

TUESDAY REVIEW

* EOS (Cubic)

- SRK | PR(77179)- PUT Calcs (Gas | Oil | Critical & Undefined) = $f(p, T, z_i)$

- Phase Amounts | Densities | Compositions
- Saturation Pressure | Flash | Stability VLE

Ch. 4 - $\mu_{i,v} = \mu_{i,l} \mid \mu_i(\text{EOS}, p, T, z_i) = RT \ln f_i$ (fugacity)

- EOS Definition

EOS MODEL {

- Table 1 - M_i T_{ci} p_{ci} ω_i S_i { C_{ai} C_{bi} } Ω_{ai} Ω_{bi}
- Table 2 - k_{sj} BIPs | $k_{ij} = k_{ji}$, $k_{ii} = 0$, $k_{\text{non-HC/HC}} \sim 0.1$
- EOS_{xx} (Tuned Model | Process Model) $xx = 15-40^+$
- EOS_x (Reservoir $x < 10$, often 6-8)

• Molar Composition (or $w_i \rightarrow m_i \rightarrow M_i \rightarrow z_i$)

- Same component state as EOS model

- Map of z_i components to EOS model components

- EOS (Thermodynamic) Validation

• K_{ii} @ p_s monotonicity of HCs $f(T_{bi}^{\text{EOS}}, M_i)$ • $K_i(p)$ not crossing for HCs• $\gamma_i^{\text{EOS}} \sim \gamma_i(M_i)$ data or correlation trend• 3-phase equilibrium test = $f(p, T, z_i)$

* C_T Characterization (Part B)

- Estimate component EOS properties Table 1 - T_{ci} p_{ci} w_i S_i

① $\gamma_i(M_i, C_f)$: vted to γ_{n+} or p_0 data/input

② (a) $T_{bi}(M_i, \gamma_i)$ correlation

$$\left. \begin{aligned} \underline{T_{ci}} \ \& \ \underline{p_{ci}} &= f(T_{bi}, \gamma_i) && \text{correlation} \\ \underline{w_i} &= f(T_{ci}, p_{ci}, T_{bi}) && \text{correlation} \end{aligned} \right\} \begin{array}{l} PVT_i \\ PVT_P \\ Prop \end{array}$$

or

$$\underline{w_i} \text{ so that } T_{bi}^{EOS} = T_{bi}(M_i, \gamma_i) \left\{ \begin{array}{l} PVT_x \\ \text{Phase Comp} \end{array} \right.$$

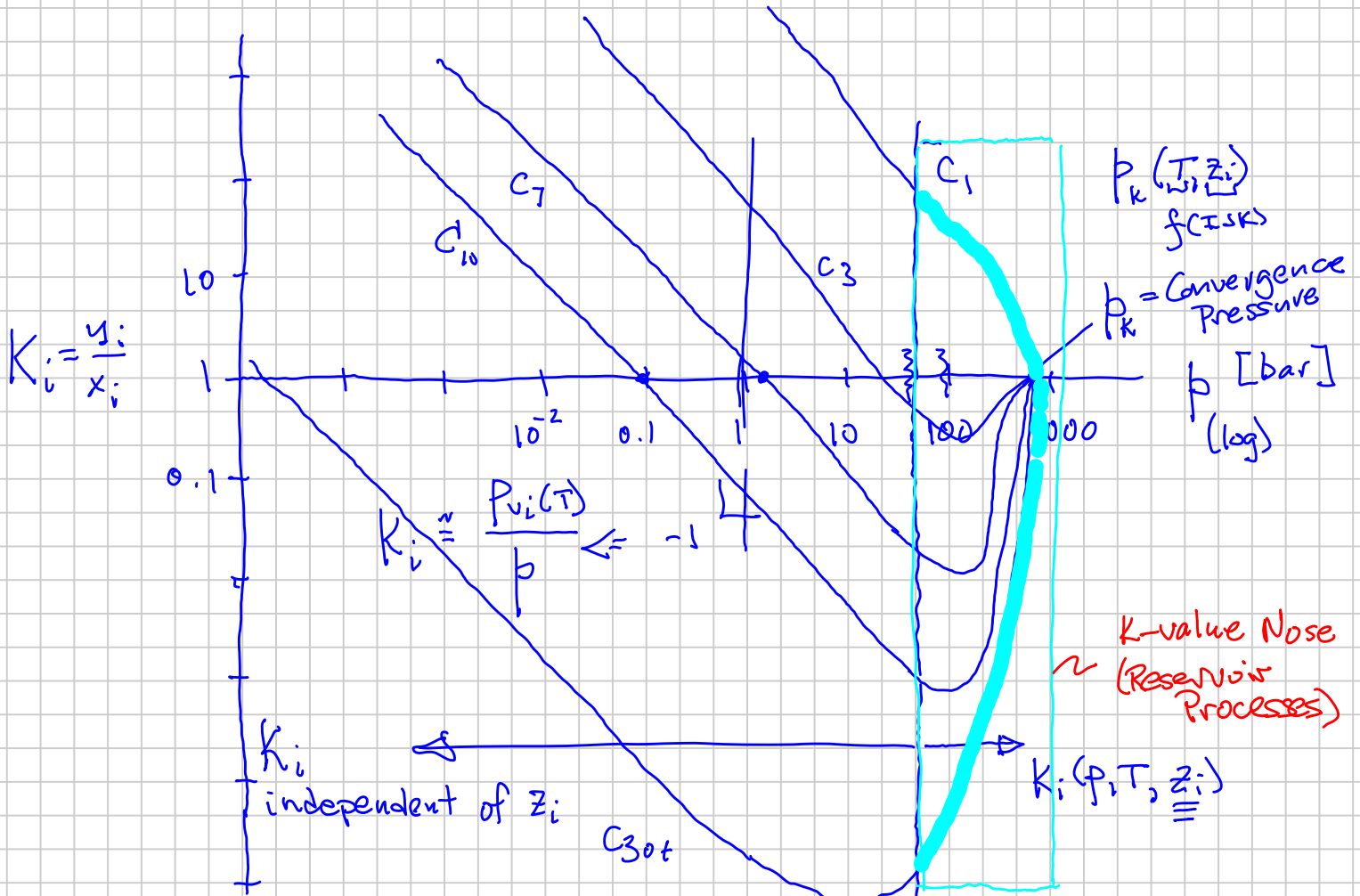
(b) PVTsim

$$\left. \begin{array}{l} SRK \\ \underline{T_{ci}} \\ \underline{p_{ci}} \\ m_i \end{array} \right\} = f(M_i, \gamma_i) \quad \left| \quad \underline{w_i} = f(m_i, EOS) \right.$$

caSep-Built

③ S_i so that $\gamma_i^{EOS} = \gamma_i(M_i, C_f)$

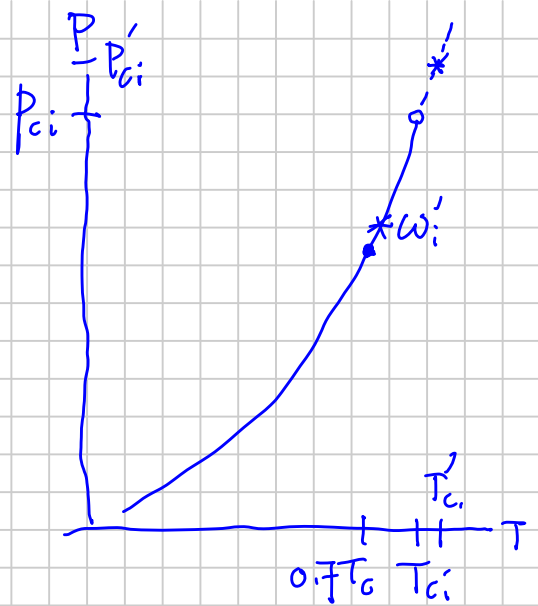
- PVTsim (<2014) S_i was not updated during regression tuning



$$\boxed{\text{EOS}} \\ p_{vi}(T) = f(T_{ci}, p_{ci}, \omega_i)$$

$$\omega_i \Rightarrow p_v @ T = 0.7 \cdot T_c$$

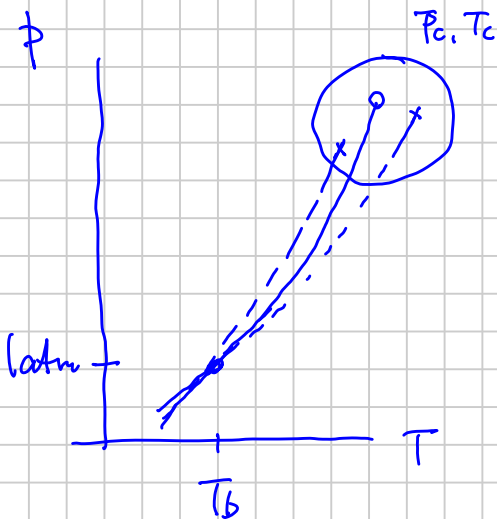
$$\omega_i \equiv -1 - \log_{10} \left(\frac{p_v @ 0.7 T_c}{p_c} \right)$$



high-p K_i (Reservoir Processes)

