

①

$$(a) \gamma_{\bar{g}} = 0.7 \quad \gamma_{API} = 35 \quad \gamma_{\bar{o}} = \frac{141.5}{131.5 + \gamma_{API}} = 0.8498$$

$$\rho_{\bar{g}} = \rho_{air,x} \cdot \gamma_{\bar{g}} = (1.223)(0.7)$$

$\rho_{\bar{g}} = 0.856 \text{ kg/m}^3$

$$\rho_{\bar{o}} = \gamma_{\bar{o}} \bar{\rho}_w$$

$= (0.8498)(999)$

$\rho_{\bar{o}} = 849 \text{ kg/m}^3$

$$(b) \text{ At } p = p_b = 275 \text{ bara}$$

$$B_o = 1.42 \text{ m}^3/\text{Sm}^3$$

$$R_s = 150 \text{ Sm}^3/\text{Sm}^3$$

$$r_s = 3 \cdot 10^{-5} \text{ Sm}^3/\text{Sm}^3$$

$$b_{gd} = 210 \text{ Sm}^3/\text{m}^3 \Rightarrow B_{gd} = 0.00476 \text{ m}^3/\text{Sm}^3$$

$$(c) \rho_o = \frac{\rho_{\bar{o}} + \rho_{\bar{g}} R_s}{B_o}$$

$$= \frac{849 + 0.856(150)}{1.42}$$

$$\rho_{\bar{g}} = \frac{\rho_{\bar{o}} + \rho_{\bar{g}} r_s}{B_{gd}}$$

$$= \frac{0.856 + 849(3 \cdot 10^{-5})}{0.00476}$$

$$\rho_o = 688 \text{ kg/m}^3$$

$$\rho_{\bar{g}} = 185 \text{ kg/m}^3$$

$$(d) p_o(D_{woc}) = p_o(D_{soc}) + \rho_o g (D_{woc} - D_{soc})$$

$$= 275 + \underbrace{688(9.8)(500)}_{\text{Pa}} \cdot 10^{-5} \text{ bar/Pa}$$

$$\rho_o(D_{woc}) = 308.7 \text{ bara}$$

$$p_g(D_{top}) = p_g(D_{soc}) + \rho_g g (D_{top} - D_{soc})$$

$$= 275 + 185(9.8)(-200) \cdot 10^{-5}$$

$$\rho_g(D_{top}) = 271.4 \text{ bara}$$

(e) B_0 @ $p = 309 \text{ bara} > p_b$ (undersaturated)

