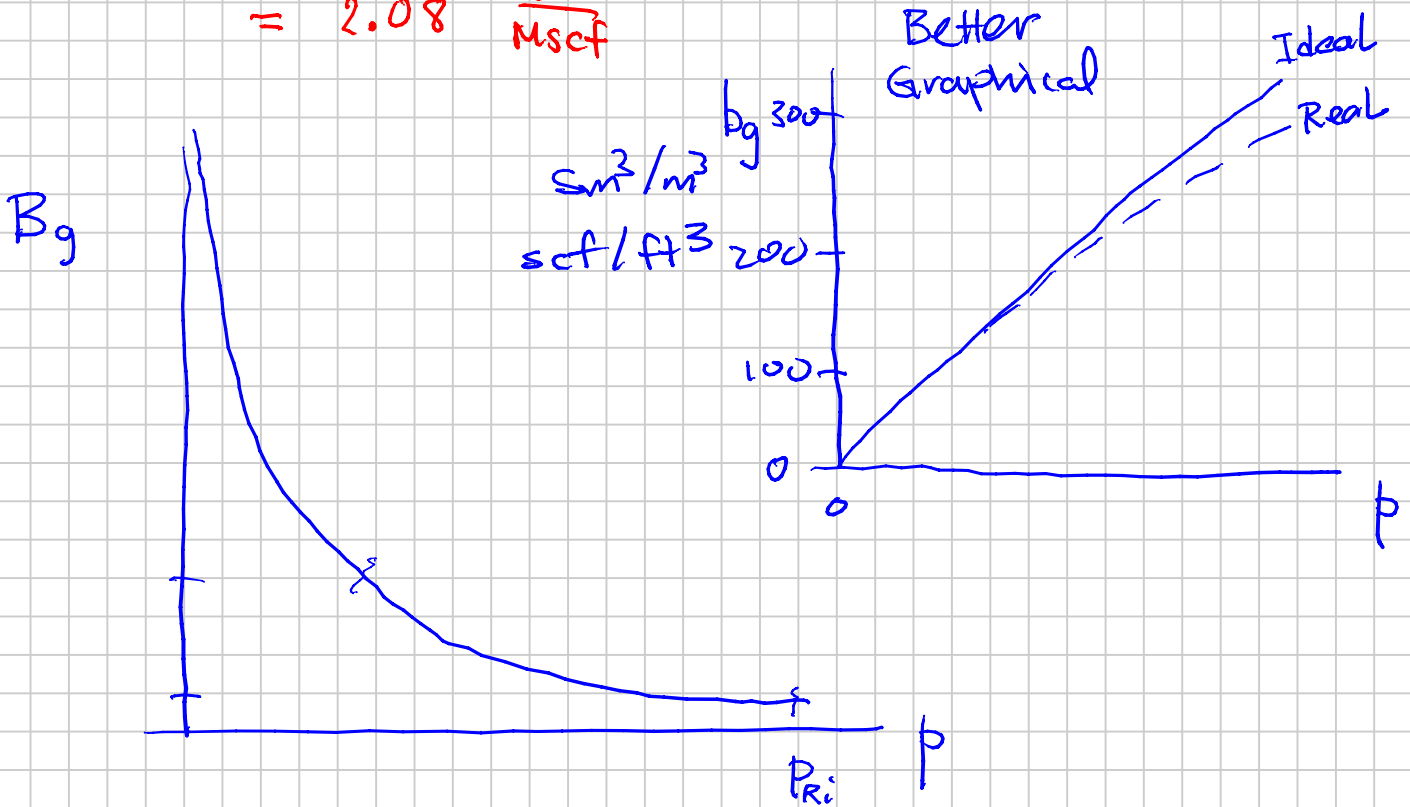
~ 100 - 300 (N.S.)

"bbl"

$$B_g \left[\frac{RB}{Mscf} \right] = 0.0117 \frac{\cancel{ft^3}}{scf} \times \frac{bbl}{5.6146 \cancel{ft^3}} \times \frac{10^3 \cancel{scf}}{Mscf}$$

"1" "1"

$$= 2.08 \frac{RB}{Mscf}$$



$$pV = nRT$$

① Gas volumes reported and measured and used are MOST often @ STC

scf Sm^3 std m^3
surface STC

② Translation from V gas volumes to moles or vice versa is common in engineering.

$$\checkmark 379 \frac{scf}{lb\text{-mole}}$$

$$\checkmark 23.6 \frac{Sm^3}{kg\text{-mole}}$$

$$\sqrt{22.4} \frac{\text{dm}^3}{\text{mol}} \quad \frac{\text{L}}{\text{g-mole}} \quad @ T_{sc} = 0^\circ\text{C}$$

$$R = 10.7315$$

$$\text{psia}, ^\circ\text{R}, \text{ft}^3, \text{lb-mole}$$

$$R = 8314.3$$

$$\text{Pa}, \text{K}, \text{m}^3, \text{kg-mole}$$

$$@ \text{STC} \quad p_{sc} = 14.696 \text{ psia} \quad 1.0135 \text{ bar(a)}$$

$$T_{sc} = 60^\circ\text{F} \quad (\text{SPE})$$

$$(\text{Alt. } \underline{20^\circ\text{C}} \text{ or } \underline{25^\circ\text{C}}) \text{ other Societies}$$

$$0^\circ\text{C}$$

Want: Surface Condition Gas "Molar Volume" v_g

$$v_g \equiv \frac{V_g(p_{sc}, T_{sc})}{n_g} = \frac{RT_{sc}}{p_{sc}}$$

Independent of the gas "makeup"

SI:

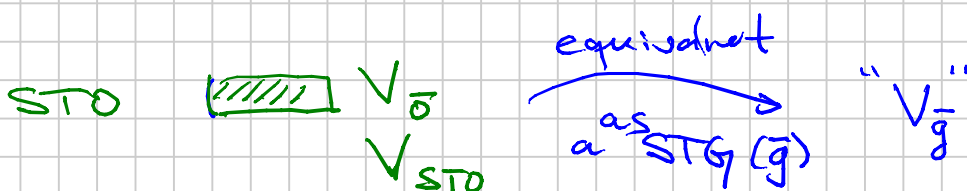
$$v_g = \frac{8314.3 (15.56 + 273.15)}{1.0135 \cdot 10^5} = 23.68 \frac{\text{m}^3}{\text{kg-mole}}$$

CH₄ CO₂ N₂
...

Field:

$$v_g = \frac{10.7315 (60 + 459.67)}{14.696} = 379.5 \frac{\text{scf}}{\text{lb-mole}}$$

Gas Equivalent of STO



$$\downarrow n_{\bar{o}} \times \frac{RT_{sc}}{P_{sc}}$$

" $v_{\bar{g}}$ "

$$[Sm^3] V_{\bar{o}} \times \frac{\rho_{\bar{o}} [kg/m^3]}{M_{\bar{o}} [kg-mole]} \times 23.68$$

$kg-mole$
 $kg-mole$

Barrel Oil Equivalent (boe) bbl

\bar{g} & \bar{o} on an "energy" basis

$\approx (6000 \text{ scf}) \approx (1 \text{ STB})$

	\bar{g}	\bar{o}	boe
North Field (Qatar)	$900 \cdot 10^{12}$ scf	(900 Tcf)	$150 \cdot 10^9$
Troll Gas	$45 \cdot 10^{12}$ scf	(45 Tcf)	$7.5 \cdot 10^9$
Ormen Lange	$10 \cdot 10^{12}$ scf	(10 Tcf)	$1.7 \cdot 10^9$