Nose Shape $\hat{p}_k : f(C_{7+}, C_{N+}, \underbrace{T_{ci}, p_{ci}, \omega_i}_{\equiv k_{ij}})$

Also impact
low-p K_i
 $p_{vi}(T)$



PVT_k
PhaseCamp

Have T_{bi} reliable (TBP)
for C_{7+} then used to
maintain during
tuning

PVT Lab Tests (Conventional)

	<u>CCE</u>	<u>SEP</u>	<u>DLE</u>	<u>CVD</u>
⊙ Lab Procedures	*			
⊙ Measured data				
⊙ Reported quantities	*			
• Uncertainties	*			
⊙ QC	*			
⊙ Applications (direct)				

Ch. 6

- ⊙ Covered
- ⊙ Partly Covered
- Not discussed

CCE: Constant Composition (Mass)

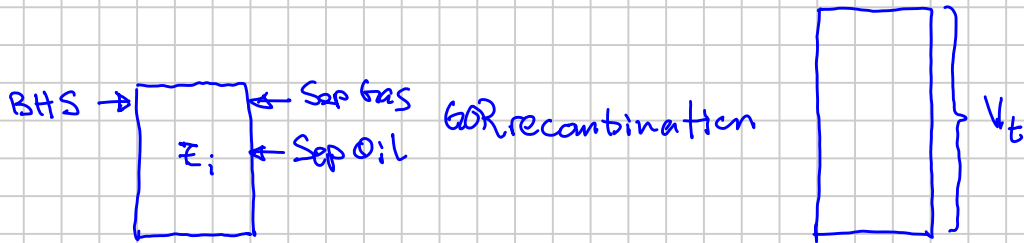
SEP: Multi-Stage Separator Test

DLE: Differential Liberation

CVD: Constant Volume Depletion

* VERY important to tuning EOS model

CCE Test

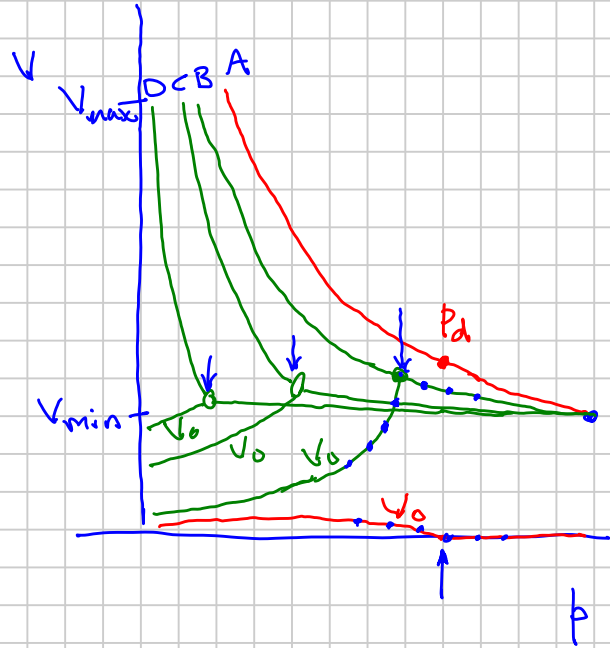
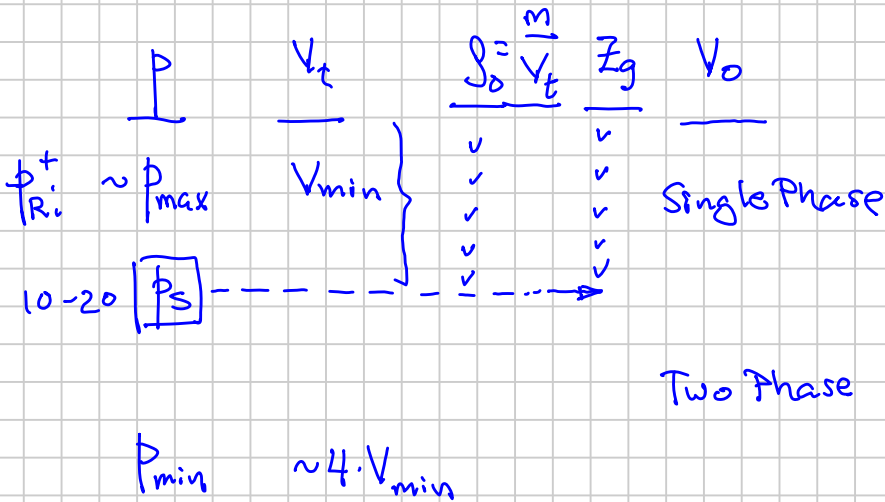


$$T = T_R \quad p_{R_i}^+ = p_i$$

P_2

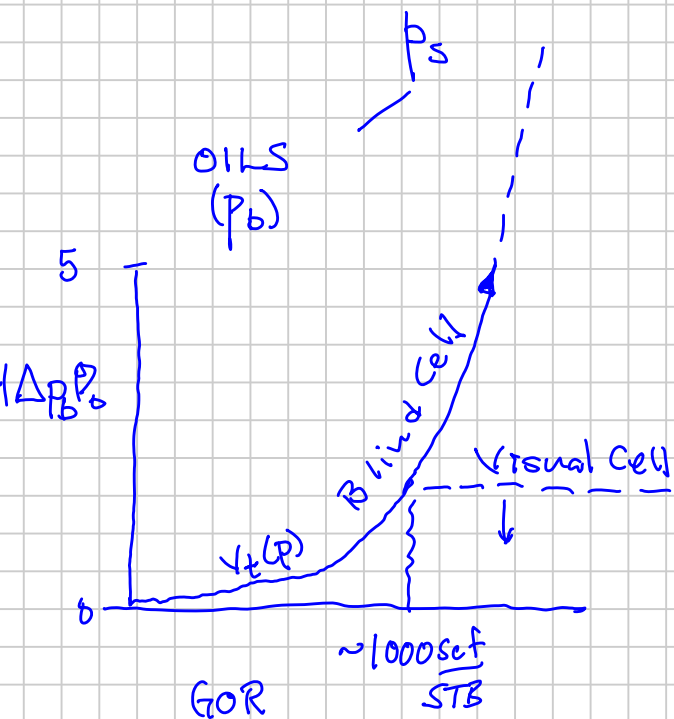
$$Z_g = \frac{PV}{nRT}$$

Know initially: $Z_i, \boxed{m}, \boxed{n}, V @ (T_R, p)$
 All Cells Visual PVT cell



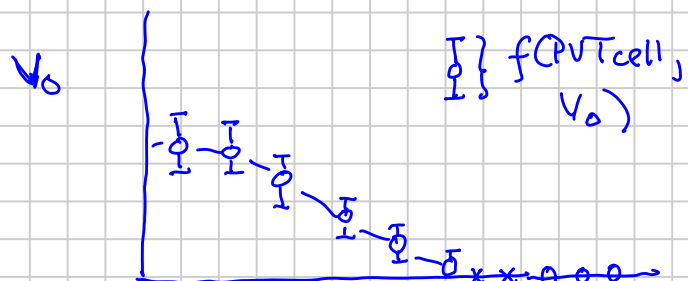
Accuracy: $\rho, Z_g \pm 1-2\%$ (expect)

$Q_{C_{gas}} = SK Z(T_R, P_r)$
 $Z_g^{SK} vs Z_{glab}$
 $\approx 3\%$



GASES (p_d)

(1-5%) \rightarrow (10% \rightarrow)
 Most liquid Dropout "Tail"



**TABLE 6.9—CCE DATA (RESERVOIR-FLUID)
FOR GOOD OIL CO. WELL 4 OIL SAMPLE**

Saturation (bubblepoint) pressure*, psig 2,620

Specific volume at saturation = $\frac{1}{\rho_{ob}}$ pressure*, ft³/lbm 0.02441

Thermal expansion of undersaturated oil at 5,000 psi = V at 220°F/V at 76°F 1.08790

Compressibility of saturated oil at $C_o = \frac{1}{V_o} \cdot \frac{\Delta V_o(5000 \rightarrow 4000)}{\Delta P}$ reservoir temperature

From 5,000 to 4,000 psi, vol/vol-psi $\left\{ \begin{array}{l} 13.48 \times 10^{-6} \\ 15.88 \times 10^{-6} \\ 18.75 \times 10^{-6} \end{array} \right.$

From 4,000 to 3,000 psi, vol/vol-psi

From 3,000 to 2,620 psi, vol/vol-psi

Pressure/Volume Relations*

Pressure (psig)	Relative volume (L) [†]	Y function [‡]
5,000	0.9639	
4,500	0.9703	
4,000	0.9771	
3,500	0.9846	
3,000	0.9929	
2,900	0.9946	
2,800	0.9964	
2,700	0.9983	
2,620**	1.0000	
2,605	1.0022	2.574
2,591	1.0041	2.688
2,516	1.0154	2.673
2,401	1.0350	2.593
2,253	1.0645	2.510
2,090	1.1040	2.422
1,897	1.1633	2.316
1,698	1.2426	2.219
1,477	1.3618	2.118
1,292	1.5012	2.028
1,040	1.7802	1.920
830	2.1623	1.823
640	2.7513	1.727
472	3.7226	1.621