$$\underline{\underline{B_0 = 1.405 \text{ m}^3/\text{Sm}^3}}$$

graphically (linearly) extrapolated

(f) $N_o = HCPV_o / \bar{B}_{oi}$

$$= 2.5 \cdot 10^8 / 1.4125$$

$$\bar{B}_{oi} = \frac{1}{2}(1.42 + 1.405) = 1.4125 \text{ m}^3/\text{Sm}^3$$

$$\underline{\underline{N_o = 1.77 \cdot 10^8 \text{ Sm}^3}}$$

$$G_o = N_o \cdot \bar{R}_{Si} \\ = (1.77 \cdot 10^8)(150)$$

$$\underline{\underline{G_o = 2.655 \cdot 10^{10} \text{ Sm}^3}}$$

$$G_g = HCPV_g / \bar{B}_{gdi}$$

$$= 10^8 / 0.00476$$

$$\underline{\underline{G_g = 2.1 \cdot 10^{10} \text{ Sm}^3}}$$

$$N_g = G_g \bar{r}_{Si} \\ = (2.1 \cdot 10^{10})(3 \cdot 10^{-5})$$

$$\underline{\underline{N_g = 6.30 \cdot 10^5 \text{ Sm}^3}}$$

$$\underline{\underline{10IP = 1.776 \cdot 10^8 \text{ Sm}^3}}$$

$$\underline{\underline{1GIP = 4.755 \cdot 10^{10} \text{ Sm}^3}}$$

(g) $M_{\bar{g}} = M_{air} \cdot \delta_{\bar{g}}$
 $= 28.97(0.7)$

$$\underline{\underline{M_{\bar{g}} = 20.28}}$$

$$M_{\bar{o}} = \frac{6084}{Y_{API} - 5.9}$$

$$\underline{\underline{M_{\bar{o}} = 209}}$$

$$(h) \quad B_{gd} = B_{gw} (1 + C_{\bar{g}} r_s) \quad \text{Eq. 7.12}$$

$$C_{\bar{g}} = 133000 \quad \frac{\gamma_{\bar{g}}}{M_{\bar{g}}} = 133000 \frac{0.8498}{209} = 541 \text{ scf/STB} \\ = 96.3 \text{ Sm}^3/\text{Sm}^3$$

$$B_{gw} = \frac{P_{sc}}{T_{sc}} \frac{T_R z_g}{P_{ri}}$$

Solve for z_g given $B_{gd} = 0.00476 \text{ m}^3/\text{Sm}^3$

$$z_g = \frac{B_{gd}}{(1 + C_{\bar{g}} r_s)} \cdot \left(\frac{T_{sc}}{T_R} \cdot \frac{P_{ri}}{P_{sc}} \right) \\ = \frac{0.00476}{1 + (96.3)(3 \cdot 10^{-5})} \left(\frac{15.56 + 273}{100 + 273} \right) \left(\frac{273}{1.0135} \right)$$

$$\underline{z_g = 0.996}$$

Standing-Katz Chart Method

$$\gamma_w = \gamma_{gr} = \gamma_g = \frac{\gamma_{\bar{g}} + 4580 r_p \gamma_{\bar{o}}}{1 + 133000 r_p \left(\frac{\gamma_{\bar{o}}}{M_{\bar{o}}} \right)} \quad \text{Eq. 3.55}$$

$$r_p = r_s = 3 \cdot 10^{-5} \frac{1}{5.615} \\ = 5.3 \cdot 10^{-6} \text{ STB/scf}$$

$$\gamma_g = \frac{0.7 + 4580 (5.3 \cdot 10^{-6})(0.8498)}{1 + 133000 (5.3 \cdot 10^{-6}) \left(\frac{0.8498}{209} \right)}$$

$$\underline{\gamma_g = 0.718 \sim 0.72}$$

Fig. 3.7 "Condensate Fluid" Curves

$$P_{pc} = 665 \text{ psia} = 45.85 \text{ bara}$$

$$T_{pc} = 385^\circ R = 214 \text{ K}$$

$$P_{pr} = 273 / 45.85 = 6$$

$$T_{pr} = (100 + 273) / 214 = 1.75$$

$$\underline{(z_g)_{\text{chart}} = 0.935}$$

$$(i) \quad B_{od,b} > (B_{od})_{SEPT test}$$

because the DLE Process is at $T = T_R$ which leads to higher K_i for heavier (C_{7+}) components and thus more vaporization of the reservoir oil. That shrinks the final oil @ P_{sc} relative to a multistage flash process at lower temperatures.

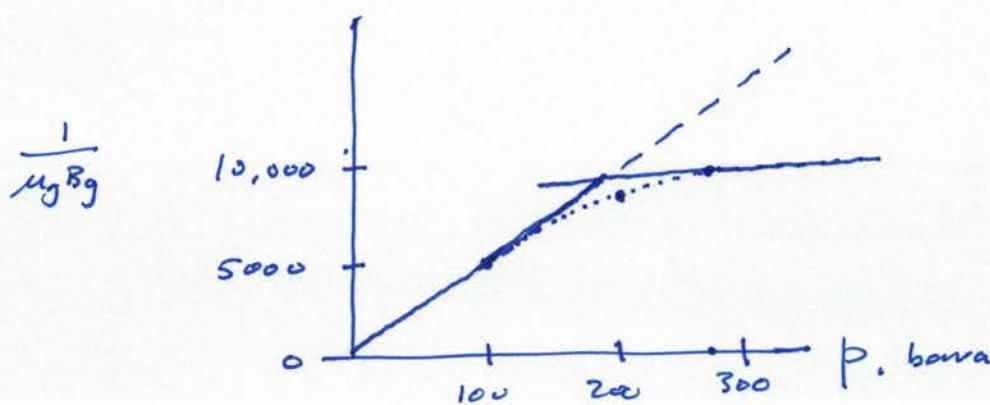
(j) $\mu_o (P > P_b)$ increases at lower pressures because
 (1) gas is coming out of solution, increasing μ_o
 (2) oil density increases, increasing μ_o