* At 220°F.
 ** Saturation pressure.
 † Relative volume = V/V_{sat} in barrels at indicated pressure per barrel at saturation pressure.
 ‡ Y function = (p_{sat} - p)/(p_{abs})(V/V_{sat} - 1).

p_d

Reported Quantities:

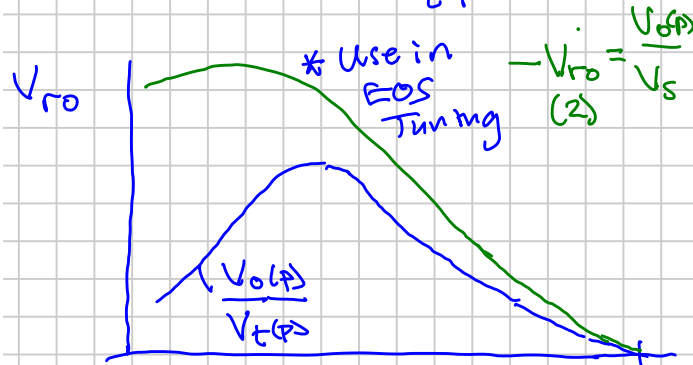
- * p_s (BP | DP)
- * $p > p_s$ ρ_o Z_g * B_g

$$B_g = \frac{p_{sc}}{p} \cdot \frac{T_R Z_g}{T}$$

*** Relative Volumes:**

$$V_{rt} \equiv V_t / V_s$$

(1) $V_{ro}^{CCE} \equiv V_o(p) / V_t(p)$



(2) $V_{ro}^{CCE} \equiv V_o(p) / V_s$ * $f(K_i)$ #

not reported, calc.

$$V_{ro} = \frac{V_o(p)}{V_s} = \frac{V_o(p)}{V_t(p)} \cdot \frac{V_t(p)}{V_s}$$

$\underbrace{\hspace{100px}}_{CCE(1)}$
 $\underbrace{\hspace{100px}}_{V_{rt}}$

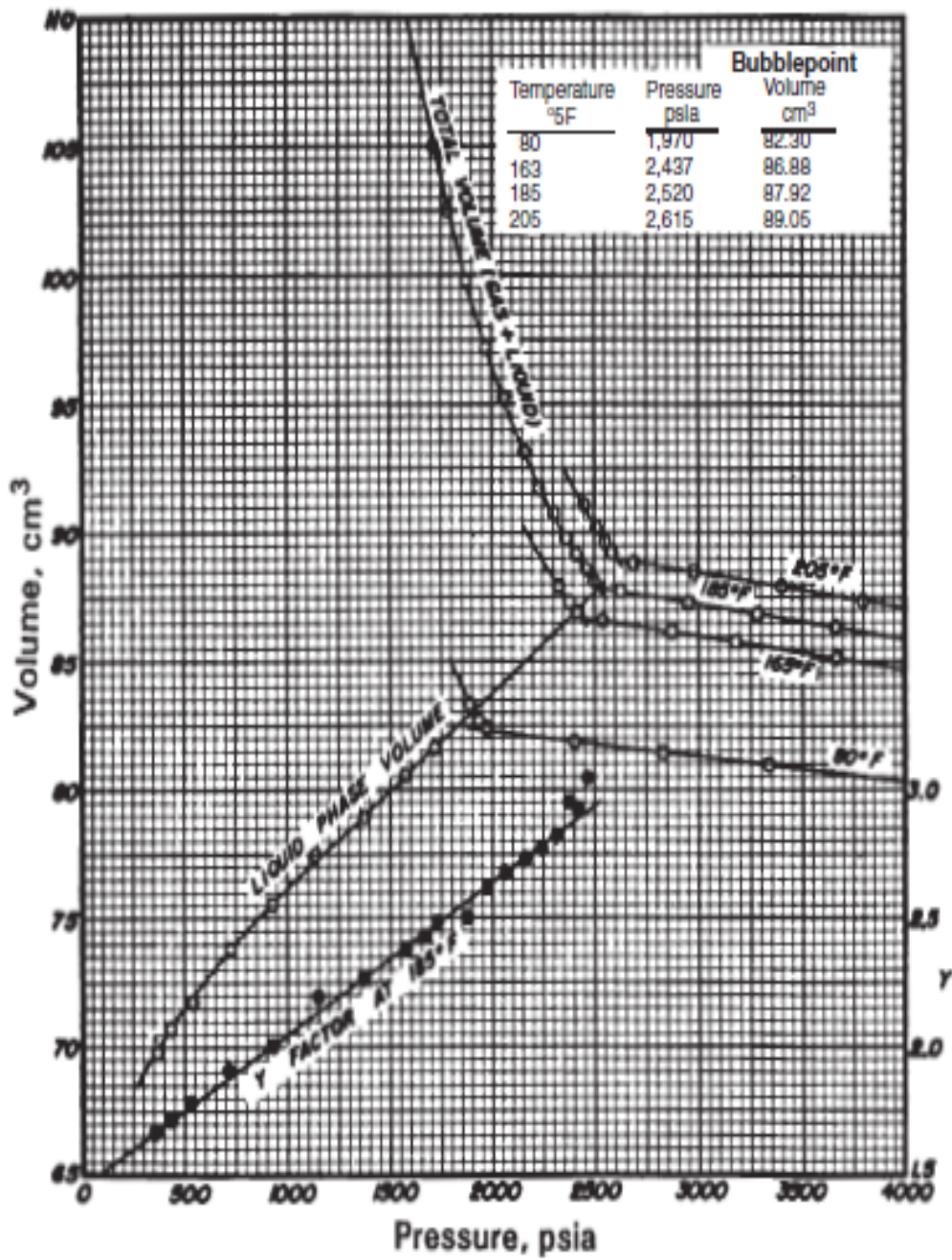
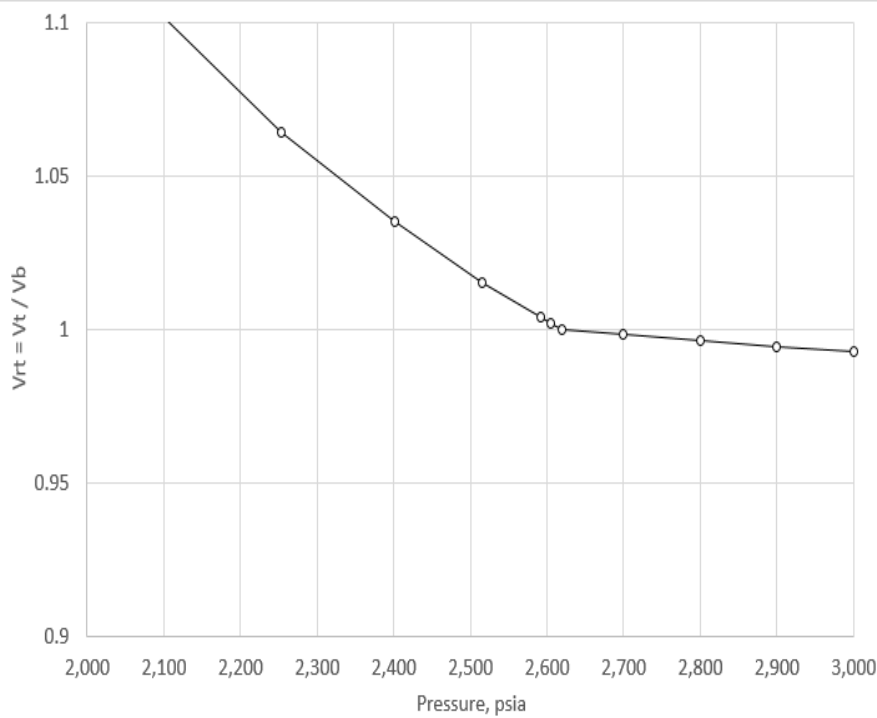
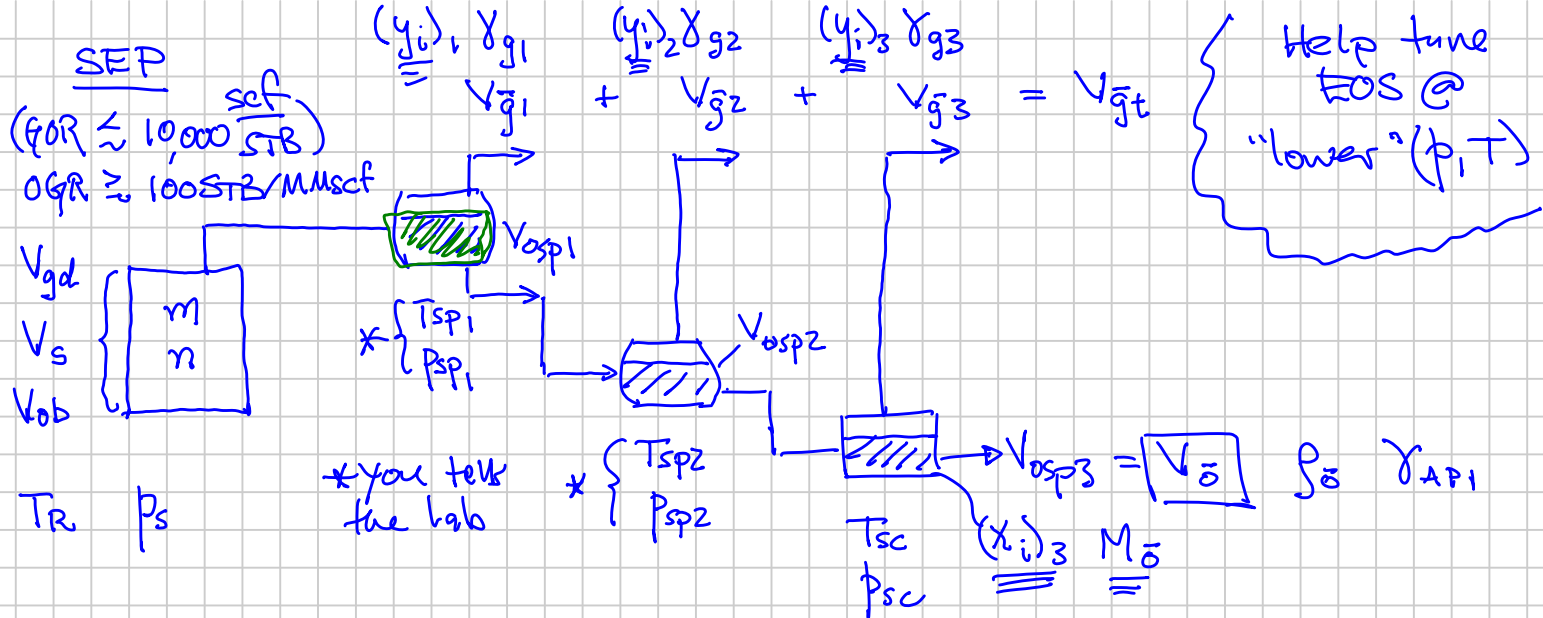


Fig. 6.5—PVT relation and plot of Y function for an oil sample at pressures below the bubblepoint.



Ch. 6 Oil



Quantities Reported

Stage	Tsp	Psp	GOR(1)	GOR(2)	Bosp or $\frac{1}{Bosp} = SF$ Sep-vol / STO	ρ_o	γ_{API}	\bar{V}_g
1			+	{ don't add !	quite unreliable (sometimes)	()		✓
2			+			()		✓
3							ρ_o	γ_{API}
			GOR_t					$\frac{1}{\bar{V}_g}$

$GOR(1) = \frac{\bar{V}_g}{V_o}$ [scf / STB]
 $GOR(2) = \frac{\bar{V}_g}{V_o(Tsp, Psp)}$ [scf / sep-bbl]

$B_{osp} = \frac{V_{osp}(Tsp, Psp)}{V_o}$

$\bar{V}_g = \frac{\sum Y_{gk} \cdot GOR(1)k}{GOR_t}$

Reflects the total average gas MW

Oil (BP) : $B_{ob} = \frac{V_{ob}}{V_o}$ [$\frac{m^3}{Sun^3}$, $\frac{bbl}{STB}$] | $R_{sb} = R_{si} = GOR_t$

Gas (DP) : $\Delta GOR_t = \frac{10^6}{GOR_t}$ [STB / MMscf] r_{si} r_{sd}

Ch. 7 "Bgd" : $B_{gd}(p) = \frac{V_{gd}}{V_{gt}}$ E100 Initial (DP) Solution OGR

↑
dry
(without
condensate)

$$\text{Geologist "B}_0\text{"} = \frac{HCPV_g \left[\frac{bbbl}{STB} \right]}{ICIP \left[\frac{STB}{STB} \right]}$$

$$= \frac{B_{gd}(P_i) \left[\frac{ft^3}{scf} \right] \cdot \frac{bbbl}{5.615 ft^3}}{r_{si} \left[\frac{STB}{MMscf} \right] \left[\frac{MMscf}{10^6 scf} \right]} \left[\frac{bbbl}{STB} \right]$$

If compositions measured: $(y_{gk})_k, x_{oi}$

Compositional Mat. Bal.

Lab Put in Cell
 Z_i

vs

Back-
Calculate

$Z_i =$

$$\frac{(\sum (y_{gk} \cdot n_{gk})) + x_{oi} \cdot n_o}{n_{gt} + n_o}$$

QC

$$(n_g)_k = (GOR_k / (RT_{sc}/P_{sc}))$$

$$n_o = V_o \cdot \frac{\sum_{oi}}{M_o}$$

QC: Material Balance on total mass:

$$\text{Reported } S_s \text{ vs } \int S_s = \frac{(\sum (m_g)_k) + m_o}{V_s}$$

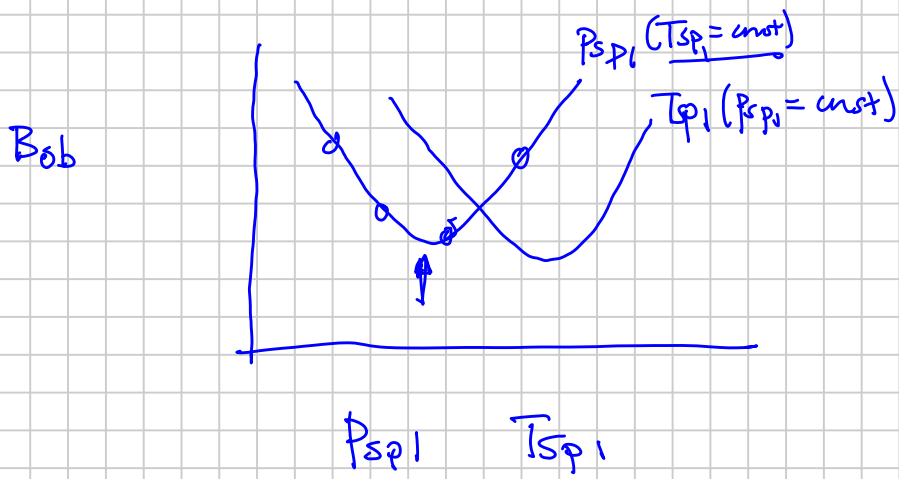
↓ $GOR_t = f(T_{sp_i}, P_{sp_i} | \dots)$

↓ Bob

$$K_i \downarrow = \frac{P_{vi}(T)}{P}$$

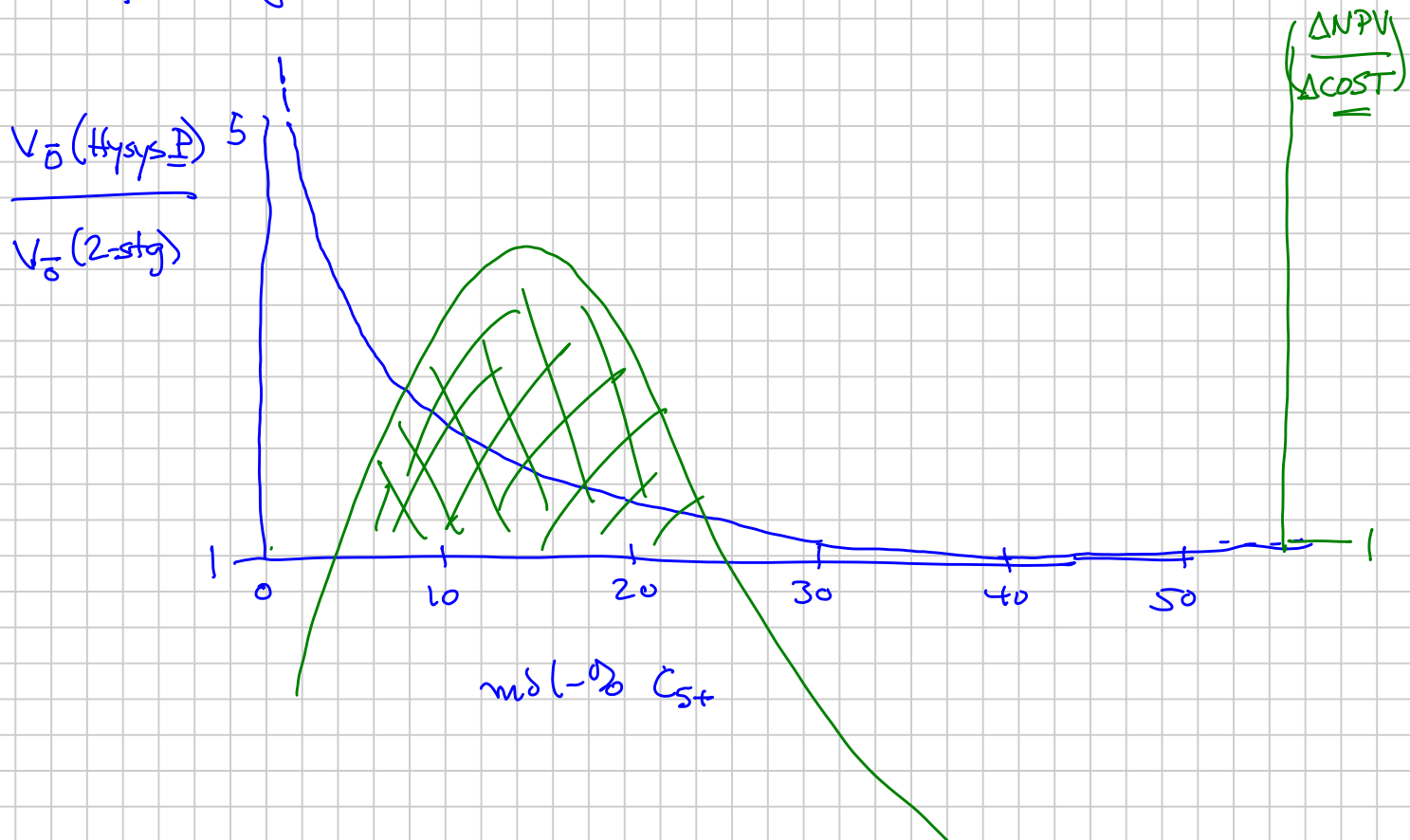
c_3, c_4, c_5, \dots

Low-p P_{sp_i} ↑ K_i
High-T T_{sp_i}
BAD



* Optimal (T_{sp1}, P_{sp1}) may change over time as the wellstream (field stream) becomes lighter (more methane).