These two reasons are inter-related, so decreasing R_s is really the main reason.

$\mu_o (P > P_b)$ increases at higher pressures because S_o increases.

(k)

<u>P</u>	<u>b_g</u>	<u>μ_g</u>	<u>$\frac{1}{\mu_g b_g}$</u>
275	222	0.0208	10673
200	158	0.0177	8926
100	80	0.0153	5229



$$(l) \quad n_{\bar{g}g} = G_g / (RT_{Sc}/P_{Sc})$$

$$R T_{sc} / p_{sc} = 23.68 \text{ } \frac{\text{m}^3}{\text{kg-mole}}$$

$$n\bar{g}_o = G_o / (RT_{sc}/\rho_{sc})$$

$$g_0 = 849 \text{ kg/m}^3$$

$$n_{\bar{g}} = N_g \left(\frac{s_0}{M_0} \right)$$

$$M_{\bar{0}} = 209 \text{ kg/kg-mole}$$

$$n_{\bar{o}o} = N_o \left(\frac{s_{\bar{o}}}{M_{\bar{o}}} \right)$$

$$\left(\frac{g_0}{M_D}\right) = 4.06 \text{ kg-mole / } \text{Sm}^3$$

$$n_{\bar{g}g} = (2.1 \cdot 10^{10}) / (23.68) = 8.87 \cdot 10^8 \text{ kg mole}$$

$$n_{\text{go}} = (2.655 \cdot 10^{10}) / (23.68) = 1.121 \cdot 10^9 \text{ kg-mole}$$

$$n_{\text{og}} = (6.30 \cdot 10^5)(4.06) = 2.558 \cdot 10^6 \text{ kg-mole}$$

$$n_{\bar{S}_0} = (1.77 \cdot 10^8)(4.06) = 7.186 \cdot 10^8 \text{ kg-mole}$$

$$y_{\bar{g}} = \frac{n_{\bar{g}g}}{n_{\bar{g}g} + n_{\bar{g}\bar{g}}} = \underline{0.9971} = y_{\bar{g}}$$

$$y_{\bar{0}} = 1 - y_{\bar{1}} = \underline{0.0029} = y_{\bar{0}}$$

$$x_{\bar{g}} = \frac{n_{\bar{g}0}}{n_{\bar{g}0} + n_{\bar{g}0}} = \underline{0.6094} = x_{\bar{g}}$$

$$x_{\bar{o}} = 1 - x_{\bar{g}} = \underline{0.3906} = x_{\bar{o}}$$

$$k_{\bar{g}} = \frac{y_{\bar{g}}}{x_{\bar{g}}} = \underline{1.636} = k_{\bar{g}}$$

$$K_{\bar{o}} = \frac{y_{\bar{o}}}{x_{\bar{o}}} = \underline{0.0074} = K_{\bar{o}}$$

$$z_{\bar{g}} = \frac{n_{\bar{g}g} + n_{\bar{g}o}}{(n_{\bar{g}g} + n_{\bar{g}o}) + (n_{\bar{o}g} + n_{\bar{o}o})} = 0.7358 = z_g$$

$$z_0 = 1 - z_{\bar{q}} = \underline{0.2642} = z_{\bar{q}}$$

(2)

(a) P^2 can be used when $P_2 \approx 150-200$ bars where $P/\mu z \sim$ linear with zero intercept.