* Optimal $V_{\bar{o}}$ is much more sensitive to (T_{sp1}, P_{sp1}) for lighter (leaner) streams:



DLE ① Only used for oils

② Blind PVT Cell

③ To get pressure dependence below p_b of

✓ - p_o (only important high k)

→ μ_o

✓ - V_o shrinkage

✓ - solution gas release

$$\left. \begin{array}{l} \checkmark - p_o \text{ (only important high } k) \\ \rightarrow \mu_o \\ \checkmark - V_o \text{ shrinkage} \\ \checkmark - \text{solution gas release} \end{array} \right\} \bar{S}_o^n \propto k_{ro} \left. \vphantom{\begin{array}{l} \checkmark - p_o \\ \rightarrow \mu_o \\ \checkmark - V_o \\ \checkmark - \text{solution gas release} \end{array}} \right\} \lambda_o$$

④ Procedure is identical to SEP

EXCEPT • $T_{sp} = T_R$

• $N_{stages} \sim 8$ (5-10)

• Last stage is a bleeding process

Reported Quantities:

p	$\frac{V_o(p)}{V_{or}} = B_{od}$	$\frac{\Delta V_g}{V_{or}} = \Delta R_{sd}$	Gas Remaining in Solution R_{sd}	p_o	Z_g	$\gamma_g (y_i)$
$p_1 = p_b$	$B_{od,b}$	0	1000			
p_2		120	880			
p_3		100	780			
\vdots						
$p_N = p_{sc}$	1.0x	50	0			

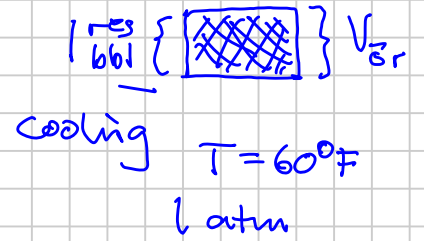
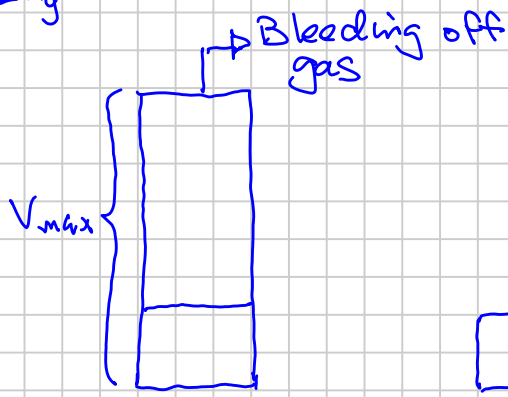
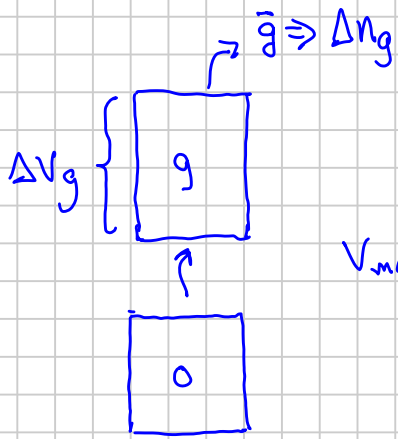
$$\begin{aligned} & \left(\sum \Delta R_{sd} \right) \\ & = \\ & R_{sd,b} \\ & = 1000 \frac{\text{scf}}{\text{res bbl}} \end{aligned}$$

- $B_{od,b} > B_{ob} \text{ (SEP)}$
- $R_{sd,b} > R_{sb} \text{ (SEP)}$
- $API_d < API \text{ (SEP)}$

$p_o = \text{calc. by material balance (Ch. 6)} =$

$$Z_g = \frac{p \Delta V_g}{\Delta V_g R T_R}$$

$$\frac{p_{or} + \sum \Delta R_{sd,k} \gamma_{g,k}}{B_{od,k}}$$



T_R
 P_{N-1}
 $\sim 250 \text{ psi}$

$P_N \rightarrow P_{sc} \rightarrow 1 \text{ atm } T_R$
 T_R

This is not used in PVT simulators } Emulate this by adding 5-10 steps from 250-15 psi

$P_{N-1} \rightarrow P_N = P_{sc}$

* Higher $T_R > 250^\circ\text{F}$ } $(B_{od}, R_{sd})_{EOS} \gg (B_{od}, R_{sd})_{lab}$
 Light oils $> 40^\circ\text{API}$

DIFFERENTIAL LIBERATION EXPERIMENT

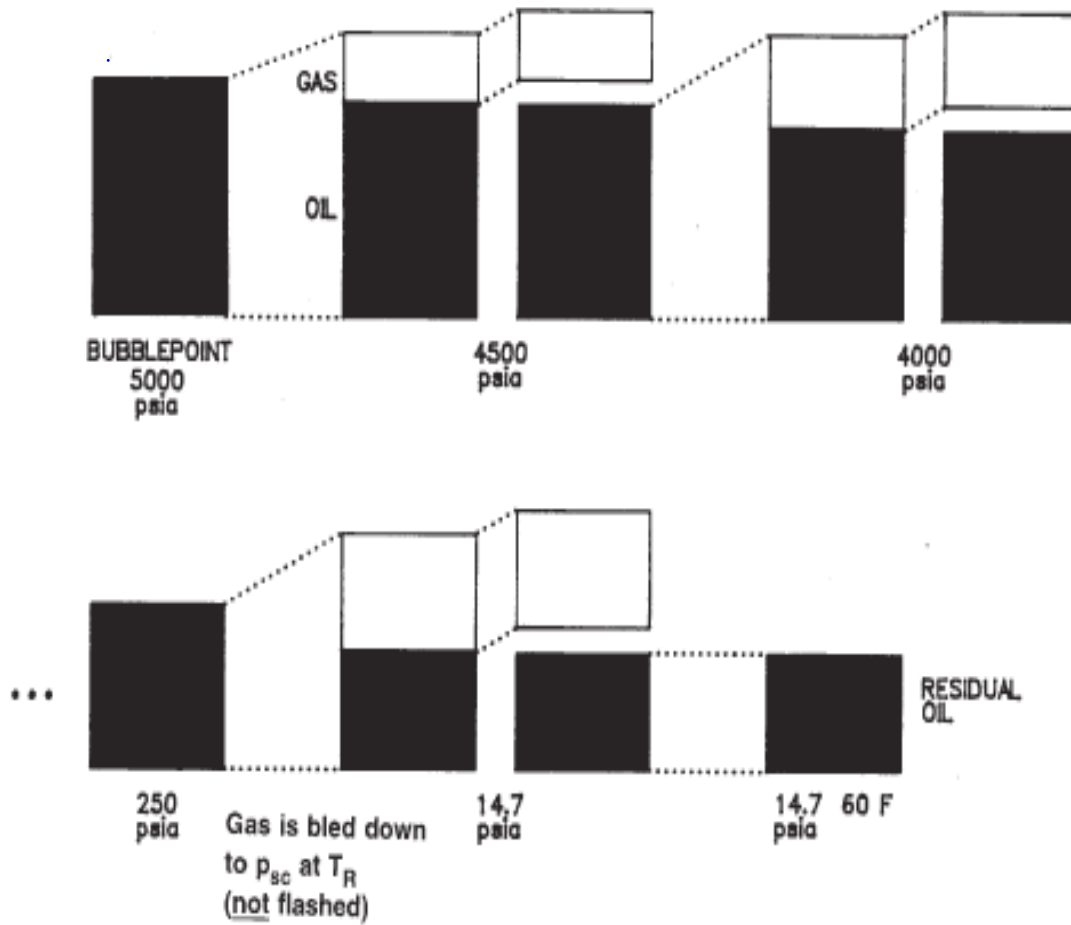


Fig. 6.6—Schematic of DLE experiment.

$$B_{td} = \frac{V_g + V_o}{V_{or}} = B_{od} + (R_{sd,b} - R_{sd}) B_g$$

Btd go Zg

TABLE 6.11—DLE DATA FOR GOOD OIL CO. WELL 4 OIL SAMPLE

Pressure (psig)	<i>R_{sd}</i>	<i>B_{od}</i>	Differential Vaporization			<i>B_g</i>	<i>γ_g</i>
	Solution GOR (scf/bbl*)	Relative Oil Volume (RB/bbl*)	Relative Total Volume (RB/bbl*)	Oil Density (g/cm ³)	Deviation Factor Z	Gas FVF (RB/bbl*)	Incremental Gas Gravity
2,620	854	1.600	1.600	0.6562			
2,350	763	1.554	1.665	0.6655	0.846	0.00685	0.825
2,100	684	1.515	1.748	0.6731	0.851	0.00771	0.818
1,850	612	1.479	1.859	0.6808	0.859	0.00882	0.797
1,600	544	1.445	2.016	0.6889	0.872	0.01034	0.791
1,350	479	1.412	2.244	0.6969	0.887	0.01245	0.794
1,110	416	1.382	2.593	0.7044	0.903	0.01552	0.809
850	354	1.351	3.169	0.7121	0.922	0.02042	0.831
600	292	1.320	4.254	0.7198	0.941	0.02931	0.881
350	223	1.283	6.975	0.7291	0.965	0.05065	0.988
159	157	1.244	14.693	0.7382	0.984	0.10834	1.213
0	0	<u>1.075</u>		0.7892			2.039
		1.000**					

DLE Viscosity Data at 220°F

Pressure (psig)	<i>Separate DLE Test</i> Oil Viscosity (cp)	<i>Calculated Gas Viscosity</i> (cp)
5,000	0.450	
4,500	0.434	
4,000	0.418	
3,500	0.401	
3,000	0.385	
2,800	0.379	
2,620	0.373	
2,350	0.396	0.0191
2,100	0.417	0.0180
1,850	0.442	0.0169
1,600	0.469	0.0160
1,350	0.502	0.0151
1,100	0.542	0.0143
850	0.592	0.0135
600	0.654	0.0126
350	0.783	0.0121
159	0.855	0.0114
0	1.286	0.0093

μ_g(P_g, T)

Gravity of residual oil = 35.1°API at 60°F.

*Barrels of residual oil.

**At 60°F.

Quantities Reported

* Don't use for "data" in EOS tuning!

$$P_i = P_d$$

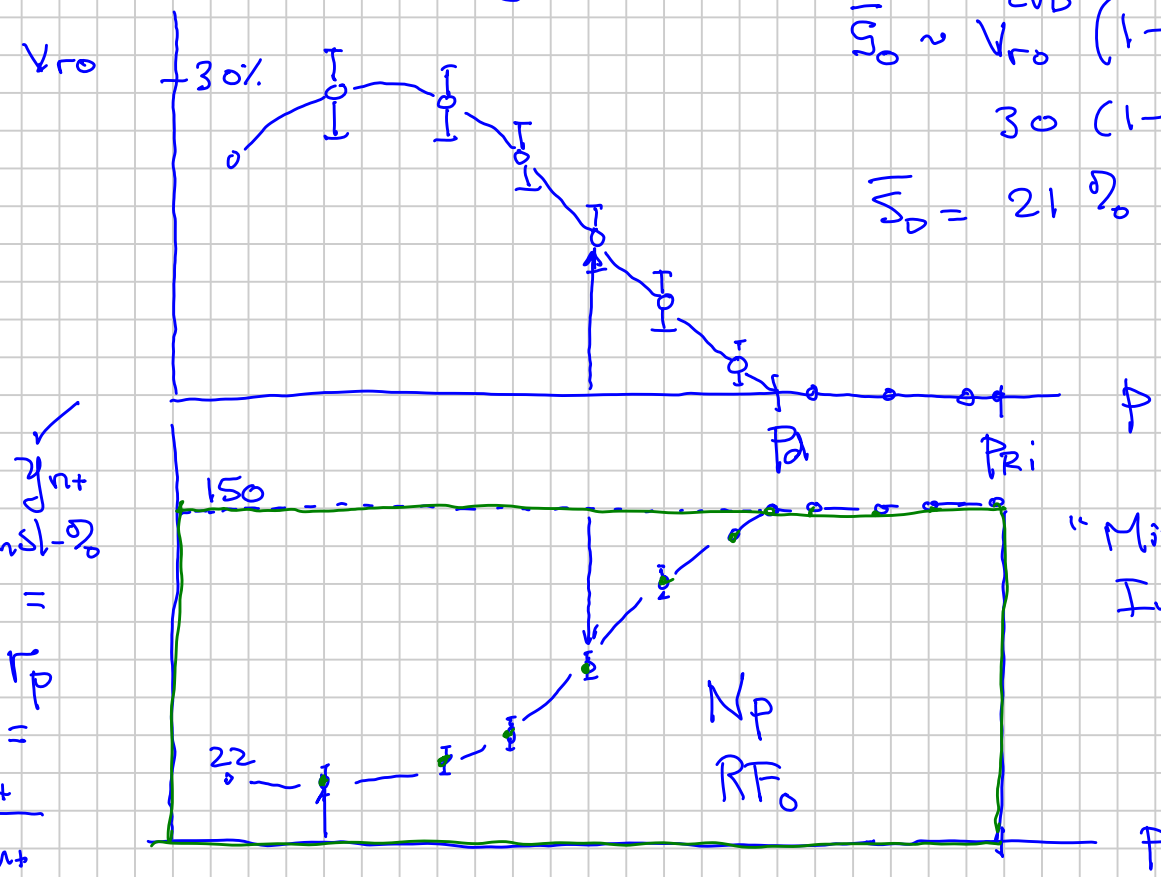
$$\frac{V_o}{V_s} = V_{ro} \%$$

$$\left[y_i \right]$$

$$\underbrace{M_{nt} \quad Y_{nt}}_C$$

$$\frac{n_p}{n_d} \left[\frac{\text{mol}}{\text{mol}} \right] \left[\frac{\text{scf}}{\text{scf}} \right]$$

$$\left[Z_g \right] \left[Z_2 \right]$$



$$\bar{S}_o \sim V_{ro} (1 - \bar{S}_w)$$

$$30 (1 - 0.3)$$

$$\bar{S}_o = 21 \% \quad k_{ro} \sim 0$$

$$CGR = \frac{y_{nt}}{1 - y_{nt}}$$

How to QC y_{nt} data in CVD:

- ① Forward Material Balance
 - CL Ruedelberger-Hinds
 - Whitsun-Torp

Back-Calc

$$n(z_i)_i = \Delta n_g y_i + n_g y_i + \hat{n}_o x_i$$

$$V_s \cdot (1 - V_{ro}) = \underbrace{V_g, P_i, T_i, Z_g}_{n_g}$$

$$\hat{n}_o = n - \Delta n_g - n_g$$

$$K_i \approx \frac{y_i}{x_i}$$

> 150 STB/MMscf

useful, accurate ONLY if you have "richer" G.C.

Very sensitive V_{ro} ✓

* (2) Backward Material Balance

BEST QC

N-1 or N

$$\underline{n_{or}} = V_{ro} \cdot \left(\frac{S_o}{M_o} \right)$$

$$(x_i)_{N-1}$$

$$\text{Initial } (z_i) = (z_{cell})_{k=1} = \left[n_{or} (x_i)_{or} + \sum_{k=N-1}^{k=1} \Delta n_{g,k} y_{i,k} \right] / \left(n_{or} + \sum_{k=N-1}^1 \Delta n_{g,k} \right)$$

vs

Lab (z_i) \Rightarrow Determine Lab "Losses" by component C_{6+}

$$(z_{6+})_{Lab} = 12 \text{ mol-\%} \quad , \quad OGR = 200 \text{ STB/MMscf}$$

$$(z_{6+})_{BMR} = 10 \text{ mol-\%} \quad : \quad = 160$$

Two-Phase Z-factor Z_2

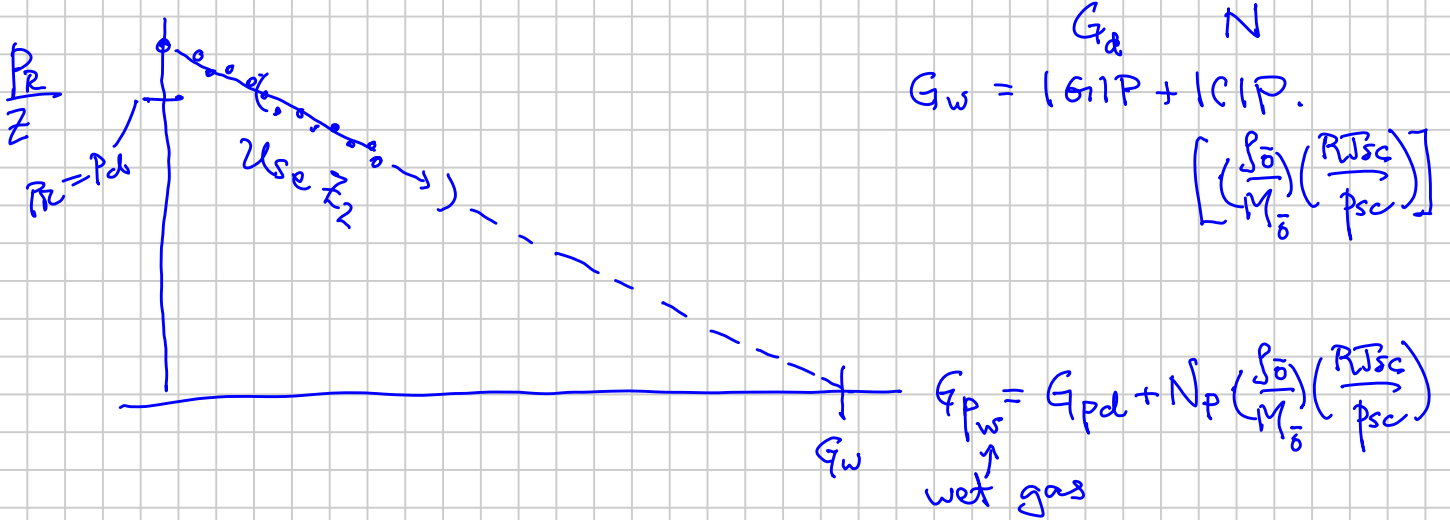
* Only be used for one application!