P_p should be used when $P_2 \gtrsim 200$ bars.

$$P_p = 2 \int_{P_0}^P \frac{P}{\mu z} dp \quad \text{or} \quad \tilde{P}_p = \int \frac{1}{\mu g B_g} dp$$

$$\mu = \mu_g$$

$$z = z_g$$

P_0 = arbitrary reference (low) pressure

(b)

$$\frac{1}{\mu g B_g} \sim \frac{P}{\mu_g z_g}$$

$$B_g = \frac{P_{sc}}{T_{sc}} \frac{T_R z_g}{P}$$

$$\frac{1}{\mu g B_g} = \underbrace{\left(\frac{T_{sc}}{P_{sc} T_R} \right)}_{\text{constant}} \left(\frac{P}{\mu_g z_g} \right)$$

(c) If $C_R = \text{constant} \Rightarrow$ Darcy eq.

i.e. not Dg_g term in C_R

(d) Friction

Gravity (e^s terms)

Eros

(e)

$$q_g = C_R (P_c^2 - P_{wf}^2)$$

$$q_g = C_T (P_c^2 - P_t^2)^{0.5}$$

$$P_c^2 = e^s P_c^2$$

$$P_{wf}^2 = e^s P_w^2$$

$$q_g = C_R e^s (P_c^2 - P_w^2)$$

$$q_g = C'_R (P_c^2 - P_t^2)$$

$$(P_c^2 - P_w^2) = \frac{1}{C'_R} q_g \quad (1')$$

$$(P_w^2 - P_t^2) = \frac{1}{C_T^2} q_g^2 \quad (2')$$

$$(P_c^2 - P_t^2) = B q_g^2 + A q_g \Rightarrow \boxed{B q_g^2 + A q_g - (P_c^2 - P_t^2) = 0}$$

$$B = \frac{1}{C_T^2} \quad A = \frac{1}{C'_R} = \frac{1}{e^s C_R}$$

$$q_g = \frac{-A + \sqrt{A^2 + 4BP_t^2}}{2B}^{1/2}$$

Question 1

$$P_t = \left[P_c^2 - (B q_g^2 + A q_g) \right]^{1/2}$$

Question 2

$$A = \frac{1}{1.3(10^{-3})} = 769 \sim 770$$

$$P_c^2 = 1200^2 / 1.3$$

$$B = \frac{1}{4^2} = \frac{1}{16} = 0.0625$$

$$P_c^2 = 1.108 \cdot 10^6 \text{ psia}^2$$

$$q_g = \frac{-770 + \sqrt{(770)^2 + 4(0.0625)((1200)^2 / 1.3 - 1200^2)}}{2(0.0625)}^{1/2}$$

$$\underline{q_g = 1258 \text{ Mscf/D}}$$

$$P_t = \left[(1200)^2 / 1.3 - ((0.0625)(1200)^2 + 770(1200)) \right]^{1/2}$$

$$\underline{P_t = 975 \text{ psia}}$$

$$(f) \quad d_T' = 2d_T \Rightarrow C_T' = C_T \left(\frac{d_T'}{d_T} \right)^{2.6} = 4 (2)^{2.6} = 24$$

$$(Kh)' = 2(Kh) \Rightarrow C_R' = 2C_R$$

$$A' = \frac{1}{1.3 C_R'} = \frac{1}{1.3 (10^{-3}) 2} = 385 \quad (\text{vs } 770)$$

$$B' = \frac{1}{C_T'^2} = \frac{1}{24^2} = 0.00174 \quad (\text{vs } 0.0625)$$

$$q_g = \frac{-385 + \left[(385)^2 + 4(0.00174)((1200)^2/1.3 - 200^2) \right]^{1/2}}{2(0.00174)}$$

$$\underline{q_g = 2740 \text{ Mscf/D}}$$

$$P_t = \left[(1200)^2/1.3 - ((0.00174)(200)^2 + 385(200)) \right]^{1/2}$$

$$\underline{P_t = 1015 \text{ psia}}$$