- Using a traditional Gas Material Balance

$$\left[1 - c_e(p_i - p)\right] \frac{p}{Z} = \left(\frac{p}{Z}\right)_i \left[1 - \frac{G_p}{G}\right]$$

Water

Z to use here at $p_r < p_d$: Z_2



Separator Liquid:

Cylinder ID No. W-209*

Cylinder ID No. W-357

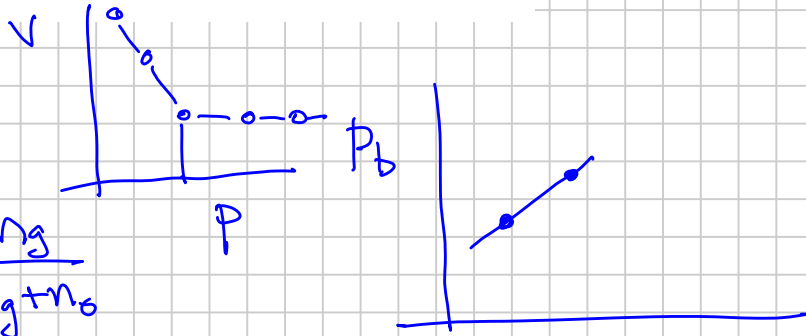
BP Pressure Temperature

{ 595 psig 77 °F }
 { 594 psig 77 °F } ✓

$P_{sp} = 650 \text{ psi}$

$T_{sp} = 940 \text{ F}$

Report Date: 7/17/2008



Eq. 6 $GOR_{sp}, P_{osp}, M_{osp} \Rightarrow F_g = \frac{n_g}{n_g + n_o}$

$z_{wi} = F_g y_i + (1 - F_g) x_i \Rightarrow C_i \Rightarrow F_g = \frac{z_i - x_i}{y_i - x_i} = 0.7805$

Component	SEPARATOR GAS		SEPARATOR OIL		WELLSTREAM	
	y_i Mole%	*	x_i Mole %	Liquid Volume %	z_{wi} Mole %	*
Hydrogen Sulfide	0.000	0.000	0.000	0.000	0.000	0.000
Nitrogen	0.484	0.000	0.072	0.018	0.393	0.000
Carbon Dioxide	2.037	0.000	0.847	0.331	1.776	0.000
Methane	82.531	0.000	17.472	6.788	68.250	0.000
Ethane	7.266	1.933	6.165	3.779	7.024	1.868
Propane	5.155	1.411	11.852	7.476	6.625	1.813
Iso-butane	0.779	0.253	3.689	2.765	1.418	0.461
N-butane	1.047	0.328	6.843	4.942	2.319	0.727
2-2 Dimethylpropane	0.000	0.000	0.066	0.058	0.014	0.006
Iso-pentane	0.228	0.083	3.026	2.538	0.842	0.306
N-pentane	0.187	0.067	3.169	2.631	0.842	0.303
2-2 Dimethylbutane	0.004	0.002	0.087	0.083	0.022	0.009
Cyclopentane	0.022	0.006	0.000	0.000	0.017	0.005
2-3 Dimethylbutane	0.000	0.000	0.473	0.444	0.104	0.042
2 Methylpentane	0.037	0.015	1.125	1.070	0.276	0.114
3 Methylpentane	0.019	0.008	0.679	0.636	0.164	0.067
Other Hexanes	0.000	0.000	0.000	0.000	0.000	0.000
n-Hexane	0.039	0.016	1.614	1.521	0.385	0.157

HOFFMAN PLOT

EQUILIBRIUM CHECK of SEPARATOR LIQUID and GAS COMPOSITIONAL ANALYSES

Separator Pressure = 650 psig

Separator Temperature = 94 °F

Components	Gas (X) Mole %	Oil (Y) Mole %	Equil. Ratio (K=Y/X)	K*Psep (psiA)	Normal BP (NBP) °R	$T_{NBP}^{-1} - T_{SEP}^{-1}$	Critical Pressure (Pc) psiA	Critical Temperature (Tc) °R	B-Factor	Graph Results	
										B(1/Tb-1/Tsp)	Log(K*Psep)
N2	0.484	0.072	6.758	4491.50	139	0.005373	493	227	551	2.958	3.652
CO2	2.037	0.847	2.404	1598.02	350	0.001048	1071	548	1811	1.898	3.204
C1	82.531	17.472	4.724	3139.57	201	0.003169	668	343	805	2.552	3.497
C2	7.266	6.165	1.179	783.39	332	0.001204	708	550	1413	1.701	2.894
C3	5.155	11.852	0.435	289.10	416	0.000598	616	666	1799	1.076	2.461
IC4	0.779	3.689	0.211	140.35	471	0.000319	529	735	2038	0.650	2.147
NC4	1.047	6.843	0.153	101.70	491	0.000231	551	765	2158	0.498	2.007
IC5	0.228	3.092	0.074	49.01	542	0.000040	490	829	2383	0.095	1.690
NC5	0.187	3.169	0.059	39.22	557	-0.000009	489	845	2483	-0.023	1.594
C6	0.121	3.978	0.030	20.22	615	-0.000181	437	913	2784	-0.504	1.306
C7+	0.165	42.822	0.004	2.56	763	-0.000496	332	1070	3607	-1.789	0.408
Total	100.000	100.000									

Hoffman - Crump - Hocott

$$K_i = \frac{P_{vi}(T)}{p} \checkmark$$

Edmister $f_{vi}(P_{ci}, T_{ci}, T_{bi}) \checkmark$

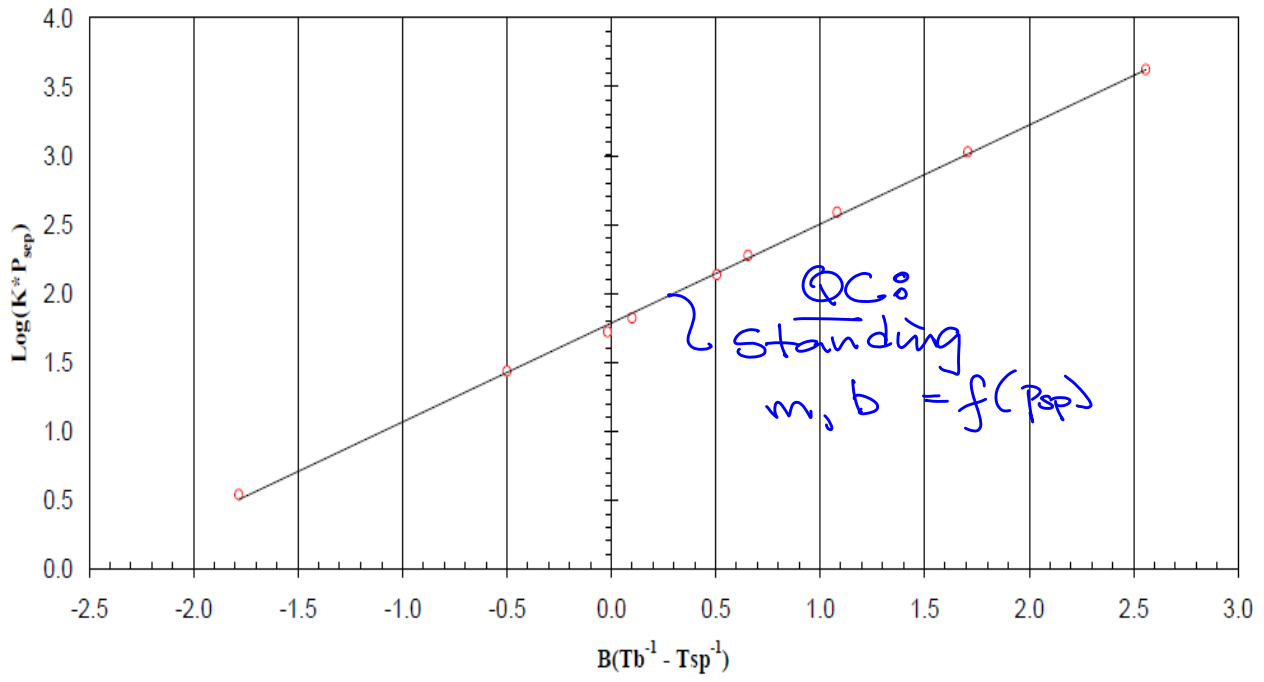
$$K_i p = A_0 + A_1 F_i$$

$$F_i = b_i \left(\frac{1}{T_{bi}} - \frac{1}{T_{sp}} \right)$$

$$b_i = \frac{\log(P_{ci}/P_{sc})}{\left(\frac{1}{T_{bi}} - \frac{1}{T_{ci}} \right)}$$

Ch. 3

Smooth Line = Equilibrium Composition



SEPARATOR CONDITIONS and FLUID PROPERTIES

Conditions	Pressure psia	Temperature °F	GOR (1)	Separator Oil Volume Factor (2)	Oil Density (3)	Gas Specific Gravity (4)
1st Stage Separator	665	94	N/A	1.2573	0.7071 ✓	0.705 ✓
2nd Stage Separator	91	80	263	{ N/A }	N/A	0.926 ✓
3rd Stage Separator	45	120	47	{ N/A }	N/A	1.264 ✓
Ambient Lab Condition	14.65	75	104	→ 1.0079	0.7888 ✓	1.758 ✓
Stock Tank	14.65	60	0	1.0000	0.7961 ✓	1.758
TOTALS	-----	-----	414		-----	-----

Stock Tank Oil Gravity: 46.06 °API at 60 °F

- (1) Gas-Oil Ratio (GOR) is the cubic feet of gas at standard conditions per barrel of stock tank oil.
- (2) Barrels of oil at indicated separator conditions per barrel of stock tank oil.
- (3) Oil Density (g/cc) at indicated separator conditions.
- (4) Air = 1.000

$$GOR_w = \overset{\text{Test}}{4320} + 414 = 4734 \frac{\text{scf}}{\text{STB}}$$

$\left(\frac{\text{scf}}{\text{STB}} \right)$ scf/Mscf

$$CGR_t = \gamma_{si} = \frac{10^6}{4734} = 211 \frac{\text{STB}}{\text{Mscf}}$$

scf/STB

TABLE 3

PRESSURE-VOLUME RELATION
OF
A 3436 Scf/Sep Bbl RESERVOIR FLUID AT 263 °F
(Constant Composition Expansion)

$$\left(\frac{G_p}{G_d}\right) = 1 - \frac{(p/z)_d}{(p/z)_i} =$$

$$\frac{1}{B_g} \left[\frac{M_{scf}}{RB} \right]$$

p Pressure, (psig)	V _{rt} Relative Volume	ρ Density, (g/cc)	Y-Function (1)	V _{ro} = V _o /V _s Retrograde Liquid Volume		Gas Deviation Factor, Z _g	Gas Expansion Factor, (4)
				% of HC Pore Volume (2)	Bbls / MMscf (3)		
11000	0.82736	0.47758	N/A	N/A	N/A	1.77731	1.69183
10440 Pres	0.83715	0.47199	N/A	N/A	N/A	1.70691	1.67193
10000	0.84486	0.46769	N/A	N/A	N/A	1.65013	1.65657
9000	0.86686	0.45581	N/A	N/A	N/A	1.52404	1.61426
8000	0.89274	0.44260	N/A	N/A	N/A	1.39542	1.56715
7000	0.92595	0.42673	N/A	N/A	N/A	1.26675	1.51055
6500	0.94639	0.41751	N/A	N/A	N/A	1.20243	1.47769
6000	0.97085	0.40699	N/A	N/A	N/A	1.13883	1.44019
5535 Psat	1.00000	0.39513	N/A	0.00%	0.000	1.08234	1.39792
5178	1.03101	N/A	2.21679	4.26%	30.252	N/A	N/A
4774	1.07253	N/A	2.19115	15.84%	112.422	N/A	N/A
4280	1.13887	N/A	2.10425	21.07%	149.544	N/A	N/A
3898	1.20738	N/A	2.01749	22.84%	162.094	N/A	N/A
3456	1.31271	N/A	1.91560	23.83%	169.096	N/A	N/A
3070	1.43810	N/A	1.82404	24.61%	174.671	N/A	N/A
2446	1.74876	N/A	1.67658	24.16%	171.474	N/A	N/A
1827	2.30634	N/A	1.54126	22.97%	163.019	N/A	N/A
1383	3.05722	N/A	1.44403	21.93%	155.621	N/A	N/A
938	4.56767	N/A	1.35256	20.78%	147.430	N/A	N/A

(1) Y-Function = Dimensionless Compressibility = $(P_{sat} - P_1) * [P_1 * (RV_1 - 1)]^{-1}$

(2) Retrograde liquid volume at the indicated pressure and reservoir temperature as a percent of the hydrocarbon pore volume at the dew point pressure and reservoir temperature.

(3) Retrograde liquid volume at the indicated pressure and reservoir temperature (Bbls) per volume of gas (MMscf) at the dew point pressure and reservoir temperature.

(4) Gas Expansion Factor = the volume of surface gas at standard conditions (Mscf) produced from one barrel of undersaturated gas at the indicated pressure and reservoir temperature.

TABLE 4

RESERVOIR GAS DEPLETION STUDY AT 263 °F

Reservoir Pressure, psig	(D.P.) 5535	4500	3500	2500	1700	900	0
Wellstream Components	mole %	mole %	mole %	mole %	mole %	mole %	mole %
Hydrogen Sulfide	0.000	0.000	0.000	0.000	0.000	0.000	0.000
Nitrogen	0.393	0.405	0.433	0.442	0.437	0.428	0.335
Carbon Dioxide	1.776	1.829	1.863	1.899	1.936	1.945	1.612
Methane	68.250	71.913	74.339	75.538	76.049	74.723	60.118
Ethane	7.024	7.178	7.244	7.464	7.621	7.691	7.178
Propane	6.625	6.501	6.501	6.501	6.501	7.112	8.091
Iso-butane	1.418	1.352	1.312	1.300	1.324	1.462	1.991
N-butane	2.319	2.178	2.113	2.113	2.113	2.360	3.243
Iso-pentane	0.857	0.800	0.766	0.724	0.690	0.815	1.213
N-pentane	0.842	0.787	0.757	0.716	0.681	0.762	1.151
Hexanes	0.968	0.857	0.771	0.701	0.647	0.701	1.268
Heptanes Plus	9.529	6.200	3.900	2.600	2.000	2.000	13.800
TOTALS	100.000	100.000	100.000	100.000	100.000	100.000	100.000

HEPTANES PLUS (C₇₊) FRACTION CHARACTERISTICS

Molecular Weight	169.530	149.703	136.501	126.627	120.849	116.400	131.537
Specific Gravity	0.8163	0.8013	0.7897	0.7803	0.7746	0.7699	1.3869

CONDENSED RETROGRADE LIQUID VOLUME

HC Pore Volume %	0.000	17.000	22.842	23.832	22.309	19.063	14.967
Bbls/MMscf of DP Gas	0.000	120.639	162.094	169.122	158.316	135.277	106.213

GAS DEVIATION FACTOR

Equilibrium Gas	1.0823	0.9202	0.8515	0.8369	0.8568	0.9069	N/A
Two-Phase	1.0823	0.9568	0.8786	0.8169	0.7645	0.6848	N/A

CUMULATIVE PRODUCED WELLSTREAM VOLUME

Vol % of Initial DP Gas	0.000	7.976	21.980	39.965	56.259	73.951	93.845
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GPM FROM CVD WELLSTREAM COMPOSITIONS

Propane plus (C ₃₊)	10.246	7.476	5.859	5.002	4.617	4.972	12.546
Butanes plus (C ₄₊)	8.432	5.697	4.080	3.223	2.837	3.026	10.332
Pentanes plus (C ₅₊)	7.245	4.575	2.991	2.138	1.744	1.811	8.668

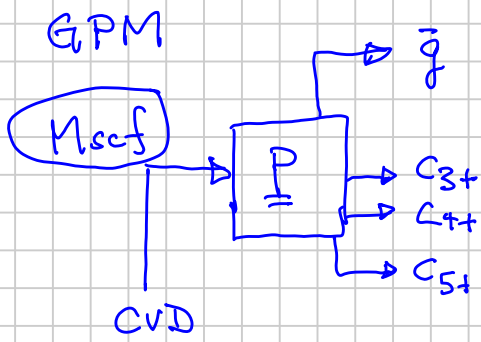
y_i

z

v_{ro}
CVD

z_{1g}
z₂

v_{pd}



gallons / Mscf
wellstreams

$$OGR = \frac{(V_{3+})_b}{V_g (C_{2-})}$$

$$GPM = \left[y_{3+} \cdot \frac{(M_{3+})}{(\rho_{h3+})} \right] (17)$$

gallons
Ideal
Volume
Mixing

$$\frac{1000 \text{ scf}}{379 \text{ scf/lbms lb}}$$